## Question Paper with Solution

## PHYSICS _ 3 Sep. _ SHIFT - 1

1. A block of mass $m=1 \mathrm{~kg}$ slides with velocity $v=6 \mathrm{~m} / \mathrm{s}$ on a frictionless horizontal surface and collides with a uniform vertical rod and sticks to it as shown. The rod is pivoted about $O$ and swings as a result of the collision making angle $\theta$ before momentarily coming to rest. If the rod has mass $M=2 \mathrm{~kg}$, and length $\mathrm{I}=1 \mathrm{~m}$, the value of $\theta$ is approximately :
(take $\mathrm{g}=10 \mathrm{~m} / \mathrm{s}^{2}$ )

(1) $49^{\circ}$
(2) $63^{\circ}$
(3) $69^{\circ}$
(4) $55^{\circ}$

Sol. 2
$\mathrm{mv} \ell=\mathrm{I} \omega$
$\omega=\frac{\mathrm{mv} \ell}{\mathrm{I}}$
$\frac{1}{2} I \omega^{2}=(m+M) g \ell_{\text {com }}(1-\cos \theta)$
$\frac{1}{2} \frac{(m v \ell)^{2}}{I}=(m+M) g \ell_{\mathrm{com}}(1-\cos \theta)$
$I=\left(\frac{M \ell^{2}}{3}+m \ell^{2}\right)$
$\mathrm{I}=\left(\frac{2}{3}+1\right)=\frac{5}{3}$
$\frac{36 \times 3}{2 \times 5}=\frac{3 \times 10 \times 2}{3}(1-\cos \theta)$
$\frac{27}{50}=(1-\cos \theta)$
$\cos \theta=1-\frac{27}{50}$
$\cos \theta=\frac{23}{50}$
$\theta=63^{\circ}$
2. A uniform thin rope of length 12 m and mass 6 kg hangs vertically from a rigid support and a block of mass 2 kg is attached to its free end. A transverse short wavetrain of wavelength 6 cm is produced at the lower end of the rope. What is the wavelength of the wavetrain (in cm ) when it reaches the top of the rope ?
(1) 12
(2) 3
(3) 9
(4) 6

## Sol. 1


$\mu=\frac{6}{12}=\frac{1}{2} \mathrm{~kg} / \mathrm{m}$
$A \rightarrow V=\sqrt{\frac{T}{\mu}}=f \lambda$
$B \rightarrow \sqrt{\frac{T^{\prime}}{\mu}}=f \lambda^{\prime}$
$\sqrt{\frac{T}{T^{\prime}}}=\frac{\lambda}{\lambda^{\prime}}$
$\sqrt{\frac{20}{80}}=\frac{6}{\lambda^{\prime}}=\lambda^{\prime}=12$
3. When a diode is forward biased, it has a voltage drop of 0.5 V . The safe limit of current through the diode is 10 mA . If a battery of emf 1.5 V is used in the circuit, the value of minimum resistance to be connected in series with the diode so that the current does not exceed the safe limit is :
(1) $300 \Omega$
(2) $200 \Omega$
(3) $50 \Omega$
(4) $100 \Omega$

Sol. 4

$V=i R$
$R=\frac{V}{i}=\frac{1}{10 m A}$
$=\frac{1000}{10}=100 \Omega$
4. Using screw gauge of pitch 0.1 cm and 50 divisions on its circular scale, the thickness of an object is measured. It should correctly be recorded as :
(1) 2.125 cm
(2) 2.124 cm
(3) 2.123 cm
(4) 2.121 cm

Sol. 2
Least count $=\frac{0.1}{50}=\frac{1}{500}=0.2 \times 10^{-2}=0.002$
when we multiply by division no.'s
it must be even because L.C. is 0.002 .
5. Model a torch battery of length $\ell$ to be made up of a thin cylindrical bar of radius 'a' and a concentric thin cylindrical shell of radius ' $b$ ' filled in between with an electrolyte of resistivity $\rho$ (see figure). If the battery is connected to a resistance of value $R$, the maximum joule heating in $R$ will take place for :

(1) $R=\frac{\rho}{2 \pi I}\left(\frac{b}{a}\right)$
(2) $R=\frac{2 \rho}{\pi \mid} \ln \left(\frac{b}{a}\right)$
(3) $R=\frac{\rho}{\pi \mid} \ln \left(\frac{b}{a}\right)$
(4) $\frac{\rho}{2 \pi \mid} \ln \left(\frac{b}{a}\right)$

Sol. 4
$\mathrm{dR}=\rho \frac{\mathrm{dr}}{2 \pi \mathrm{r} \ell}$
$\int \mathrm{dR}=\frac{\rho}{2 \pi \ell} \int_{\mathrm{a}}^{\mathrm{b}} \frac{1}{\mathrm{r}} \mathrm{dr}$
$\mathrm{R}=\frac{\rho}{2 \pi \ell}[\ln (\mathrm{r})]_{\mathrm{a}}^{\mathrm{b}}$
$R=\frac{\rho}{2 \pi \ell} \ln \left(\frac{b}{a}\right)$

$r=R$
For Max heat transfer
6. Consider a gas of triatomic molecules. The molecules are assumed to be triangular and made of massless rigid rods whose vertices are occupied by atoms. The internal energy of a mole of the gas at temperature T is :

(1) $\frac{3}{2} R T$
(2) 3 RT
(3) $\frac{5}{2} R T$
(4) $\frac{9}{2} R T$

Sol. 2
$U=\frac{f}{2} n R T$
$U=\frac{6}{2} n R T$
$U=3 n R T \quad n=1$
$U=3 R T$
7. An elliptical loop having resistance $R$, of semi major axis $a$, and semi minor axis $b$ is placed in magnetic field as shown in the figure. If the loop is rotated about the $x$-axis with angular frequency $\omega$, the average power loss in the loop due to Joule heating is:


