FINAL JEE-MAIN EXAMINATION - JUNE, 2022

(Held On Saturday 25thJune, 2022)

TIME : 3:00 PM to 6:00 PM

PHYSICS

SECTION-A

1. Given below are two statements. One is labelled as Assertion A and the other is labelled as Reason R. Assertion A :Two identical balls A and B thrown with same velocity 'u' at two different angles with horizontal attained the same range R. If A and B reached the maximum height h_1 and h_2 respectively, then $R = 4\sqrt{h_1h_2}$

Reason R: Product of said heights.

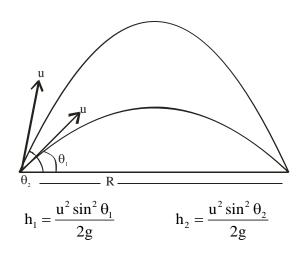
$$h_1h_2 = \left(\frac{u^2\sin^2\theta}{2g}\right) \cdot \left(\frac{u^2\cos^2\theta}{2g}\right)$$

Choose the CORRECT answer :

- (A) Both A and R are true and R is the correct explanation of A.
- (B) Both A and R are true but R is NOT the correct explanation of A.
- (C) A is true but R is false
- (D) A is false but R is true

Official Ans. by NTA (A)

Sol. For same range $\theta_1 + \theta_2 = 90^\circ$



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$$h_1h_2 = \frac{u^2 \sin^2 \theta_1}{2g} \times \frac{u^2 \sin^2 \theta_2}{2g}$$
$$\theta_2 = 90 - \theta_1$$
$$h_1h_2 = \frac{u^2 \sin^2 \theta_1}{2g} \cdot \frac{u^2 \cos^2 \theta_1}{2g}$$
$$= \left[\frac{u^2 \sin \theta_1 \cos \theta_1}{2g}\right]^2$$
$$= \left[\frac{u^2 \sin \theta_1 \cos \theta_1}{2g} \times \frac{2}{2}\right]^2 = \frac{R^2}{16}$$
$$R = 4\sqrt{h_1h_2}$$

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So R is correct explanation of A

2. Two buses P and Q start from a point at the same time and move in a straight line and their positions are represented by $X_P(t) = \alpha t + \beta t^2$ and $X_Q(t) = ft - t^2$. At what time, both the buses have same velocity ?

(A)
$$\frac{\alpha - f}{1 + \beta}$$

(B) $\frac{\alpha + f}{2(\beta - 1)}$
(C) $\frac{\alpha + f}{2(1 + \beta)}$
(D) $\frac{f - \alpha}{2(1 + \beta)}$

Official Ans. by NTA (D)

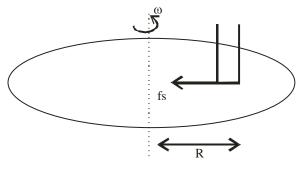
Sol.
$$X_{P}(t) = \alpha t + \beta t^{2}$$
 $X_{Q} = ft - t^{2}$
 $V_{P}(t) = \alpha + 2\beta t$ $V_{Q} = f - 2t$
 $V_{P} = V_{Q}$
 $\alpha + 2\beta t = f - 2t$
 $t = \frac{f - \alpha}{2\beta + 2}$

3. A disc with a flat small bottom beaker placed on it at a distance R from its center is revolving about an axis passing through the center and perpendicular to its plane with an angular velocity ω. The coefficient of static friction between the bottom of the beaker and the surface of the disc is μ. The beaker will revolve with the disc if :

(A)
$$R \le \frac{\mu g}{2\omega^2}$$
 (B) $R \le \frac{\mu g}{\omega^2}$
(C) $R \ge \frac{\mu g}{2\omega^2}$ (D) $R \ge \frac{\mu g}{\omega^2}$

Official Ans. by NTA (B)

Sol. For beaker to move with disc





We know that $f_s \leq f_{s \max}$

 $m\omega^2 R \leq \mu mg$

$$R \leq \frac{\mu g}{\omega^2}$$

4. A solid metallic cube having total surface area 24 m² is uniformly heated. If its temperature is increased by 10°C, calculate the increase in volume of the cube (Given : $\alpha = 5.0 \times 10^{-4} \text{ °C}^{-1}$)

Sol. Increase in volume
$$\Delta V = \gamma V_0 \Delta T$$

 $\gamma = 3\alpha$
So $\Delta V = (3\alpha) V_0 \Delta T$
Total surface area = $6a^2$, where a is side length
 $24 = 6a^2$ a= 2m
Volume $V_0 = (2)^3 = 8m^3$
 $\Delta V = (3 \times 5 \times 10^{-4})(8) \times 10$
 $= 1.2 \times 10^5 \text{ cm}^3$

- 5. A copper block of mass 5.0 kg is heated to a temperature of 500°C and is placed on a large ice block. What is the maximum amount of ice that can melt? [Specific heat of copper: $0.39 \text{ Jg}^{-1} \circ \text{C}^{-1}$ and latent heat of fusion of water : 335 J g⁻¹] (A) 1.5 kg (B) 5.8 kg (C) 2.9 kg (D) 3.8 kg **Official Ans. by NTA (C)**
- Sol. Heat given by block to get 0°C temperature

$$\Delta Q_1 = 5 \times (0.39 \times 10^3) \times (500 - 0)$$

= 975×10^3 J Heat absorbed by ice to melt m mass

$$\Delta \mathbf{Q}_2 = \mathbf{m} \times \left(335 \times 10^3\right) \mathbf{J}$$

$$\Delta Q_1 = \Delta Q_2$$

m×(335×10³) = 975×10³

$$m = \frac{975}{335} = 2.910 \, \text{kg}$$

6. The ratio of specific heats $\left(\frac{C_P}{C_V}\right)$ in terms of degree of freedom (f) is given by:

(A)
$$\left(1 + \frac{f}{3}\right)$$
 (B) $\left(1 + \frac{2}{f}\right)$
(C) $\left(1 + \frac{f}{2}\right)$ (D) $\left(1 + \frac{1}{f}\right)$

Official Ans. by NTA (B)

Sol. Molar heat capacity at constant volume $C_v = \frac{fR}{2}$ where f is degree of freedom.

Molar heat capacity at constant pressure can be written as $C_P = R + C_V = R + \frac{fR}{2} = \left(1 + \frac{f}{2}\right)R$ So $\frac{C_P}{C_v} = 1 + \frac{2}{f}$

7. For a particle in uniform circular motion, the acceleration \vec{a} at any point P(R, θ) on the circular path of radius R is (when θ is measured from the positive x –axis and v is uniform speed) :

(A)
$$-\frac{v^2}{R}\sin\theta\hat{i} + \frac{v^2}{R}\cos\theta\hat{j}$$

(B) $-\frac{v^2}{R}\cos\theta\hat{i} + \frac{v^2}{R}\sin\theta\hat{j}$
(C) $-\frac{v^2}{R}\cos\theta\hat{i} - \frac{v^2}{R}\sin\theta\hat{j}$
(D) $-\frac{v^2}{R}\hat{i} + \frac{v^2}{R}\hat{j}$
Official Ans. by NTA (C)

Sol. $a = |\vec{a}| = \frac{V^2}{R}$ $\vec{a} = -a\cos\theta\hat{i} - a\sin\theta\hat{j}$ $= -\frac{V^2}{R}\cos\theta\hat{i} - \frac{V^2}{R}\sin\theta\hat{j}$

8. Two metallic plates form a parallel plate capacitor. The distance between the plates is 'd'. A metal sheet of thickness $\frac{d}{2}$ and of area equal to area of each plate is introduced between the plates. What will be the ratio of the new capacitance to the original capacitance of the capacitor ? (A) 2:1 (B) 1:2

Official Ans.	by NTA (A)
(C) 1:4	(D) 4:1
(A) 2.1	(D) 1.2

Sol.
$$C_1 = \frac{\epsilon_0 A}{d}$$

 $C_2 = \frac{\epsilon_0 A}{\frac{d}{2} + \frac{d/2}{\infty}} = \frac{2\epsilon_0 A}{d}$
 $\frac{C_2}{C_1} = \frac{2}{1}$

9. Two cells of same emf but different internal resistances r_1 and r_2 are connected in series with a resistance R. The value of resistance R, for which the potential difference across second cell is zero, is

(A)
$$r_2 - r_1$$
 (B) $r_1 - r_2$

(C)
$$r_1$$
 (D) r_2

Official Ans. by NTA (A)

Sol.
$$I = \frac{2E}{R + r_1 + r_2} \dots (i)$$

$$A = \frac{E}{r_2 - B} + \frac{E}{r_1 - r_1}$$
But $V_A - V_B = E - Ir_2 = 0$

$$\Rightarrow I = \frac{E}{r_2} \quad \dots (ii)$$

Comparing values of I from (i) and (ii)

$$\frac{\mathrm{E}}{\mathrm{r}_2} = \frac{2\mathrm{E}}{\mathrm{R} + \mathrm{r}_1 + \mathrm{r}_2}$$
$$\implies \mathrm{R} = \mathrm{r}_2 - \mathrm{r}_1$$

10. Given below are two statements:

Statement - I: Susceptibilities of paramagnetic and ferromagnetic substances increase with decrease in temperature.