(1) $\frac{\pi a b B \omega}{\mathrm{R}}$
(2) $\frac{\pi^{2} a^{2} b^{2} B^{2} \omega^{2}}{R}$
(3) $\frac{\pi^{2} a^{2} b^{2} B^{2} \omega^{2}}{2 R}$
(4) Zero

Sol. 3
$\phi=B A \cos \omega t$
$\epsilon=\frac{\mathrm{d} \phi}{\mathrm{dt}}=\mathrm{BA} \omega \sin \omega \mathrm{t}$
$\epsilon=\mathrm{B} \pi \mathrm{ab} \omega \sin \omega \mathrm{t}$
$\epsilon_{0}=\mathrm{B} \pi \mathrm{ab} \omega$
$P=\frac{\epsilon^{2}}{R}=\frac{B^{2} \pi^{2} a^{2} b^{2} \omega^{2}}{R}\left(\sin ^{2} \omega t\right)$
$P_{a v}=\frac{B^{2} \pi^{2} a^{2} b^{2} \omega^{2}}{2 R}$
8. A balloon filled with helium ( $32^{\circ} \mathrm{C}$ and 1.7 atm.) bursts. Immediately afterwards the expansion of helium can be considered as :
(1) reversible isothermal
(2) irreversible isothermal
(3) reversible adiabatic
(4) irreversible adiabatic

## Sol. 4

irreversible adiabatic $\rightarrow$ Because Energy can not be restored.
9. When the wavelength of radiation falling on a metal is changed from 500 nm to 200 nm , the maximum kinetic energy of the photoelectrons becomes three times larger. The work function of the metal is close to :
(1) 1.02 eV
(2) 0.61 eV
(3) 0.52 eV
(4) 0.81 eV

Sol. 2
$K E_{\text {max }}=\frac{h c}{\lambda_{1}}-\phi$
$3 K E_{\text {max }}=\frac{h c}{\lambda_{2}}-\phi$
$3\left(\frac{\mathrm{hc}}{\lambda_{1}}-\phi\right)=\frac{\mathrm{hc}}{\lambda_{2}}-\phi$
$\frac{3 \mathrm{hc}}{\lambda_{1}}-\frac{\mathrm{hc}}{\lambda_{2}}=2 \phi$
$\frac{3 \times 1240}{500}-\frac{1240}{200}=2 \phi$
$\phi=0.6$
10. Two isolated conducting spheres $S_{1}$ and $S_{2}$ of radius $\frac{2}{3} R$ and $\frac{1}{3} R$ have $12 \mu C$ and $-3 \mu C$ charges, respectively, and are at a large distance from each other. They are now connected by a conducting wire. A long time after this is done the charges on $S_{1}$ and $S_{2}$ are respectively :
(1) $6 \mu \mathrm{C}$ and $3 \mu \mathrm{C}$
(2) $4.5 \mu \mathrm{C}$ on both
(3) $+4.5 \mu \mathrm{C}$ and $-4.5 \mu \mathrm{C}$
(4) $3 \mu \mathrm{C}$ and $6 \mu \mathrm{C}$

Sol. 1

$\mathrm{q}_{1}+\mathrm{q}_{2}=9 \mu \mathrm{C}$
$\frac{K q_{1}}{\frac{2}{3} R}=\frac{K q_{2}}{\frac{1}{3} R}$
$\mathrm{q}_{1}=2 \mathrm{q}_{2}$
$3 q_{2}=9 \mu \mathrm{c}$
$\mathrm{q}_{2}=3 \mu \mathrm{c}$
$\mathrm{q}_{1}=6 \mu \mathrm{c}$
11. In a radioactive material, fraction of active material remaining after time $t$ is $9 / 16$. The fraction that was remaining after $t / 2$ is :
(1) $\frac{3}{4}$
(2) $\frac{7}{8}$
(3) $\frac{4}{5}$
(4) $\frac{3}{5}$

Sol. 1
$N=N_{0} e^{-\lambda t}$
$\left(\frac{N}{N_{0}}\right)=e^{-\lambda t}$
$\frac{9}{16}=\mathrm{e}^{-\lambda \mathrm{t}}$
$\frac{\mathrm{N}}{\mathrm{N}_{0}}=\mathrm{e}^{-\mathrm{tt} / 2}$
$\frac{\mathrm{N}}{\mathrm{N}_{0}}=\left(\frac{9}{16}\right)^{\frac{1}{2}}=\frac{3}{4}$
12. Moment of inertia of a cylinder of mass $M$, length $L$ and radius $R$ about an axis passing through its centre and perpendicular to the axis of the cylinder is $I=M\left(\frac{R^{2}}{4}+\frac{L^{2}}{12}\right)$. If such a cylinder is to be made for a given mass of a material, the ratio $L / R$ for it to have minimum possible $I$ is :
(1) $\frac{2}{3}$
(2) $\frac{3}{2}$
(3) $\sqrt{\frac{2}{3}}$
(4) $\sqrt{\frac{3}{2}}$

Sol. 4
$\mathrm{M}=\mathrm{d} . \mathrm{V}$
$M=d \pi R^{2} L$
$I=M\left(\frac{R^{2}}{4}+\frac{L^{2}}{12}\right)$
$I=M\left(\frac{M}{4 d \pi L}+\frac{L^{2}}{12}\right)$
$I=\left(\frac{M^{2}}{4 d \pi L}+\frac{M L^{2}}{12}\right)$
$\frac{\mathrm{dI}}{\mathrm{dL}}=\frac{\mathrm{M}^{2}}{4 \mathrm{~d} \pi}\left(\frac{-1}{\mathrm{~L}^{2}}\right)+\frac{2 \mathrm{LM}}{12}=0$
$\frac{\mathrm{M}^{2}}{4 \mathrm{~d} \pi \mathrm{~L}^{2}}=\frac{2 \mathrm{LM}}{12}$
$\frac{d \pi R^{2} L}{4 d \pi L^{2}}=\frac{L}{6}$
$\frac{\mathrm{R}^{2}}{\mathrm{~L}^{2}}=\frac{2}{3}$
$\frac{\mathrm{R}}{\mathrm{L}}=\sqrt{\frac{2}{3}} \Rightarrow \frac{\mathrm{~L}}{\mathrm{R}}=\sqrt{\frac{3}{2}}$
13. A satellite is moving in a low nearly circular orbit around the earth. Its radius is roughly equal to that of the earth's radius $R_{e}$. By firing rockets attached to it, its speed is instantaneously increased in the direction of its motion so that it become $\sqrt{\frac{3}{2}}$ times larger. Due to this the farthest distance from the centre of the earth that the satellite reaches is $R$. Value of $R$ is :
(1) $2 R_{e}$
(2) $3 R_{e}$
(3) $4 R_{e}$
(4) $2.5 \mathrm{R}_{\mathrm{e}}$

## Sol. 2

$$
V_{0}=\sqrt{\frac{G M}{R_{e}}} \times \sqrt{\frac{3}{2}}
$$


$m V_{0} R_{e}=m V R$
$\sqrt{\frac{3}{2}} \sqrt{\frac{G M}{R_{e}}} R_{e}=V R$
$-\frac{\mathrm{GMm}}{\mathrm{R}_{\mathrm{e}}}+\frac{1}{2} m v_{0}^{2}=-\frac{\mathrm{GMm}}{\mathrm{R}}+\frac{1}{2} m v^{2}$
$-\frac{G M m}{R_{e}}+\frac{1}{2} m \frac{3}{2} \frac{G M}{R_{e}}=-\frac{G M m}{R}+\frac{1}{2} m \frac{3}{2} \frac{G M}{R_{e}} \frac{R_{e}^{2}}{R^{2}}$
$-\frac{1}{R_{e}}+\frac{3}{4 R_{e}}=-\frac{1}{R}+\frac{3}{4} \frac{R_{e}}{R^{2}}$
$-\frac{1}{4 R_{e}}=-\frac{1}{R}+\frac{3}{4} \frac{R_{e}}{R^{2}}$
By further calculating
$R=3 R_{e}$
14. Pressure inside two soap bubbles are 1.01 and 1.02 atmosphere, respectively. The ratio of their volumes is :
(1) $4: 1$
(2) $2: 1$
(3) $0.8: 1$
(4) $8: 1$

Sol. 4
$P_{\text {in }}=P_{0}+\frac{4 T}{R_{1}}$
$1.01=1+\frac{4 T}{R_{1}}$
$0.01=\frac{4 T}{R_{1}}$
$0.02=\frac{4 \mathrm{~T}}{\mathrm{R}_{2}}$
$\frac{1}{2}=\frac{R_{2}}{R_{1}}$
$\frac{\mathrm{V}_{1}}{\mathrm{~V}_{2}}=\frac{8}{1}$
15. In a Young's double slit experiment, light of 500 nm is used to produce an interference pattern. When the distance between the slits is 0.05 mm , the angular width (in degree) of the fringes formed on the distance screen is close to :
(1) $0.17^{\circ}$
(2) $0.07^{\circ}$
(3) $0.57^{\circ}$
(4) $1.7^{\circ}$

Sol. 3
$B=\frac{\lambda D}{d}$
$\theta=\frac{\beta}{D}=\left(\frac{\lambda}{d}\right)$
$\theta=\frac{500 \times 10^{-9}}{\left(0.05 \times 10^{-3}\right)}$
$\theta=\frac{5 \times 10^{-2}}{5}=\frac{5}{100 \times 5}$
$\theta^{\circ}=\frac{5}{100} \times \frac{180}{5 \times \pi}$
$\theta^{\circ}=0.57^{\circ}$
16. A $750 \mathrm{~Hz}, 20 \mathrm{~V}(\mathrm{rms})$ source is connected to a resistance of $100 \Omega$, an inductance of 0.1803 H and a capacitance of $10 \mu \mathrm{~F}$ all in series. The time in which the resistance (heat capacity $2 \mathrm{~J} /{ }^{\circ} \mathrm{C}$ ) will get heated by $10^{\circ} \mathrm{C}$. (assume no loss of heat to the surroundings) is close to :
(1) 245 s
(2) 365 s
(3) 418 s
(4) 348 s

Sol. 4
$\mathrm{f}=750 \mathrm{~Hz}, \mathrm{~V}_{\mathrm{rms}}=20 \mathrm{~V}, \mathrm{R}=100 \Omega, \mathrm{~L}=0.1803 \mathrm{H}$
$\mathrm{C}=10 \mu \mathrm{f}$
$\mathrm{S}=2 \mathrm{~J} /{ }^{\circ} \mathrm{C}$
$Z=\sqrt{R^{2}+\left(X_{L}-X_{C}\right)^{2}}=\sqrt{R^{2}+\left(2 \pi f L-\frac{1}{2 \pi f c}\right)^{2}}$
$|z|=\sqrt{(100)^{2}+\left(2 \times 3.14 \times 750 \times 0.1803-\frac{1}{2 \times 3.14 \times 750 \times 10^{-5}}\right)^{2}}$
$|Z|=834 \Omega$
In AC
$\mathrm{P}=\mathrm{V}_{\mathrm{rms}} \mathrm{i}_{\mathrm{rms}} \cos \phi$
$P=\left(V_{\text {rms }} \cdot \frac{V_{\text {rms }}}{|Z|} \cdot \frac{R}{|Z|}\right)$
$P=\left(\frac{V_{r m s}}{|Z|}\right)^{2} R$
$P=\left(\frac{20}{834}\right)^{2} \times 100=0.0575 \mathrm{~J} / \mathrm{S}$
$\mathrm{P} \times \mathrm{t}=\mathrm{S} \Delta \theta$
$t=\frac{2(10)}{0.0575}$
$\mathrm{t}=348 \mathrm{sec}$
17. Magnitude of magnetic field (in SI units) at the centre of a hexagonal shape coil of side $10 \mathrm{~cm}, 50$ turns and carrying current I (Ampere) in units of $\frac{\mu_{0} I}{\pi}$ is :
(1) $250 \sqrt{3}$
(2) $50 \sqrt{3}$
(3) $500 \sqrt{3}$
(4) $5 \sqrt{3}$