Statement – II: Diamagnetism is a result of orbital motions of electrons developing magnetic moments opposite to the applied magnetic field.

Choose the **CORRECT** answer from the options given below : -

(A) Both statement – I and statement -II are true.

(B) Both statement – I and Statement – II are false.

(C) Statement – I is true but statement – II is false.

(D) Statement-I is false but Statement-II is true.

Official Ans. by NTA (A)

Sol. According to curie's law, magnetic susceptibility is inversely proportional to temperature for a fixed value of external magnetic field i.e. $\chi = \frac{C}{T}$.

The same is applicable for ferromagnet & the relation is given as $\chi = \frac{C}{T - T_C}$ (T_c is curie

temperature)

Diamagnetism is due to non-cooperative behaviour of orbiting electrons when exposed to external magnetic field.

Hence option (A).

11. A long solenoid carrying a current produces a magnetic field B along its axis. If the current is doubled and the number of turns per cm is halved, the new value of magnetic field will be equal to

(A) B	(B) 2 B
(C) 4 B	(D) $\frac{B}{2}$

Official Ans. by NTA (A)

Sol. $B_1 = \mu_0 nI$

$$B_2 = \mu_0 \left(\frac{n}{2}\right) (2I)$$
$$\Rightarrow B_1 = B_2$$

12. A sinusoidal voltage V(t) = 210 sin 3000t volt is applied to a series LCR circuit in which L = 10 mH, C = 25 μ F and R = 100 Ω . The phase difference (Φ) between the applied voltage and resultant current will be :

(A) $\tan^{-1}(0.17)$ (B) $\tan^{-1}(9.46)$ (C) $\tan^{-1}(0.30)$ (D) $\tan^{-1}(13.33)$

Official Ans. by NTA (A)

Sol.
$$X_L = 10^{-2} \times 3000 = 30\Omega$$

$$X_{C} = \frac{1}{3000 \times 25 \times 10^{-6}} = \frac{40}{3} \Omega$$
$$X = X_{L} - X_{C}$$
$$= 30 - \frac{40}{3} = \frac{50}{3}$$
$$\tan \delta = \frac{X}{R} = \frac{50}{3 \times 100} = \frac{1}{6}$$
$$\delta = \tan^{-1} \left(\frac{1}{6}\right) = \tan^{-1} (0.17)$$

13. The electromagnetic waves travel in a medium at a speed of 2.0×10^8 m/s. The relative permeability of the medium is 1.0. The relative permittivity of the medium will be:

(A) 2.25	(B) 4.25
(C) 6.25	(D) 8.25

Official Ans. by NTA (A)

Sol.
$$V = 2 \times 10^8 \text{ m/s}$$

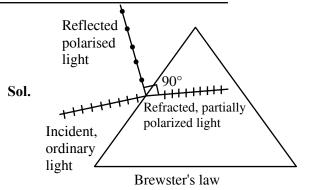
 $C = 3 \times 10^8 \text{ m/s}$
 $\frac{C}{V} = \sqrt{\mu_r \epsilon_r}$
 $\frac{9}{4} = 1 \times \epsilon_r$
 $\epsilon_r = \frac{9}{4} = 2.25$

- 14. The interference pattern is obtained with two coherent light sources of intensity ratio 4 :1. And the ratio $\frac{I_{max} + I_{min}}{I_{max} I_{min}}$ is $\frac{5}{x}$. Then, the value of x will be equal to :
 - (A) 3 (B) 4 (C) 2 (D) 1
 - Official Ans. by NTA (B)

Sol.
$$\frac{I_1}{I_2} = 4$$
$$\frac{I_{max}}{I_{min}} = \left[\frac{\sqrt{I_1} + \sqrt{I_2}}{\sqrt{I_1} - \sqrt{I_2}}\right]^2$$
$$\frac{I_{max}}{I_{min}} = \left[\frac{2\sqrt{I_2} + \sqrt{I_2}}{2\sqrt{I_2} - \sqrt{I_2}}\right]^2$$
$$\frac{I_{max}}{I_{min}} = 9$$
$$\frac{I_{max} + I_{min}}{I_{max} - I_{min}} = \frac{10}{8}$$
$$\frac{5}{x} = \frac{10}{8}$$
$$x = 4$$

- **15.** A light whose electric field vectors are completely removed by using a good Polaroid, allowed to incident on the surface of the prism at Brewster's angle. Choose the most suitable option for the phenomenon related to the prism.
 - (A) Reflected and refracted rays will be perpendicular to each other
 - (B) Wave will propagate along the surface of prism
 - (C) No refraction, and there will be total reflection of light.
 - (D) No reflection and there will be total transmission of light.

Official Ans. by NTA (D)



But as the incident light electric field vectors are completely removed so there will be no reflection and there will be total transmission of light, explained by an experiment in NCERT.

[Reference NCERT Part-2 Pg-380, (A special case of total transmission)]

Note : Since direction of polarization is not mentioned hence most suitable option (D) corresponding to case in which electric field is absent perpendicular to plane consisting incident and normal.

16. A proton, a neutron, an electron and an α -particle have same energy. If $\lambda_p, \lambda_n, \lambda_e$ and λ_α are the de Broglie's wavelengths of proton, neutron, electron and α particle respectively, then choose the correct relation from the following :

(A)
$$\lambda_p = \lambda_n > \lambda_e > \lambda_\alpha$$

(B) $\lambda_\alpha < \lambda_n < \lambda_p < \lambda_e$
(C) $\lambda_e < \lambda_p = \lambda_n > \lambda_\alpha$
(D) $\lambda_e = \lambda_p = \lambda_n = \lambda_\alpha$

Official Ans. by NTA (B)

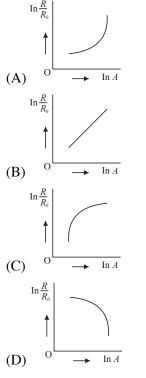
Sol.
$$\lambda = \frac{h}{\sqrt{2Em}}$$

 $\lambda \propto \frac{1}{\sqrt{m}}$
 $\therefore \quad \lambda_e > \lambda_p > \lambda_n > \lambda_\alpha$

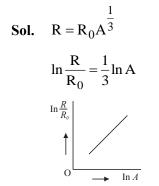
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17. Which of the following figure represents the variation of $In\left(\frac{R}{R_0}\right)$ with In A(If R = radius of a

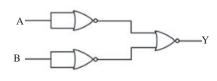
nucleus and A = its mass number)



Official Ans. by NTA (B)



18. Identify the logic operation performed by the given circuit :



(A) AND gate(B) OR gate(C) NOR gate(D) NAND gate

Official Ans. by NTA (A)

Sol. =
$$\left[\overline{\left[\overline{A+A}\right] + \left[\overline{B+B}\right]}\right]$$

$$Y = \overline{\overline{A} + \overline{B}} (D' \text{ MORGAN LAW})$$
$$Y = AB$$

19. Match List I with List II

List	-I	List –	II
A	Facsimile	I.	Static Document Image
В.	Guided media Channel	II.	Local Broadcast Radio
C.	Frequency Modulation	III.	Rectangular wave
D.	Digital Signal	IV.	Optical Fiber

Choose the correct answer from the following options :

(A) A –IV, B-III, C-II, D-I
(B)A-I, B-IV, C-II, D-III
(C) A –IV, B-II, C-III, D-I
(D) A-I, B-II, C-III, D-IV
Official Ans. by NTA (B)

- Sol. Question based on the theory given in NCERT.
- 20. If n represents the actual number of deflections in a converted galvanometer of resistance G and shunt resistance S. Then the total current I when its figure of merit is K will be :

(A)
$$\frac{KS}{(S+G)}$$
 (B) $\frac{(G+S)}{nKS}$
(C) $\frac{nKS}{(G+S)}$ (D) $\frac{nK(G+S)}{S}$

Official Ans. by NTA (D)

Sol.

$$I = \frac{I_g}{I - I_g} G$$
Figure of merit $\frac{I_g}{\theta} = K$

$$I_g = Kn$$

$$I = \frac{I_g}{S} (G + S)$$

$$I = \frac{nK}{S} (G + S)$$

SECTION-B

1. For $z = a^2 x^3 y^{\frac{1}{2}}$, where 'a' is a constant. If percentage error in measurement of 'x' and 'y' are 4% and 12%, respectively, then the percentage error for 'z' will be %.

Official Ans. by NTA (18)

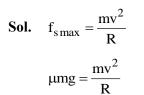
Sol.
$$z = a^2 x^3 y^{1/2}$$

$$\frac{\Delta z}{z} = \frac{2\Delta a}{a} + \frac{3\Delta x}{x} + \frac{1}{2}\frac{\Delta y}{y}$$

a is constant

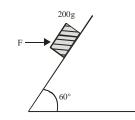
$$\frac{\Delta z}{z} \times 100 = 3(4\%) + \frac{1}{2}(12\%) = 18\%$$

Official Ans. by NTA (24)



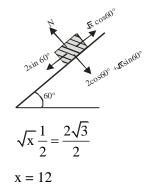
$$v = \sqrt{\mu Rg}$$
$$\frac{v_2}{v_1} = \sqrt{\frac{R_2}{R_1}}$$
$$\frac{v_2}{30} = \sqrt{\frac{48}{75}}$$
$$v_2 = 24 \text{ m/s}$$

3. A block of mass 200 g is kept stationary on a smooth inclined plane by applying a minimum horizontal force $F = \sqrt{x}N$ as shown in figure. The value of x = _____.