Sol. 3


$$
\begin{aligned}
& \tan 30=\frac{x}{d} \\
& d=\frac{x}{\tan 30} \\
& d=\frac{5 \times 10^{-2}}{\frac{1}{\sqrt{3}}}
\end{aligned}
$$

$$
d=5 \sqrt{3} \times 10^{-2}
$$

$\mathbf{B}=\frac{6 \times \mu_{0} \mathrm{IN}}{4 \pi d}\left(\sin \theta_{1}+\sin \theta_{2}\right)$
$B=\frac{\mu_{0} I}{4 \pi} \times \frac{50 \times 6}{5 \sqrt{3} \times 10^{-2}}(\sin 30+\sin 30)$
$B=\frac{10 \times 6 \times 100}{\sqrt{3} \times 4}\left(\frac{\mu_{0} I}{\pi}\right)$
$B=500 \sqrt{3}$
18. The magnetic field of a plane electromagnetic wave is
$\vec{B}=3 \times 10^{-8} \sin [200 \pi(y+c t)] \hat{i} T$
where $\mathrm{c}=3 \times 10^{8} \mathrm{~ms}^{-1}$ is the speed of light.
The corresponding electric field is:
(1) $\vec{E}=-9 \sin [200 \pi(y+c t)] \hat{k} V / m$
(2) $\vec{E}=9 \sin [200 \pi(y+c t)] \hat{k} V / m$
(3) $\vec{E}=-10^{-6} \sin [200 \pi(y+c t)] \hat{k} V / m$
(4) $\vec{E}=3 \times 10^{-8} \sin [200 \pi(y+c t)] \hat{k} V / m$

Sol. 1
$E=B C$
$E=3 \times 10^{-8} \times 3 \times 10^{8}$
$E=9$
$\vec{S}=\frac{\vec{E} \times \vec{B}}{r}$

$E=9 \sin [200 \pi(y+c t)(-\hat{k})]$
19. A charged particle carrying charge $1 \mu \mathrm{C}$ is moving with velocity $(2 \hat{i}+3 \hat{j}+4 \hat{k}) \mathrm{ms}^{-1}$. If an external magnetic field of $(5 \hat{i}+3 \hat{j}-6 \hat{k}) \times 10^{-3} \mathrm{~T}$ exists in the region where the particle is moving then the force on the particle is $\vec{F} \times 10^{-9} \mathrm{~N}$. The vector $\overrightarrow{\mathrm{F}}$ is :
(1) $-0.30 \hat{i}+0.32 \hat{j}-0.09 \hat{k}$
(2) $-3.0 \hat{i}+3.2 \hat{j}-0.9 \hat{k}$
(3) $-30 \hat{i}+32 \hat{j}-9 \hat{k}$
(4) $-300 \hat{i}+320 \hat{j}-90 \hat{k}$

Sol. 3
$\vec{F}=q(\vec{V} \times \vec{B})$
$\vec{F}=10^{-6}(2 \hat{i}+3 \hat{j}+4 \hat{k}) \times(5 \hat{i}+3 \hat{j}-6 \hat{k}) \times 10^{-3}$
$\vec{F}=(-30 \hat{i}+32 \hat{j}-9 \hat{k}) \times 10^{-9}$
20. In the circuit shown in the figure, the total charge is $750 \mu \mathrm{C}$ and the voltage across capacitor $\mathrm{C}_{2}$ is 20 V . Then the charge on capacitor $\mathrm{C}_{2}$ is :

(1) $650 \mu \mathrm{C}$
(2) $450 \mu \mathrm{C}$
(3) $590 \mu \mathrm{C}$
(4) $160 \mu \mathrm{C}$

Sol. 3

$\mathrm{Q}_{2}=\mathrm{CV}=8 \mu \mathrm{f} 20=160 \mu \mathrm{C}$
$\mathrm{Q}_{1}=750-160=590 \mu \mathrm{C}$
21. A person of 80 kg mass is standing on the rim of a circular platform of mass 200 kg rotating about its axis at 5 revolutions per minute (rpm). The person now starts moving towards the centre of the platform. What will be the rotational speed (in rpm) of the platform when the person reaches its centre $\qquad$ -
Sol. 9
$\mathrm{I}_{1} \omega_{1}=\omega_{2} \mathrm{I}_{2}$

$$
\left(\frac{\mathrm{MR}^{2}}{2}+m R^{2}\right) \omega_{1}=\omega_{2} \frac{M R^{2}}{2}
$$

$\left(1+\frac{2 \mathrm{mR}^{2}}{\mathrm{MR}^{2}}\right) \omega_{1}=\omega_{2}$
$\left(1+\frac{2 \times 80}{200}\right) \omega_{1}=\omega_{2}$
$\omega_{2}=\omega_{1} 1.8$
$2 \pi f_{2}=2 \pi f_{1} \times 1.8$
$\mathrm{f}_{2}=5 \times 1.8=9$
22. An observer can see through a small hole on the side of a jar (radius 15 cm ) at a point at height of 15 cm from the bottom (see figure). The hole is at a height of 45 cm . When the jar is filled with a liquid up to a height of 30 cm the same observer can see the edge at the bottom of the jar. If the refractive index of the liquid is $N / 100$, where $N$ is an integer, the value of $N$ is $\qquad$ .


Sol. 1.58

$\mu \sin i=1 \sin 45$
$\mu \frac{15}{\sqrt{1125}}=\frac{1}{\sqrt{2}}$
$\mu=\frac{\sqrt{1125}}{15 \sqrt{2}}$
$\mu=\sqrt{\frac{1125}{450}}$
$\mu=1.58$
$\mu=\frac{\mathrm{N}}{100}=1.58$
$\Rightarrow \mathrm{N}=158$
23. A cricket ball of mass 0.15 kg is thrown vertically up by a bowling machine so that it rises to a maximum height of 20 m after leaving the machine. If the part pushing the ball applies a constant force $F$ on the ball and moves horizotally a distance of 0.2 m while launching the ball, the value of $F($ in $N)$ is $\left(g=10 \mathrm{~ms}^{-2}\right)$ $\qquad$ -.
Sol. 150
$\mathrm{m}=0.15 \mathrm{~kg}$


$$
\begin{aligned}
\mu & =\sqrt{2 \mathrm{gh}} \\
\mu & =\sqrt{2 \times 10 \times 20} \\
\mu & =20 \mathrm{~m} / \mathrm{sec}
\end{aligned}
$$

$\mathrm{Fd}=\frac{1}{2} m v^{2}$
$F \times 0.2=\frac{1}{2} \times \frac{0.15}{100} \times 400$
$F=\frac{15 \times 200 \times 10}{100 \times 0.2}=150$

24. When a long glass capillary tube of radius 0.015 cm is dipped in a liquid, the liquid rises to a height of 15 cm within it. If the contact angle between the liquid and glass to close to $0^{\circ}$, the surface tension of the liquid, in milliNewton $\mathrm{m}^{-1}$, is
$\left[\rho_{\text {(liquid) }}=900 \mathrm{kgm}^{-3}, \mathrm{~g}=10 \mathrm{~ms}^{-2}\right.$ ] (Give answer in closest integer) $\qquad$ .
Sol. 101
$r=0.015$

$h=\frac{2 T}{\rho g r}$
$T=\frac{h \rho g r}{2}$
$\mathrm{T}=\frac{15 \times 10^{-2} \times 900 \times 10 \times 0.015 \times 10^{-2}}{2 \times 10^{3}}$
$\mathrm{T}=101 \mathrm{milli} \mathrm{N} / \mathrm{m}$
25. A bakelite beaker has volume capacity of 500 cc at $30^{\circ} \mathrm{C}$. When it is partially filled with $\mathrm{V}_{\mathrm{m}}$ volume (at $30^{\circ} \mathrm{C}$ ) of mercury, it is found that the unfilled volume of the beaker remains constant as temperature is varied. If $\gamma_{\text {(beaker) }}=6 \times 10^{-6}{ }^{\circ} \mathrm{C}^{-1}$ and $\gamma_{\text {(mercury) }}=1.5 \times 10^{-4}{ }^{\circ} \mathrm{C}^{-1}$, where $\gamma$ is the coefficient of volume expansion, then $\mathrm{V}_{\mathrm{m}}$ (in cc) is close to
Sol. 20
$\Delta V=v \gamma \Delta T$
$\mathrm{v}_{1} \gamma_{1}=\mathrm{v}_{2} \gamma_{2}$
$500 \mathrm{cc} \times 6 \times 10^{-6}=\mathrm{V}_{\mathrm{m}} \times 1.5 \times 10^{-4}$
$V_{m}=\frac{500 \times 6 \times 10^{-6}}{1.5 \times 10^{-4}}=\frac{30}{1.5}$
$V_{m}=20 c c$

## Question paper with Solution

## CHEMISTRY _ 3 Sep. _ SHIFT - 1

1. It is true that:
(1) A second order reaction is always a multistep reaction
(2) A first order reaction is always a single step reaction
(3) A zero order reaction is a multistep reaction
(4) A zero order reaction is a single step reaction

Sol. 3

## Factual

2. An acidic buffer is obtained on mixing :
(1) 100 mL of 0.1 M HCl and 200 mL of $0.1 \mathrm{M} \mathrm{CH}_{3} \mathrm{COONa}$
(2) 100 mL of 0.1 M HCl and 200 mL of 0.1 M NaCl
(3) 100 mL of $0.1 \mathrm{M} \mathrm{CH}_{3} \mathrm{COOH}$ and 100 mL of 0.1 M NaOH
(4) 100 mL of $0.1 \mathrm{M} \mathrm{CH}_{3} \mathrm{COOH}$ and 200 mL of 0.1 M NaOH

Sol. 1
$2 \mathrm{HCl}+\mathrm{CH}_{3} \mathrm{COO}^{-} \longrightarrow \mathrm{CH}_{3} \mathrm{COOH}+\mathrm{OH}^{-}$
1020
$\begin{array}{lll}\mathrm{X} & 10 & 10\end{array}$
L
Acidic buffer
3. The Kjeldahl method of Nitrogen estimation fails for which of the following reaction products?
(a)

(b)

(c)

(d)

(1) (a), (c) and (d)
(2) (b) and (c)
(3) (c) and (d)
(4) (a) and (d)

Sol. 3
(A)

(B)

(C)

(D)


Diazo compound and inorganic nitrogen can't be estimeted by kjeldal method.
4. If the boiling point of $\mathrm{H}_{2} \mathrm{O}$ is 373 K , the boiling point of $\mathrm{H}_{2} \mathrm{~S}$ will be :
(1) greater than 300 K but less than 373 K
(2) equal to 373 K
(3) more than 373 K
(4) less than 300 K

## Sol. 4

Less than 300 K (factual)
5. The complex that can show optical activity is :
(1) cis $-\left[\mathrm{CrCl}_{2}(\mathrm{ox})_{2}\right]^{3-}$ (ox = oxalate)
(2) trans - $\left[\mathrm{Fe}\left(\mathrm{NH}_{3}\right)_{2}(\mathrm{CN})_{4}\right]$
(3) trans $-\left[\mathrm{Cr}\left(\mathrm{Cl}_{2}\right)(\mathrm{ox})_{2}\right]^{3-}$
(4) cis $-\left[\mathrm{Fe}\left(\mathrm{NH}_{3}\right)_{2}(\mathrm{CN})_{4}\right]^{-}$