Official Ans. by NTA (12)

Sol. mg = 2N



4. Moment of Inertia (M.I.) of four bodies having same mass 'M' and radius '2R' are as follows:
I₁= M.I. of solid sphere about its diameter
I₂ = M.I. of solid cylinder about its axis
I₃ = M.I. of solid circular disc about its diameter
I₄ = M.I. of thin circular ring about its diameter
If 2(I₂ + I₃) + I₄ = x. I₁ then the value of x will be

Official Ans. by NTA (5)

7

Sol.
$$I_1 = \frac{2}{5}M(2R)^2 = \frac{8}{5}MR^2$$

 $I_1 = \frac{1}{2}M(2R)^2 = 2MR^2$
 $I_3 = \frac{M(2R)^2}{4} = MR^2$
 $I_4 = \frac{M(2R)^2}{2} = 2MR^2$
 $2(I_2 + I_3) + I_4 = x I_1$
 $8MR^2 = x\frac{8}{5}MR^2$
 $x = 5$

5. Two satellites S₁ and S₂ are revolving in circular orbits around a planet with radius $R_1 = 3200$ km and $R_2 = 800$ km respectively. The ratio of speed of satellite S_1 to the speed of satellite S_2 in their respective orbits would be $\frac{1}{x}$ where x =

Official Ans. by NTA (2)

Sol.
$$V = \frac{GM}{r} \Rightarrow \frac{V_1}{V_2} = \sqrt{\frac{800}{3200}} = \frac{1}{2}$$

When a gas filled in a closed vessel is heated by 6. raising the temperature by 1°C, its pressure increase by 0.4%. The initial temperature of the gas is_____K.

Official Ans. by NTA (250)

Sol. pV = nRT $\Delta P.V = nR\Delta T$ $\frac{\Delta P}{P} = \frac{\Delta T}{T} = \frac{0.4}{100}$ \Rightarrow

$$\Rightarrow T = \frac{100 \times 1}{0.4} = 250 \text{K}$$

7. 27 identical drops are charged at 22V each. They combine to form a bigger drop. The potential of the bigger drop will be _____ V.

Official Ans. by NTA (198)

Sol.
$$q \rightarrow nq$$

 $n\frac{4}{3}\pi r^3 = \frac{4}{3}\pi (r')^3$
 $\Rightarrow r' = n\frac{1}{3}r$

a \na

8.

$$V = \frac{kq}{r} \propto \frac{n}{n^{1/3}} \propto n^{2/3} \propto 27^{2/3} \Rightarrow v' = 9V = 9 \times 22 = 198$$

The length of a given cylindrical wire is increased to double of its original length. The percentage increase in the resistance of the wire will be _%.

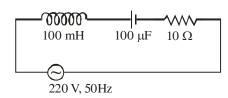
Official Ans. by NTA (300)

Sol. V' = V

$$\ell'A = \ell A$$

 $2\ell A' = \ell A$
 $A' = \frac{A}{2}$
 $R = \rho \frac{\ell}{A} \dots (i)$
 $\ell' = 2\ell$
 $A' = \frac{A}{2}$
 $R' = \frac{\rho \ell'}{A'} = \frac{\rho 2\ell}{\frac{A}{2}}$
 $R' = \frac{4\rho \ell}{A}$
 $R' = 4R$ from equation (i)
% increase in resistance
 $= \frac{R'-R}{R} \times 100 = \frac{4R-R}{R} \times 100$
 $= 300\%$

9. In a series LCR circuit, the inductance, capacitance and resistance are L = 100mH, C = 100µF and $R = 10\Omega$ respectively. They are connected to an AC source of voltage 220V and frequency of 50 Hz. The approximate value of current in the circuit will be_____A.

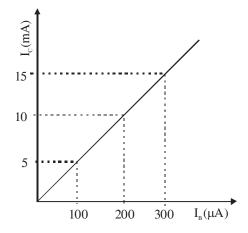


Official Ans. by NTA (22)

Sol.
$$X_L = \omega L = 2\pi \times 50 \times 10^{-1} = 10\pi$$

 $X_X = \frac{1}{\omega C} = \frac{1}{2\pi \times 50} \times 10^4 = \frac{100}{\pi}$
 $R = 10\Omega$
 $Z = \sqrt{\left(10\pi - \frac{100}{\pi}\right)^2 + 10^2} \approx 10\Omega$
 $i = \frac{E}{2} \approx \frac{220}{10} \approx 22$ Amp

10. In an experiment of CE configuration of n-p-n transistor, the transfer characteristics are observed as given in figure.



If the input resistance is 200Ω and output resistance is 60Ω the voltage gain in this experiment will be _____

Official Ans. by NTA (15)

Sol. Voltage Gain = $\frac{I_C}{I_B} \times \frac{R_0}{R_I} = \frac{10 \times 10^{-3}}{200 \times 10^{-6}} \times \frac{60}{200} = 15$

FINAL JEE-MAIN EXAMINATION - JUNE, 2022 (Held On Saturday 25[®] June, 2022) TIME: 3:00 PM to 6:00 PM **CHEMISTRY TEST PAPER WITH SOLUTION SECTION-A Official Ans. by NTA (D)** 1. The minimum energy that must be possessed by photons in order to produce the photoelectric effect **Sol.** $\Delta T = i k_f \times m$ with platinum metal is: $0.2 = i \times 1.86 \times \frac{0.7}{93} \times \frac{1000}{42}$ [Given: The threshold frequency of platinum is 1.3 $\times 10^{15} \text{ s}^{-1}$ and h = 6.6 $\times 10^{-34} \text{ J s.}$] $i = \frac{0.2 \times 93 \times 6}{1.86 \times 100}$ (A) 3.21×10^{-14} J (B) 6.24×10^{-16} J i = 0.60(C) 8.58×10^{-19} J (D) 9.76×10^{-20} J $2A \rightleftharpoons A_2$ **Official Ans. by NTA (C)** $1-\alpha \qquad \frac{\alpha}{2}$ **Sol.** W = hv $= 6.6 \times 10^{-34} \times 1.3 \times 10^{15}$ $i = 1 - \alpha + \frac{\alpha}{2}$ $= 8.58 \times 10^{-19} \text{ J}$ $i=1-\frac{\alpha}{2}$ At 25°C and 1 atm pressure, the enthalpy of 2. combustion of benzene (1) and acetylene (g) are $1 - \frac{\alpha}{2} = 0.60$ -3268 kJ mol⁻¹ and -1300 kJ mol⁻¹, respectively. The change in enthalpy for the reaction $1 - 0.60 = \frac{\alpha}{2}$ $3 C_2H_2(g) \rightarrow C_6H_6(l)$, is $(A) + 324 \text{ kJ mol}^{-1}$ $\alpha = 0.80$ (B) $+632 \text{ kJ mol}^{-1}$ The K_{sp} for bismuth sulphide (Bi₂S₃) is $(D) - 732 \text{ kJ mol}^{-1}$ $(C) - 632 \text{ kJ mol}^{-1}$ 4. 1.08×10^{-73} . The solubility of Bi₂S₃ in mol L⁻¹ at **Official Ans. by NTA (C)** 298 K is (B) 2.7×10^{-12} (A) 1.0×10^{-15} **Sol.** $\Delta H = \sum \Delta H_{Combustion}$ (Reactant) - $\sum \Delta H_{Combustion}$ (C) 3.2×10^{-10} (D) 4.2×10^{-8} (Product) **Official Ans. by NTA (A)** $= 3 \times (-1300) - [-3268]$ $= -632 \text{ kJ mol}^{-1}$ Solute A associates in water. When 0.7 g of solute Sol. $Bi_2S_3 \Longrightarrow 2Bi^{3+} + 3S^{2-}$ 3. $\begin{array}{c} 2s \\ k_{sp} = (2s)^2 (3s)^3 \end{array}$ 3s A is dissolved in 42.0 g of water, it depresses the freezing point by 0.2°C. The percentage $=4s^{2} \times 27(s)^{3}$ association of solute A in water, is $= 108 (s)^{5}$ [Given : Molar mass of A = 93 g mol⁻¹. Molal $(s)^5 = \frac{1.08 \times 10^{-73}}{108}$ depression constant of water is 1.86 K kg mol⁻¹] (A) 50 % (B) 60 % \Rightarrow s = 10⁻¹⁵ (C) 70 % (D) 80 %

5.	Match List I with List II.		
	List I	List II	
	A. Zymase	I. Stomach	
	B. Diastase	II. Yeast	
	C. Urease	III. Malt	
	D. Pepsin	IV. Soyabean	
	Choose the correct answ	ver from the options given	
	below:		
	(A) A-II, B-III, C-I, D-IV		
	(B) A-II, B-III, C-IV, D-I		
	(C) A-III, B-II, C-IV, D-I		
	(D) A-III, B-II, C-I, D-IV		
	Official Ans. by NTA (I	3)	
Sol.	Zymase naturally occur	rs in yeast.	
	Diastase is found in ma	ılt.	
	Urease is found in soya	ibean	
	Pepsin is found in stor	ach	
6.	The correct order of elec	etron gain enthalpies of Cl,	
	F, Te and Po is		
	(A) F < Cl < Te < Po	(B) Po < Te < F < Cl	
	(C) Te < Po < Cl < F	(D) Cl < F < Te < Po	
	Official Ans. by NTA (I	D)	
Sol.	As Cl has maximum electron affinity among all		
	elements.		
	Element	Δ_{eg} H (kJ/mol)	
	F Cl	-328 -349	
	Те	-190	
	Po	-174	
7.	Given below are two stat		
/•			
	Statement I: During electrolytic refining, blister copper deposits precious metals Statement II: In the process of obtaining pure		
	*	method, copper blister is	
	used to make the anode.		

In the light of the above statements, choose the correct answer from the options given below.
(A) Both Statement I and Statement II are true.
(B) Both Statement I and Statement II are false.
(C) Statement I is true but Statement II is false.
(D) Statement I is false but Statement II is true.
Official Ans. by NTA (A)

- **Sol.** In the electro-refining, impure metal (here blister copper) is used as an anode while precious metal like Au, Pt get deposited as anode mud.
- 8. Given below are two statements one is labelled as
 Assertion A and the other is labelled as Reason R:
 Assertion A : The amphoteric nature of water is explained by using Lewis acid/base concept.

Reason R : Water acts as an acid with NH_3 and as a base with H_2S .

In the light of the above statements choose the correct answer from the options given below :

- (A) Both A and R are true and R is the correct explanation of A.
- (B) Both A and R are true but R is NOT the correct explanation of A.
- (C) A is true but R is false.
- (D) A is false but R is true.

Official Ans. by NTA (D)

Sol. $H_2S + H_2O \Longrightarrow H_3O^+ + HS^-$ Acid Base

 $\underset{Acid}{\text{H}_2\text{O}} + \underset{Base}{\text{NH}_3} \xrightarrow{} \text{NH}_4\text{OH}$

9. The correct order of reduction potentials of the following pairs is

A. Cl_2/Cl^-

- B. I_2/I^-
- C. Ag^+/Ag
- D. Na⁺/Na
- E. Li⁺/Li

Choose the correct answer from the options given below.

(A) A > C > B > D > E
(B) A > B > C > D > E
(C) A > C > B > E > D
(D) A > B > C > E > D
Official Ans. by NTA (A)

Sol.
$$E_{Cl_2/Cl^-}^{\circ} = +1.36 V$$

 $E_{I_2/I^-}^{\circ} = +0.54 V$
 $E_{Ag^+/Ag}^{\circ} = +0.80 V$
 $E_{Na^+/Na}^{\circ} = -2.71 V$
 $E_{Li^+/Li}^{\circ} = -3.05 V$

10. The number of bridged oxygen atoms present in compound B formed from the following reactions is

$$Pb(NO_3)_2 \xrightarrow{673 \text{ K}} A + PbO + O_2$$

 $A \xrightarrow{\text{Dimerise}} B$

(A) 0	(B) 1
(C) 2	(D) 3

Official Ans. by NTA (A)

Sol.

P

$$b(NO_3)_2 \xrightarrow{\Delta} PbO + NO_2 + O_2$$

(A)

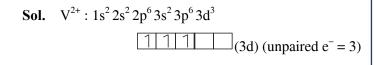
 $NO_{A} \xrightarrow{\text{Dimerise}} N_{2}O_{4}$



(no bridged oxygen)

 The metal ion (in gaseous state) with lowest spinonly magnetic moment value is

(A) V ²⁺	(B) Ni ²⁺
(C) Cr^{2+}	(D) Fe ²⁺
Official Ans. by NTA (B)	



Ni²⁺:
$$1s^{2} 2s^{2} 2p^{6} 3s^{2} 3p^{6} 3d^{8}$$

$$\boxed{111111} (3d) \text{ (unpaired } e^{-} = 2)$$
Cr²⁺: $1s^{2} 2s^{2} 2p^{6} 3s^{2} 3p^{6} 3d^{4}$

$$\boxed{111111} (3d) \text{ (unpaired } e^{-} = 4)$$
Fe²⁺: $1s^{2} 2s^{2} 2p^{6} 3s^{2} 3p^{6} 3d^{6}$

$$\boxed{111111} (3d) \text{ (unpaired } e^{-} = 4)$$

12. Given below are two statements: one is labelled as Assertion A and the other is labelled as Reason R Assertion A: Polluted water may have a value of BOD of the order of 17 ppm.

Reason R: BOD is a measure of oxygen required to oxidise both the biodegradable and nonbiodegradable organic material in water.

In the light of the above statements, choose the most appropriate answer from the options given below.

(A) Both A and R are correct and R is the correct explanation of A.

(B) Both A and R are correct but R is NOT the correct explanation of A.

- (C) A is correct but R is not correct.
- (D) A is not correct but R is correct.

Official Ans. by NTA (C)

Sol. Clean water have BOD less than 5 ppm while highly polluted water has BOD greater or equal to 17 ppm. So, assertion is correct.

BOD is measure of oxygen required to oxidise only bio-degradable organic matter. So, reason is false.

13. Given below are two statements: one is labelled as Assertion A and the other is labelled as Reason R. Assertion A: A mixture contains benzoic acid and napthalene. The pure benzoic acid can be separated out by the use of benzene.

Reason R: Benzoic acid is soluble in hot water.

In the light of the above statements, choose the most appropriate answer from the options given below.

(A) Both A and R are true and R is the correct explanation of A.

(B) Both A and R are true but R is NOT the correct explanation of A.

(C) A is true but R is false.

(D) A is false but R is true.

Official Ans. by NTA (D)

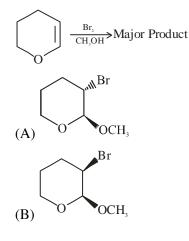
Sol. Benzoic acid and Napthalene can be effectively separated by crystallization. Benzoic acid is soluble in hot water whereas Napthalene is insoluble.

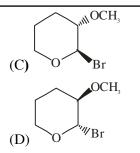
Hence assertion is incorrect but reason is correct

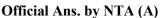
- During halogen test, sodium fusion extract is boiled with concentrated HNO₃ to
 - (A) remove unreacted sodium
 - (B) decompose cyanide or sulphide of sodium
 - (C) extract halogen from organic compound
 - (D) maintain the pH of extract

Official Ans. by NTA (B)

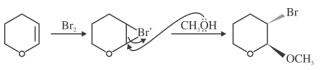
- **Sol.** Sodium fusion extract is boiled with concentrated HNO₃ to remove sodium cyanide and sodium sulphide
- **15.** Amongst the following, the major product of the given chemical reaction is



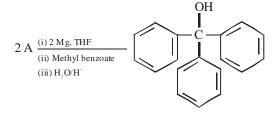




Sol.



16. In the given reaction

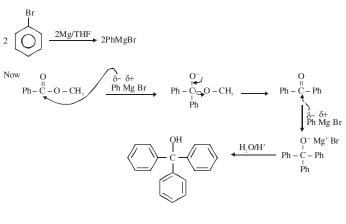


'A' can be

- (A) benzyl bromide (B) bromobenzene
- (C) cyclohexyl bromide (D) methyl bromide

Official Ans. by NTA (B)

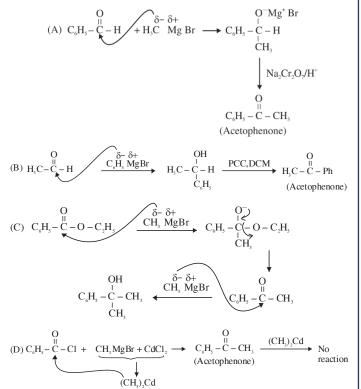
Sol.



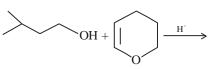
17. Which of the following conditions or reaction sequence will NOT give acetophenone as the major product ? (A) (a) $C_{o}H_{3}$ $H + CH_{3}MgBr$ (b) $Na_{2}Cr_{2}O_{7}$, H^{+} (B) (a) $H_{3}C$ $H + C_{o}H_{3}MgBr$ (b) PCC, DCM (C) $C_{o}H_{3}$ $OC_{2}H_{5} + 2CH_{3}MgBr$ (D) $C_{o}H_{5}$ $Cl + CH_{3}MgBr + CdCl_{2}$

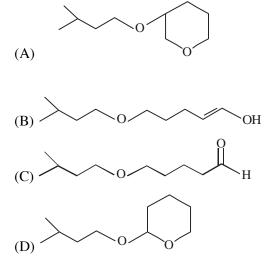
Official Ans. by NTA (C)

Sol.