Sol. 1

cis $-\left[\mathrm{CrCl}_{2}(\mathrm{ox})_{2}\right]^{3-}(\mathrm{ox}=$ oxalate $)$

trans $-\left[\mathrm{Fe}\left(\mathrm{NH}_{3}\right)_{2}(\mathrm{CN})_{4}\right]^{-}$

trans $-\left[\mathrm{Cr}\left(\mathrm{Cl}_{2}\right)(\mathrm{ox})_{2}\right]^{3-}$

cis - $\left[\mathrm{Fe}\left(\mathrm{NH}_{3}\right)_{2}(\mathrm{CN})_{4}\right]^{-}$
6. Which one of the following compounds possesses the most acidic hydrogen?
(1) $\mathrm{H}_{3} \mathrm{C}-\mathrm{C} \equiv \mathrm{C}-\mathrm{H}$
(2)

(3)

(4)


Sol. 3

has most acidic hydrogen among given compound, this is due to strong -M effect of -CN group which stabilize -ve charge significantly.
7. Aqua regia is used for dissolving noble metals ( $\mathrm{Au}, \mathrm{Pt}$, etc.). The gas evolved in this process is :
(1) $\mathrm{N}_{2} \mathrm{O}_{3}$
(2) $\mathrm{N}_{2}$
(3) $\mathrm{N}_{2} \mathrm{O}_{5}$
(4) NO

Sol. 4
$\mathrm{Au}+\mathrm{HNO}_{3}+\mathrm{HCl} \rightarrow \mathrm{HAuCl}_{4}+\mathrm{NO}+\mathrm{H}_{2} \mathrm{O}$
$\mathrm{Pt}+\underbrace{\mathrm{HNO}_{3}+\mathrm{HCl}}_{\text {aquar regia }} \rightarrow \mathrm{H}_{2} \mathrm{PtCl}_{6}+\mathrm{NO}+\mathrm{H}_{2} \mathrm{O}$
8. The antifertilituy drug "Novestrol" can react with :
(1) $\mathrm{Br}_{2} /$ water; $\mathrm{ZnCl}_{2} / \mathrm{HCl} ; \mathrm{FeCl}_{3}$
(2) $\mathrm{Br}_{2} /$ water; $\mathrm{ZnCl}_{2} / \mathrm{HCl} ; \mathrm{NaOCl}$
(3) Alcoholic $\mathrm{HCN} ; \mathrm{NaOCl} ; \mathrm{ZnCl}_{2} / \mathrm{HCl}$
(4) $\mathrm{ZnCl}_{2} / \mathrm{HCl} ; \mathrm{FeCl}_{3}$; Alcoholic HCN

Sol. 1
Novestrol


It can reacts with $\mathrm{Br}_{2} /$ water due to presence of unsaturation, with $\mathrm{ZnCl}_{2} / \mathrm{HCl}$ due to -OH group and with $\mathrm{FeCl}_{3}$ due to phenol.
9. Which of the following compounds produces an optically inactive compound on hydrogenation?
(1)

(2)

(3)

(4)


Sol. 3

10. Of the species, $\mathrm{NO}^{2}, \mathrm{NO}^{+}, \mathrm{NO}^{2+}$ and $\mathrm{NO}^{-}$, the one with minimum bond strength is :
(1) $\mathrm{NO}^{-}$
(2) $\mathrm{NO}^{+}$
(3) $\mathrm{NO}^{2+}$
(4) NO

Sol. 1
B.O. $\mathrm{NO}^{-}=2$
$\mathrm{BO} \mathrm{NO}^{+}=3$
$\mathrm{BO} \mathrm{NO}{ }^{2+}=2.5$
$\mathrm{BO} \mathrm{NO}=2.5$
B. $O \propto \frac{1}{\text { B.L }}$
11. Glycerol is separated in soap industries by :
(1) Fractional distillation
(2) Distillation under reduced pressure
(3) Differential extraction
(4) Steam distillation

Sol. 2
conceptual
Glycerol is separated in soap industries by distillation under reduced pressure
12. Thermal power plants can lead to :
(1) Ozone layer depletion
(2) Blue baby syndrome
(3) Eutrophication
(4) Acid rain

Sol. 4
Refer enviornmental chemistry
It emits $\mathrm{CO}_{2}$ that combine with mositure of atmosphere and forms $\mathrm{H}_{2} \mathrm{CO}_{3}$ (carbonic acid)
13. Henry's constant (in kbar) for four gases $\alpha, \beta, \gamma$ and $\delta$ in water at 298 K is given below :

|  | $\alpha$ | $\beta$ | $\gamma$ | $\delta$ |
| :---: | :---: | :---: | :---: | :---: |
| $\mathrm{K}_{\mathrm{H}}$ | 50 | 2 | $2 \times 10^{-5}$ | 0.5 |

(density of water $=10^{3} \mathrm{~kg} \mathrm{~m}^{-3}$ at 298 K )
This table implies that :
(1) solubility of $\gamma$ at 308 K is lower than at 298 K
(2) The pressure of a 55.5 molal solution of $\delta$ is 250 bar
(3) $\alpha$ has the highest solubility in water at a given pressure
(4) The pressure of a 55.5 molal solutio of $\gamma$ is 1 bar

Sol. 1
$\mathrm{p}=\mathrm{K}_{\mathrm{H}} \mathrm{X}$ mol fraction of gas in liquid.
On increasing tamp, ' $\mathrm{K}_{\mathrm{H}}^{\prime}$ increases
Hence solubility $\downarrow$
therefore, option 1
14. The electronic spectrum of $\left[\mathrm{Ti}\left(\mathrm{H}_{2} \mathrm{O}\right)_{6}\right]^{3+}$ shows a single broad peak with a maximum at $20,300 \mathrm{~cm}^{-1}$. The crystal field stabillization energy (CFSE) of the complex ion, in $\mathrm{kJ} \mathrm{mol}^{-1}$, is :
( $1 \mathrm{~kJ} \mathrm{~mol}-1=83.7 \mathrm{~cm}^{-1}$ )
(1) 83.7
(2) 242.5
(3) 145.5
(4) 97