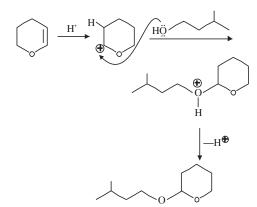
18. The major product formed in the following reaction, is



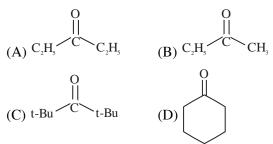


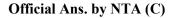
Official Ans. by NTA (D)

Sol.



19. Which of the following ketone will NOT give enamine on treatment with secondary amines? [where t-Bu is -C(CH₃)₃]





Sol. Enamine formation is an example of nucleophilic addition elimination reaction

Since in ketone $\begin{array}{c} H_3C & 0 & CH_3 \\ H_3C & C & -C & -CH_3 \\ H_3C & C & C & -CH_3 \\ H_3C & CH_3 & CH_3 \end{array}$ Carbonyl

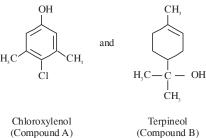
Group is highly sterically hindered hence attack of nucleophile will not be possible.

- An antiseptic dettol is a mixture of two compounds 20. 'A' and 'B' where A has 6π electrons and B has 2π electrons. What is 'B'? (A) Bithionol

 - (B) Terpineol
 - (C) Chloroxylenol
 - (D) Chloramphenicol
 - Official Ans. by NTA (B)

Sol. Dettol is mixture of

It has 6πe⁻



(Compound B) It has 2πe⁻

Hence compound 'B' is Terpineol.

SECTION-B

1. A protein 'A' contains 0.30% of glycine (molecular weight 75). The minimum molar mass of the protein 'A' is $\times 10^3$ g mol⁻¹ [nearest integer]

Official Ans. by NTA (25)

Sol. 0.30 % glycine is equal to 75

$$1 \% \longrightarrow \frac{75}{0.30}$$
$$100 \% \longrightarrow \frac{75}{0.30} \times 100$$
$$= 25000 \text{ g}$$

2. A rigid nitrogen tank stored inside a laboratory has a pressure of 30 atm at 06:00 am when the temperature is 27 °C. At 03:00 pm, when the temperature is 45°C, the pressure in the tank will be _____atm. [nearest integer]

Official Ans. by NTA (32)

Sol.
$$\frac{P_1}{T_1} = \frac{P_2}{T_2}$$

 $\frac{30}{300} = \frac{P_2}{318}$
 $P_2 = \frac{30}{300} \times 318$
 $= \frac{1}{10} \times 318$
 $= 32$

3. Amongst BeF₂, BF₃, H₂O, NH₃, CCl₄ and HCl, the number of molecules with non-zero net dipole moment is _____.

Official Ans. by NTA (3)

- **Sol.** BeF₂, BF₃ and CCl₄ $\Rightarrow \mu_{net} = 0$ H₂O, NH₃ and HCl $\Rightarrow \mu_{net} \neq 0$
- 4. At 345 K, the half life for the decomposition of a sample of a gaseous compound initially at 55.5 kPa was 340 s. When the pressure was 27.8 kPa, the half life was fund to be 170 s. The order of the reaction is_____. [integer answer]

Official Ans. by NTA (0)

Sol.
$$t_{1/2} \times \frac{1}{[P_0]^{n-1}}$$

 $\frac{t_1}{t_2} = \frac{(P_2)^{n-1}}{(P_1)^{n-1}}$
 $\frac{340}{170} = \left(\frac{27.8}{55.5}\right)^{n-1}$
 $\Rightarrow 2 = \frac{1}{(2)^{n-1}}$
 $n = 0$
5. A solution of Fe₂(SO₄)₃ is e

A solution of Fe₂(SO₄)₃ is electrolyzed for 'x' min with a current of 1.5 A to deposit 0.3482 g of Fe. The value of x is_____. [nearest integer] Given : $1 \text{ F} = 96500 \text{ C mol}^{-1}$ Atomic mass of $Fe = 56 \text{ g mol}^{-1}$ Official Ans. by NTA (20)

Sol. $Fe^{3+} + 3e^{-} \longrightarrow Fe$ $3F \longrightarrow 1$ mole Fe is deposited For 56 g \longrightarrow 3 × 96500 (required charge) For 1g $\longrightarrow \frac{3 \times 96500}{56}$ (required charge) For 0.3482 g $\longrightarrow \frac{3 \times 96500}{56} \times 0.3482$ = 1800.06Q = it1800.06 = 1.5 t $t = 20 \min$ 6. Consider the following reactions : $PCl_3 + H_2O \longrightarrow A + HCl$ $A + H_2O \longrightarrow B + HCl$ number of ionisable protons present in the product B_____. Official Ans. by NTA (2) **Sol.** $PCl_3+H_2O \xrightarrow{Partial} PCl_2(OH)$ (or) $PCl(OH)_2 +$ $\begin{array}{c} \text{PCl}_2(\text{OH}) \text{ (or) } \text{PCl}(\text{OH})_2 \xrightarrow{\text{water}} \begin{array}{c} O \\ H \\ P \\ OH \end{array} + \text{HCl} \\ H \end{array}$ no. of ionisable protons in B = 27. Amongst $FeCl_3.3H_2O_3$ $K_3[Fe(CN)_6]$ [Co(NH₃)₆]Cl₃, the spin-only magnetic moment value of the inner-orbital complex that absorbs

Official Ans. by NTA (2)

[nearest integer]

Sol. $[Fe(H_2O)_3Cl_3]$, $\underbrace{K_3[Fe(CN)_6], [Co(NH_3)_6]Cl_3}_{inner orbital complexes}$

 $K_3[Fe(CN)_6]$ has more value of Δ_0 than that of $[Co(NH_3)_6]Cl_3$; as $\overline{C}N$ is stronger ligand.

light at shortest wavelength is _____ B.M.

More $\Delta_0 \Rightarrow$ smaller value of absorbed λ

$$\begin{array}{cccc} K_{3}[Fe(CN)_{6}]: \ Fe^{3+} & : \ 3d^{5} & 4s^{0} & 4p^{0} \\ & & & \\ \hline 1 & & \\ 1 \ unpaired \ e^{-} & d^{2}sp^{3} \end{array}$$

Spin only magnetic moment (μ) = $\sqrt{3}$ BM = 1.732 BM

Rounding off $\Rightarrow 2$

- 8. The Novolac polymer has mass of 963 g. The number of monomer units present in it are Official Ans. by NTA (9)
- OH CH₂OH Monomer unit of Novolac is Sol. its

molecular mass is 124 amu.

Upon considering molecular weight of polymer as 963 amu (In question its given as 963 gram) Now if during formation of Novolac, (n-1) unit of water are removed then

 $n \times 124 = 963 + [18 \times (n-1)]$

n = 9

HC1

and

- 9. How many of the given compounds will give a positive Biuret test _____ ? Glycine, Glycylalanine, Tripeptide, Biuret Official Ans. by NTA (2)
- Biuret test is given by all proteins and peptides Sol. having atleast two peptide linkages. Hence positive test must be given by tripeptide and Biuret.
- 10. The neutralization occurs when 10 mL of 0.1 M acid 'A' is allowed to react with 30 mL of 0.05 M base M(OH)₂. The basicity of the acid 'A' is_____. [M is a metal] Official Ans. by NTA (3)

Sol. Acid + Base
$$\longrightarrow$$
 Salt + H₂O
0.1 M M(OH)₂
10ml 0.05 M
30 ml
at equivalence point
equivalent of acid = equivalent of base
0.1 × 10 × n = 30 × 0.05 × 2
n = 3

7

FINAL JEE-MAIN EXAMINATION - JUNE, 2022 (Held On Saturday 25thJune, 2022) TIME: 3:00 PM to 6:00 PM **TEST PAPER WITH SOLUTION MATHEMATICS SECTION-A** 3. Let z_1 and z_2 be two complex numbers such that Let $A = \{x \in R : |x+1| < 2\}$ and 1. $\overline{z}_1 = i\overline{z}_2$ and $\arg\left(\frac{z_1}{\overline{z}_2}\right) = \pi$. Then $B = \{x \in R : |x-1| \ge 2\}$. Then which one of the following statements is NOT true ? (A) arg $z_2 = \frac{\pi}{4}$ (B) arg $z_2 = -\frac{3\pi}{4}$ (A) A - B = (-1, 1)(B) B - A = R - (-3, 1)(C) arg $z_1 = \frac{\pi}{4}$ (D) arg $z_1 = -\frac{3\pi}{4}$ (C) $A \cap B = (-3, -1]$ (D) $A \cup B = R - [1, 3)$ Official Ans. by NTA (B) **Official Ans. by NTA (C) Sol.** A: $x \in (-3, 1)$ B: $x \in (-\infty, -1] \cup [3, \infty)$ **Sol.** $\overline{z}_1 = i\overline{z}_2$ $B - A = (-\infty, -3] \cup [3, \infty) = R - (-3, 3)$ $z_1 = -iz_2$ 2. Let $a, b \in \mathbb{R}$ be such that the equation $\arg\left(\frac{z_1}{\overline{z_2}}\right) = \pi$ $ax^2 - 2bx + 15 = 0$ has a repeated root α . If α and β are the roots of the equation $x^2 - 2bx + 21 = 0$, $\operatorname{arg}\left(-\operatorname{i}\frac{Z_{2}}{\overline{Z_{2}}}\right) = \pi$ $\operatorname{arg}(Z_{2}) = \theta$ then $\alpha^2 + \beta^2$ is equal to: (A) 37 (B) 58 $-\frac{\pi}{2} + \theta + \theta = \pi$ (C) 68 (D) 92 Official Ans. by NTA (B) $2\theta = \frac{3\pi}{2}$ Sol. $ax^2 - 2bx + 15 = 0$ $\arg(z_2) = \theta = \frac{3\pi}{4}$, $\arg z_1 = \frac{\pi}{4}$ $2\alpha = \frac{2b}{\alpha}, \alpha^2 = \frac{15}{\alpha}$ 4. The system of equations $\frac{\alpha}{2} = \frac{15}{2h}$ -kx + 3y - 14z = 25 $\alpha = \frac{15}{1}$ -15x + 4y - kz = 3-4x + y + 3z = 4 $x^2 - 2bx + 21 = 0$ is consistent for all k in the set $\left(\frac{15}{h}\right)^2 - 2b\left(\frac{15}{h}\right) + 21 = 0$ (A)R $(B) R - \{-11, 13\}$ (C) $R - \{13\}$ (D) $R - \{-11, 11\}$ $b^2 = 25$ $\alpha + \beta = 2b, \ \alpha\beta = 21$ **Official Ans. by NTA (D)** $\alpha^2 + \beta^2 = 4b^2 - 42$ = 58

Sol.
$$\Delta = \begin{vmatrix} -k & 3 & -14 \\ -15 & 4 & -k \\ -4 & 1 & 3 \end{vmatrix} = 121 - k^{2}$$

$$\Delta \neq 0 \qquad k \in \mathbb{R} - \{11, -11\} \text{ (Unique sol.)}$$

If k = 11

$$\Delta_{z} = \begin{vmatrix} -11 & 3 & 25 \\ -15 & 4 & 3 \\ -4 & 1 & 4 \end{vmatrix} \neq 0$$

No solution
If k = -11

$$\Delta_{z} = \begin{vmatrix} 11 & 3 & 25 \\ -15 & 4 & 3 \\ -4 & 1 & 4 \end{vmatrix} \neq 0$$

No solution
5.
$$\lim_{x \to \frac{\pi}{2}} \left(\tan^{2} x \left(\left(2\sin^{2} x + 3\sin x + 4 \right)^{\frac{1}{2}} - \left(\sin^{2} x + 6\sin x + 2 \right)^{\frac{1}{2}} \right) \right)$$

is equal to
(A) $\frac{1}{12}$
(B) $-\frac{1}{18}$
(C) $-\frac{1}{12}$
(D) $-\frac{1}{6}$

Official Ans. by NTA (A)

Sol.

$$\lim_{x \to \frac{\pi}{2}} \tan^2 x \left[\sqrt{2 \sin^2 x + 3 \sin x + 4} - \sqrt{\sin^2 x + 6 \sin x + 2} \right] =$$

$$\lim_{x \to \frac{\pi}{2}} \frac{\tan^2 x \left[\sin^2 x - 3 \sin x + 2 \right]}{\sqrt{9} + \sqrt{9}}$$

$$= \lim_{x \to \frac{\pi}{2}} \frac{\tan^2 x (\sin x - 1)(\sin x - 2)}{6}$$

$$= \frac{1}{6} \lim_{x \to \frac{\pi}{2}} \tan^2 x (1 - \sin x)$$

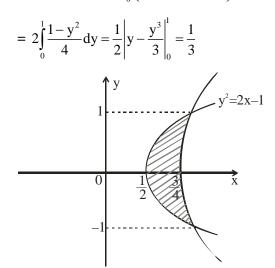
$$= \frac{1}{6} \lim_{x \to \frac{\pi}{2}} \frac{\sin^2 x (1 - \sin x)}{(1 - \sin x)(1 + \sin x)} = \frac{1}{12}$$

6. The area of the region enclosed between the parabolas $y^2 = 2x - 1$ and $y^2 = 4x - 3$ is (A) $\frac{1}{x}$ (B) $\frac{1}{x}$

(C)
$$\frac{2}{3}$$
 (D) $\frac{3}{4}$

Official Ans. by NTA (A)

Sol. Required area = $2\int_{0}^{1} \left(\frac{y^2+3}{4} - \frac{y^2+1}{2}\right) dy$



7. The coefficient of x^{101} in the expression $(5+x)^{500} + x(5+x)^{499} + x^2(5+x)^{498} + \dots x^{500}$,

x > 0, is

(A)
$${}^{501}C_{101}(5){}^{399}$$
 (B) ${}^{501}C_{101}(5){}^{400}$
(C) ${}^{501}C_{100}(5){}^{400}$ (D) ${}^{500}C_{101}(5){}^{399}$

Official Ans. by NTA (A)

Sol. $(5+x)^{500} + x(5+x)^{499} + x^2(5+x)^{498} + \dots + x^{500}$

$$=\frac{(5+x)^{501}-x^{501}}{(5+x)-x}=\frac{(5+x)^{501}-x^{501}}{5}$$

 \Rightarrow coefficient x¹⁰¹ in given expression

$$=\frac{{}^{501}\mathrm{C}_{101}5^{400}}{5}={}^{501}\mathrm{C}_{101}5^{399}$$

8. The sum $1 + 2 \cdot 3 + 3 \cdot 3^2 + \dots + 10 \cdot 3^9$ is equal to (A) $\frac{2 \cdot 3^{12} + 10}{4}$ (B) $\frac{19 \cdot 3^{10} + 1}{4}$ (C) $5 \cdot 3^{10} - 2$ (D) $\frac{9 \cdot 3^{10} + 1}{2}$

Official Ans. by NTA (B)

Sol.
$$S = 1 \cdot 3^{0} + 2 \cdot 3^{1} + 3 \cdot 3^{2} + \dots + 10.3^{9}$$

 $3S = 1 \cdot 3^{1} + 2.3^{2} \dots + 9 \times 3^{9} + 10 \times 3^{10}$
 $-2S = (1 \cdot 3^{0} + 3^{1} + 3^{2} \dots 3^{9}) - 10.3^{10}$
 $S = 5 \times 3^{10} - \left(\frac{3^{10} - 1}{4}\right)$
 $S = \frac{20 \cdot 3^{10} - 3^{10} + 1}{4} = \frac{19 \cdot 3^{10} + 1}{4}$

9. Let P be the plane passing through the intersection of the planes

 $\vec{r} \cdot (\hat{i} + 3\hat{j} - \hat{k}) = 5$ and $\vec{r} \cdot (2\hat{i} - \hat{j} + \hat{k}) = 3$, and the point (2,1,-2). Let the position vectors of the points X and Y be $\hat{i} - 2\hat{j} + 4\hat{k}$ and $5\hat{i} - \hat{j} + 2\hat{k}$ respectively. Then the points (A) X and X + Y are on the same side of P

(B) Y and Y - X are on the opposite sides of P

(C) X and Y are on the opposite sides of $\ensuremath{\mathsf{P}}$

(D) X + Y and X – Y are on the same side of P Official Ans. by NTA (C)

Sol. $P_1 + \lambda P_2 = 0$ $\Rightarrow (x + 3y - z - 5) + \lambda(2x - y + z - 3) = 0$ (2,1,-2) lies on this plane

 $\therefore \lambda = 1 \Longrightarrow plane \text{ is } 3x + 2y - 8 = 0$

10. A circle touches both the y-axis and the line x + y = 0. Then the locus of its center is

(A)
$$y = \sqrt{2}x$$
 (B) $x = \sqrt{2}y$

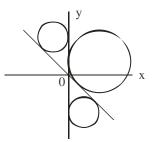
(C)
$$y^2 - x^2 = 2xy$$
 (D) $x^2 - y^2 = 2xy$

Sol. Let (h, k) is centre of circle

$$\left|\frac{\mathbf{h}-\mathbf{k}}{\sqrt{2}}\right| = \left|\mathbf{h}\right|$$

 $k^2 - h^2 + 2hk = 0$

 \therefore Equation of locus is $y^2 - x^2 + 2xy = 0$



11. Water is being filled at the rate of $1 \text{ cm}^3 / \text{sec}$ in a right circular conical vessel (vertex downwards) of height 35 cm and diameter 14 cm. When the height of the water level is 10 cm, the rate (in cm² / sec) at which the wet conical surface area of the vessel increases is

C)
$$\frac{\sqrt{26}}{5}$$
 (D) $\frac{\sqrt{26}}{10}$

Official Ans. by NTA (C)