Sol. 4
$\left[\mathrm{Ti}\left(\mathrm{H}_{2} \mathrm{O}\right)_{6}\right]^{3+} \mathrm{Ti}^{3+} 3 \mathrm{~d}^{1}$ in octahedral field of ligend


CFSE $=-0.4 \Delta_{0}$
CFSE $=\frac{-0.4 \times 20300}{83.7}$
$=97 \mathrm{~kJ} \mathrm{~mol}$
15. The atomic number of the element unnilennium is:
(1) 109
(2) 102
(3) 119
(4) 108

Sol. 1
Unnilennium 109
16. An organic compound [A], molecular formula $\mathrm{C}_{10} \mathrm{H}_{20} \mathrm{O}_{2}$ was hydrolyzed with dilute sulphuric acid to give a carboxylic acid [B] and an alcohol [C]. Oxidation of [C] with $\mathrm{CrO}_{3}-\mathrm{H}_{2} \mathrm{SO}_{4}$ produced [B]. Which of the following strucutres are not possible for $[A]$ ?
(1) $\left(\mathrm{CH}_{3}\right)_{3}-\mathrm{C}-\mathrm{COOCH}_{2} \mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}$
(2)

(3)

(4)


Sol. 2

(A) $\downarrow \mathrm{H}_{2} \mathrm{O} / \mathrm{H}^{+}$

(C)
(B)

17. The mechanism of $S_{N} 1$ reaction is given as :
$R-X \underset{\text { Ion pair }}{R^{\oplus} X^{\Theta}} \rightarrow R^{\oplus} \| X^{\Theta} \xrightarrow{Y^{\Theta}} R-Y+X^{\Theta}$
Solvent Separated ion
pair
A student writes general characteristics based on the given mechanism as :
(a) The reaction is favoured by weak nucleophiles.
(b) $R^{\oplus}$ would be easily formed if the substituents are bulky.
(c) The reaction is accompanied by racemization.
(d) The reaction is favoured by non-polar solvents.

Which observations are correct?
(1) (a) and (b)
(2) (a), (b) and (c)
(3) (a) and (c)
(4) (b) and (d)

Sol. 2
Statement (a), (b) \& (c) are correct for $\mathrm{S}_{\mathrm{N}}{ }^{1}$ reaction mechanism.
18. Tyndall effect is observed when:
(1) The diameter of dispersed particles is much smaller than the wavelength of light used.
(2) The diameter of dispersed particles is much larger than the wavelength of light used.
(3) The refractive index of dispersed phase is greater than that of the dispersion medium.
(4) The diameter of dispersed particles is similar to the wavelenght of light used.

Sol. 4
Diameter of dispersed particles should not be much smaller than wavelength of light used. Refer topic surface chemistry
19. Let $\mathrm{C}_{\mathrm{NaCl}}$ and $\mathrm{C}_{\mathrm{BaSO}_{4}}$ be the conductances (in S ) measured for saturated aqueous solutions of NaCl and BaSO4, respectively, at a temperature T .
Which of the following is false?
(1) $\mathrm{C}_{\text {NaCl }}\left(T_{2}\right)>C_{\text {NaCl }}\left(T_{1}\right)$ for $T_{2}>T_{1}$
(2) $C_{\text {BasO }_{4}}\left(T_{2}\right)>C_{\text {BaSo }_{4}}\left(T_{1}\right)$ for $T_{2}>T_{1}$
(3) Ionic mobilities of ions from both salts increase with $T$.
(4) $\mathrm{C}_{\mathrm{NaCl}} \gg \mathrm{C}_{\mathrm{BaSO}_{4}}$ at a given T

Sol. 4
Ionic
$\mathrm{C}_{\mathrm{NaCl}} \gg \mathrm{C}_{\mathrm{BaSO}_{4}}$ at temp ${ }^{\prime} \mathrm{T}^{\prime}$
20. In a molecule of pyrophosphoric acid, the number of $\mathrm{P}-\mathrm{OH}, \mathrm{P}=\mathrm{O}$ and $\mathrm{P}-\mathrm{O}-\mathrm{P}$ bonds/moiety(ies) respectively are :
(1) 3, 3 and 3
(2) 4, 2 and 1
(3) 2, 4 and 1
(4) 4, 2 and 0

Sol. 2

$\mathrm{P}-\mathrm{OH}$ bonds $=4$
$\mathrm{P}=\mathrm{O}$ bonds $=2$
$P-O-P$ linkage $=1$
Ans. 4, 2, 1
21. The mole fraction of glucose $\left(\mathrm{C}_{6} \mathrm{H}_{12} \mathrm{O}_{6}\right)$ in an aqueous binary solution is 0.1 . The mass percentage of water in it, to the nearest integer, is $\qquad$ —.
Sol. $47 \%$
$\mathrm{X}_{\text {Glucose }}=0.1$
mass $\%$ of glucose $=\frac{0.1 \times 180}{0.1 \times 180+0.9 \times 18} \times 100$

$$
\begin{aligned}
& =\frac{1800}{18+16.2} \\
& =\frac{1800}{34.2} \% \\
& =52.63 \% \\
& =53 \%
\end{aligned}
$$