Sol. From figure $\frac{r}{h} = \frac{7}{35} \Rightarrow h = 5r$ $\overrightarrow{f_{h}} = 1 \Rightarrow \frac{d}{dt} \left(\frac{\pi r^{2} h}{3} \right) = 1$ $\Rightarrow \frac{d}{dt} \left(\frac{5\pi}{3} r^{3} \right) = 1 \Rightarrow r^{2} \frac{dr}{dt} = \frac{1}{5\pi}$ Let we conical surface area = S $= \pi r \ell = \pi r \sqrt{h^{2} + r^{2}}$ $= \sqrt{26}\pi r^{2} \Rightarrow \frac{dS}{dt} = 2\sqrt{26}\pi r \frac{dr}{dt}$ When h = 10 then r = 2 $\Rightarrow \frac{dS}{dt} = \frac{2\sqrt{26}}{10}$

12. If
$$b_n = \int_0^{\frac{\pi}{2}} \frac{\cos^2 nx}{\sin x} dx$$
, $n \in N$, then
(A) $b_3 - b_2$, $b_4 - b_3$, $b_5 - b_4$ are in an A.P. with
common difference -2
(B) $\frac{1}{b_3 - b_2}$, $\frac{1}{b_4 - b_3}$, $\frac{1}{b_5 - b_4}$ are in an A.P. with
common difference 2
(C) $b_3 - b_2$, $b_4 - b_3$, $b_5 - b_4$ are in a G.P.
(D) $\frac{1}{b_3 - b_2}$, $\frac{1}{b_4 - b_3}$, $\frac{1}{b_5 - b_4}$ are in an A.P. with
common difference -2
Official Ans. by NTA (D)
Sol. $b_n = \int_{0}^{\frac{\pi}{2}} \frac{1 + \cos 2nx}{\sin x} dx$

$$b_{n+1} - b_n = \int_0^{\pi/2} \frac{\cos^2(n+1)x - \cos^2 nx}{\sin x} dx$$

= $\int_0^{\pi/2} \frac{-\sin(2n+1)x \sin x}{\sin x} dx$
= $\left(\frac{\cos(2n+1)x}{2n+1}\right)_0^{\pi/2} = \frac{-1}{2n+1}$
 $\frac{1}{b_3 - b_2}, \frac{1}{b_4 - b_3}, \frac{1}{b_5 - b_4}$ are in A.P. with c.d.= -2

13. If y = y(x) is the solution of the differential equation $2x^2 \frac{dy}{dx} - 2xy + 3y^2 = 0$ such that $y(e) = \frac{e}{3}$, then y(1) is equal to (A) $\frac{1}{3}$ (B) $\frac{2}{3}$

(C)
$$\frac{3}{2}$$
 (D) 3

Official Ans. by NTA (B)

Sol.
$$\frac{dy}{dx} - \frac{y}{x} = -\frac{3}{2} \left(\frac{y}{x}\right)^2 \qquad y = vx$$
$$\frac{dv}{v^2} = -\frac{3dx}{2x}$$
$$-\frac{1}{v} = -\frac{3}{2} \ell n |x| + C$$
$$-\frac{x}{y} = \frac{-3}{2} \ell n |x| + C$$
$$x = e, \ y = \frac{e}{3}$$
$$C = -\frac{3}{2}$$
When $x = 1, \ y = \frac{2}{3}$

14. If the angle made by the tangent at the point (x_0, y_0) on the curve $x = 12(t + \sin t \cos t)$,

y =
$$12(1 + \sin t)^2$$
, $0 < t < \frac{\pi}{2}$, with the positive x-axis
is $\frac{\pi}{3}$, then y₀ is equal to
(A) $6(3 + 2\sqrt{2})$ (B) $3(7 + 4\sqrt{3})$
(C) 27 (D) 48
Official Ans. by NTA (C)

Sol.
$$\frac{dy}{dx} = \frac{2(1+\sin t) \times \cos t}{1+\cos 2t}$$
$$\Rightarrow \frac{2(1+\sin t)\cos t}{2\cos^2 t} = \sqrt{3}$$
$$\Rightarrow t = \frac{\pi}{6}, y_0 = 27$$

15. The value of $2\sin(12^\circ) - \sin(72^\circ)$ is :

(A)
$$\frac{\sqrt{5}(1-\sqrt{3})}{4}$$
 (B) $\frac{1-\sqrt{5}}{8}$
(C) $\frac{\sqrt{3}(1-\sqrt{5})}{2}$ (D) $\frac{\sqrt{3}(1-\sqrt{5})}{4}$

Official Ans. by NTA (D)

Sol. $\sin 12^\circ + \sin 12^\circ - \sin 72^\circ$ $= \sin 12^\circ - 2\cos 42^\circ \sin 30^\circ$ $= \sin 12^\circ - \sin 48^\circ$ $= -2\cos 30^\circ \sin 18^\circ$ $= -2 \times \frac{\sqrt{3}}{2} \times \frac{\sqrt{5} - 1}{4}$ $= \frac{\sqrt{3}}{4}(1 - \sqrt{5})$

16. A biased die is marked with numbers 2,4, 8, 16, 32, 32 on its faces and the probability of getting a face with mark n is $\frac{1}{n}$. If the die is thrown thrice, then the probability, that the sum of the numbers obtained is 48, is

(A)
$$\frac{7}{2^{11}}$$
 (B) $\frac{7}{2^{12}}$
(C) $\frac{3}{2^{10}}$ (D) $\frac{13}{2^{12}}$

Official Ans. by NTA (D)

Sol.
$$P(n) = \frac{1}{n}$$

 $P(2) = \frac{1}{2}$ $P(8) = \frac{1}{8}$
 $P(4) = \frac{1}{4}$ $P(16) = \frac{1}{16}$
 $P(32) = \frac{2}{32}$
Possible cases
16, 16, 16 and 32, 8, 8
Probability $= \frac{1}{16^3} + \frac{2}{32} \times \frac{1}{8} \times \frac{1}{8} \times 3 = \frac{13}{16^3}$
17. The negation of the Boolean expression
 $((\sim q) \land p) \Rightarrow ((\sim p) \lor q)$ is logically equivalent to
(A) $p \Rightarrow q$ (B) $q \Rightarrow p$

(C)
$$\sim (p \Rightarrow q)$$
 (D) $\sim (q \Rightarrow p)$

Official Ans. by NTA (C)

Sol. $\sim p \lor q \equiv p \rightarrow q$ $\sim q \land p \equiv \sim (p \rightarrow q)$ Negation of $\sim (p \rightarrow q) \rightarrow (p \rightarrow q)$ is $\sim (p \rightarrow q) \land (\sim (p \rightarrow q))$ i.e. $\sim (p \rightarrow q)$

18. If the line y = 4 + kx, k > 0, is the tangent to the parabola $y = x - x^2$ at the point P and V is the vertex of the parabola, then the slope of the line through P and V is :

(A)
$$\frac{3}{2}$$
 (B) $\frac{26}{9}$
(C) $\frac{5}{2}$ (D) $\frac{23}{6}$

Official Ans. by NTA (C)

Sol. Slope of tangent at P = Slope of line AP

$$y'|_{P} = 1 - 2\alpha = \frac{\alpha - \alpha^2 - 4}{\alpha}$$

Solving $\alpha = -2 \Rightarrow P(-2, -6)$

Slope of PV =
$$\frac{5}{2}$$

A(0,4)
 $V(1/2, 1/4)$
 $P(\alpha, \alpha - \alpha^2)$
P($\alpha, \alpha - \alpha^2$)
N
19. The value of $\tan^{-1}\left(\frac{\cos\left(\frac{15\pi}{4}\right) - 1}{\sin\left(\frac{\pi}{4}\right)}\right)$ is equal to
(A) $-\frac{\pi}{4}$ (B) $-\frac{\pi}{8}$

(C)
$$-\frac{5\pi}{12}$$
 (D) $-\frac{4\pi}{9}$

Official Ans. by NTA (B)

Sol.
$$\tan^{-1}\left[\frac{\cos\left(4\pi - \frac{\pi}{4}\right) - 1}{\sin\frac{\pi}{4}}\right] \Rightarrow \tan^{-1}\left(\frac{\cos\frac{\pi}{4} - 1}{\sin\frac{\pi}{4}}\right)$$

 $\tan^{-1}\left(\frac{1 - \sqrt{2}}{1}\right) = -\frac{\pi}{8}$
20. The line $y = x + 1$ meets the ellipse $\frac{x^2}{4} + \frac{y^2}{2} = 1$

20. The line y = x + 1 meets the ellipse $\frac{4}{4} + \frac{5}{2} = 1$ at two points P and Q. If r is the radius of the circle with PQ as diameter then $(3r)^2$ is equal to (A) 20 (B) 12

(C) 11 (D) 8

Official Ans. by NTA (A)

Sol. Ellipse $x^2 + 2y^2 = 4$

Line y = x + 1Point of intersection

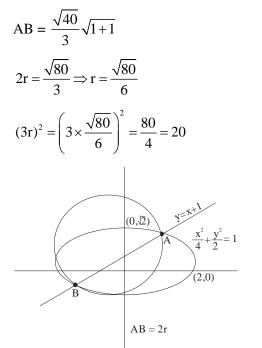
$$x^{2} + 2(x+1)^{2} = 4$$

$$3x^{2} + 4x - 2 = 0$$

$$|x_{1} - x_{2}| = \frac{\sqrt{40}}{3}$$

$$AB = 2r = |x_1 - x_2|\sqrt{1 + m^2}$$

m is slope of given line



SECTION-B

1. Let
$$A = \begin{pmatrix} 2 & -2 \\ 1 & -1 \end{pmatrix}$$
 and $B = \begin{pmatrix} -1 & 2 \\ -1 & 2 \end{pmatrix}$. Then the

number of elements in the set

$$\label{eq:and_states} \begin{split} \{(n,m)\,:\,n,m\in\{1,2,....,10\} \ \ and \ \ \ nA^n+mB^m=I\} \\ is ____ \end{split}$$

Official Ans. by NTA (1)

Sol.
$$A^2 = A$$
 and $B^2 = B$

Therefore equation $nA^n + mB^m = I$ becomes

nA + mB = I, which gives m = n = 1

Only one set possible

2. Let $f(x) = [2x^2 + 1]$ and $g(x) = \begin{cases} 2x - 3, & x < 0 \\ 2x + 3, & x \ge 0 \end{cases}$

where [t] is the greatest integer $\leq t$. Then, in the open interval (-1, 1), the number of points where fog is discontinuous is equal to _____

Official Ans. by NTA (62)

Sol.
$$f(g(x)) = [2g^{2}(x)] + 1$$

= $\begin{cases} [2(2x-3)^{2}] + 1; x < 0 \\ [2(2x+3)^{2}] + 1; x \ge 0 \end{cases}$

∴ fog is discontinuous whenever $2(2x-3)^2$ or $2(2x+3)^2$ belongs to integer except x = 0.

 \therefore 62 points of discontinuity.

3. The value of b > 3 for which

$$12\int_{3}^{b} \frac{1}{(x^{2}-1)(x^{2}-4)} dx = \log_{e}\left(\frac{49}{40}\right), \text{ is equal to}$$

Official Ans. by NTA (6)

Sol.
$$\frac{12}{3} \left[\int_{3}^{b} \left(\frac{1}{x^{2} - 4} - \frac{1}{x^{2} - 1} \right) dx \right] = \log \frac{49}{40}$$
$$\frac{12}{3} \left[\frac{1}{4} \ell n \left| \frac{x - 2}{x + 2} \right| - \frac{1}{2} \ell n \left| \frac{x - 1}{x + 1} \right| \right]_{3}^{b} = \log \frac{49}{40}$$
$$\ell n \frac{(b - 2)(b + 1)^{2}}{(b + 2)(b - 1)^{2}} = \ell n \frac{49}{50}$$
$$b = 6$$

4. If the sum of the coefficients of all the positive even powers of x in the binomial expansion of

$$\left(2x^3+\frac{3}{x}\right)^{10}$$
 is $5^{10}-\beta\cdot3^9$, then β is equal to _____

Official Ans. by NTA (83)

- Sol. $T_{r+1} = {}^{10} C_r (2x^3)^{10-r} \left(\frac{3}{x}\right)^r$ = ${}^{10} C_r 2^{10-r} 3^r x^{30-4r}$ Put r = 0, 1, 2, 7 and we get $\beta = 83$ 5. If the mean deviation about the mean of the
- numbers 1, 2, 3,, n, where n is odd, is $\frac{5(n+1)}{n}$, then n is equal to _____

Official Ans. by NTA (21)

- Sol. Mean deviation about mean of first n natural numbers is $\frac{n^2 - 1}{4n}$ \therefore n = 21
- 6. Let $\vec{b} = \hat{i} + \hat{j} + \lambda \hat{k}, \lambda \in \mathbb{R}$. If \vec{a} is a vector such that $\vec{a} \times \vec{b} = 13\hat{i} - \hat{j} - 4\hat{k}$ and $\vec{a} \cdot \vec{b} + 21 = 0$, then $(\vec{b} - \vec{a}) \cdot (\hat{k} - \hat{j}) + (\vec{b} + \vec{a}) \cdot (\hat{i} - \hat{k})$ is equal to Official Ans. by NTA (14)

Sol.
$$(\vec{a} \times \vec{b}).\vec{b} = 0$$

 $\Rightarrow 13 - 1 - 4\lambda = 0 \Rightarrow \lambda = 3$
 $\Rightarrow \vec{b} = \hat{i} + \hat{j} + 3\hat{k} \Rightarrow \vec{a} \times \vec{b} = 13\hat{i} - \hat{j} - 4\hat{k}$
 $\Rightarrow (\vec{a} \times \vec{b}) \times \vec{b} = (13\hat{i} - \hat{j} - 4\hat{k}) \times (\hat{i} + \hat{j} + 3\hat{k})$
 $\Rightarrow -21\vec{b} - 11\vec{a} = \hat{i} - 43\hat{j} + 14\hat{k}$
 $\Rightarrow \vec{a} = -2\hat{i} + 2\hat{j} - 7\hat{k}$

Now $(\vec{b} - \vec{a}) \cdot (\hat{k} - \hat{j}) + (\vec{b} + \vec{a}) \cdot (\hat{i} - \hat{k}) = 14$

7. The total number of three-digit numbers, with one digit repeated exactly two times, is

Official Ans. by NTA (243)

- Sol. If 0 taken twice then ways = 9 If 0 taken once then ${}^{9}C_{1} \times 2 = 18$ If 0 not taken then ${}^{9}C_{1} \, {}^{8}C_{1}.3 = 216$ Total = 243 8. Let $f(x) = |(x-1)(x^{2}-2x-3)| + x - 3, x \in \mathbb{R}$. If m
 - Let $\Gamma(x) = |(x-1)(x-2x-3)| + x-3$, $x \in \mathbb{R}$. If iff and M are respectively the number of points of local minimum and local maximum of f in the interval (0, 4), then m + M is equal to _____

Official Ans. by NTA (3)

Sol.
$$f(x) = \begin{cases} (x^2 - 1)(x - 3) + (x - 3), x \in (0, 1] \cup [3, 4) \\ -(x^2 - 1)(x - 3) + (x - 3), x \in [1, 3] \end{cases}$$

$$\Rightarrow f'(x) = \begin{cases} 3x^2 - 6x, x \in (0, 1) \cup (3, 4) \\ -3x^2 + 6x + 2, x \in (1, 3) \end{cases}$$

f(x) is non-derivable at x = 1 and x = 3

also f'(x) = 0 at x = 1 +
$$\sqrt{\frac{5}{3}} \implies m + M = 3$$

7

Sol.

9. Let the eccentricity of the hyperbola
$$\frac{x^2}{a^2} - \frac{y^2}{b^2} = 1$$

be $\frac{5}{4}$. If the equation of the normal at the point $\left(\frac{8}{\sqrt{5}}, \frac{12}{5}\right)$ on the hyperbola is $8\sqrt{5}x + \beta y = \lambda$, then $\lambda - \beta$ is equal to

Official Ans. by NTA (85)

 $e^{2} = 1 + \frac{b^{2}}{a^{2}} = \frac{25}{16} \Rightarrow \frac{b^{2}}{a^{2}} = \frac{9}{16}$ (1) Sol. $A\left(\frac{8}{\sqrt{5}},\frac{12}{5}\right)$ satisfies $\frac{x^2}{a^2}-\frac{y^2}{b^2}=1$ $\Rightarrow \frac{64}{5a^2} - \frac{144}{25b^2} = 1$ (2) Solving (1) & (2) $b = \frac{6}{5} a = \frac{8}{5}$ Normal at A is $\frac{\sqrt{5}a^2x}{8} + \frac{5b^2y}{12} = a^2 + b^2$ Comparing it $8\sqrt{5}x + \beta y = \lambda$ Gives $\lambda = 100$, $\beta = 15$ $\lambda - \beta = 85$ Let l_1 be the line in xy-plane with x and y 10. intercepts $\frac{1}{8}$ and $\frac{1}{4\sqrt{2}}$ respectively, and l_2 be the line in zx-plane with x and z intercepts $-\frac{1}{8}$ and

$$-\frac{1}{6\sqrt{3}}$$
 respectively. If d is the shortest distance
between the line l_1 and l_2 , then d⁻² is equal to
Official Ans. by NTA (51)

$$8x + 4\sqrt{2}y = 1, z = 0$$

$$\Rightarrow \frac{x - \frac{1}{8}}{1} = \frac{y - 0}{-\sqrt{2}} = \frac{z - 0}{0} = \lambda$$

$$-8x - 6\sqrt{3}z = 1, y = 0$$

$$\Rightarrow \frac{x + \frac{1}{8}}{3\sqrt{3}} = \frac{y - 0}{0} = \frac{z - 0}{-4}$$

$$\left| \frac{1}{4} \quad 0 \quad 0 \\ 1 \quad -\sqrt{2} \quad 0 \\ 3\sqrt{3} \quad 0 \quad -4 \right| = \sqrt{2}$$

$$d = \frac{1}{\sqrt{51}}$$

$$\frac{1}{d^2} = 51$$