$\therefore$ mass $\%$ of $\mathrm{H}_{2} \mathrm{O}=47 \%$
22. The volume strength of $8.9 \mathrm{M} \mathrm{H}_{2} \mathrm{O}_{2}$ solution calculated at 273 K and 1 atm is $\qquad$ . $(R=0.0821 L$ atm K- $\mathrm{mol}^{-1}$ ) (rounded off ot the nearest integer)
Sol. 100
Vol. strength $=\frac{8.9}{2} \times \frac{0.821 \times 273}{1}$

$$
\begin{aligned}
& =99.73 \\
& =100
\end{aligned}
$$

23. An element with molar mass $2.7 \times 10^{-2} \mathrm{~kg} \mathrm{~mol}^{-1}$ forms a cubic unit cell with edge length 405 pm . If its density is $2.7 \times 10^{3} \mathrm{~kg} \mathrm{~m}^{-3}$, the radius of the element is approximately $\qquad$ $\times 10^{-12} \mathrm{~m}$ (to the nearest integer).
Sol. 143
Density $=\frac{\mathrm{Z} \times \mathrm{GMM}}{\mathrm{N}_{\mathrm{A}} \times \mathrm{a}^{3}}$
$2.7 \times 10^{3}=\frac{\mathrm{Z} \times 2.7 \times 10^{-2}}{6.023 \times 10^{23} \times\left(405 \times 10^{-12}\right)^{3}}$
$Z=6.023 \times 405 \times 405 \times 405 \times 10^{23-36+3+2}$
$Z=6.023 \times 405 \times 405 \times 405 \times 10^{-8}$
$Z=4$
FCC
$4 R=\sqrt{2} \times a$
$R=\frac{405}{2 \sqrt{2}} \times 10^{-12}=143.21 \times 10^{-12} \mathrm{~m}$
$=143$ ans
24. The total number of monohalogenated organic products in the following (including stereoisomers) reaction is $\qquad$ _.

A

$$
\xrightarrow[\text { (ii) } \mathrm{X}_{2} / \Delta]{\text { (i) } \mathrm{H}_{2} / \mathrm{Ni} / \Delta}
$$

(Simplest
optically
active
alkene)

Sol. 8

25. The photoelectric current from Na (Work function, $\mathrm{w}_{0}=2.3 \mathrm{eV}$ ) is stopped by the output voltage of the cell $\mathrm{Pt}(\mathrm{s}) \mathrm{H}_{2}(\mathrm{~g}, 1 \mathrm{Bar}) \mathrm{HCl}$ (aq. $\left.\mathrm{pH}=1\right)|\mathrm{AgCl}(\mathrm{s})| \mathrm{Ag}(\mathrm{s})$.
The pH of aq. HCl required to stop the photoelectric current form $\mathrm{K}\left(\mathrm{w}_{0}=2.25 \mathrm{eV}\right)$, all other conditions remaining the same, is $\qquad$ $\times 10^{-2}$ (to the nearest integer).
Given,
$2.303 \frac{R T}{\mathrm{~F}}=0.06 \mathrm{~V} ; \mathrm{E}_{\mathrm{AgCl}|\mathrm{Ag}| \mathrm{Cl}^{-}}^{0}=0.22 \mathrm{~V}$
Sol. 58

```
Energy of photon \(=2.3-\mathrm{E}_{\text {cell }}\{\) for Na\(\}\)
Energy of photon \(=2.25-\mathrm{E}_{\text {cell }}\{\) for K\(\}\)
\(\mathrm{E}_{\text {cell }}\left\{\right.\) for ' \(\left.\mathrm{Na}^{\prime}\right\}+0.05=\mathrm{E}_{\text {cell }}\left\{\right.\) for \(\left.{ }^{\text {cell }} \mathrm{K}^{\prime}\right\}\)
\(0.22+0.06 \log \left[\mathrm{H}^{+}\right]\left[\mathrm{Cl}^{-}\right]+0.05=0.22+0.06 \log \left[\mathrm{H}^{+}\right]\left[\mathrm{Cl}^{-}\right]\)
\(6 \log \left(10^{-2}\right)+5=6 \log \left[\mathrm{H}^{+}\right]\left[\mathrm{Cl}^{-}\right]\)
\(\log \left(10^{-12}\right)+\log \left(10^{5}\right)=\log \left\{\left[\mathrm{H}^{+}\right]\left[\mathrm{Cl}^{-}\right]\right\}^{6}\)
\(\left\{\left[\mathrm{H}^{+}\right]\left[\mathrm{Cl}^{-}\right]\right\}^{6}=10^{-7}\)
\(\left[\mathrm{H}^{+}\right]^{12}=10^{-7}\)
\(\mathrm{pH}=\frac{7}{12}=0.58\)
\(=58 \times 10^{-2}=58\) Ans
```


## Question Paper with Solution

## MATHEMATICS _ 3 Sep. _ SHIFT - 1

Q. 1 The value of $\left(2 .{ }^{1} P_{0}-3 \cdot{ }^{2} P_{1}+4 \cdot{ }^{3} P_{2}-\ldots .\right.$. up to $51^{\text {th }}$ term $)+\left(1!-2!+3!-\ldots \ldots\right.$ up to $51^{\text {th }}$ term $)$ is equal to:
(1) $1-51(51)$ !
(2) $1+(52)$ !
(3) 1
(4) $1+(51)$ !

Sol. 2
2. ${ }^{1} P_{0}=\lfloor 2$
3. ${ }^{2} P_{1}=\underline{3}$
4. ${ }^{3} P_{2}=\lfloor 4$
$(\underline{2}-\underline{3}+\underline{4}-\underline{5}+\ldots \ldots . \underline{5} 2)+(\underline{1}-\underline{2}+\underline{3}-\underline{4} \ldots \ldots+\underline{51})$
$=\boxed{52}+1$
Q. 2 Let $P$ be a point on the parabola, $y^{2}=12 x$ and $N$ be the foot of the perpendicular drawn from $P$ on the axis of the parabola. A line is now drawn through the mid-point $M$ of $P N$, parallel to its axis which meets the parabola at $Q$. If the $y$-intercept of the line $N Q$ is $\frac{4}{3}$, then:
(1) $P N=4$
(2) $M Q=\frac{1}{3}$
(3) $\mathrm{PN}=3$
(4) $M Q=\frac{1}{4}$

Sol. 4


Q (h, 3t) lie on
Parabola
$9 t^{2}=12 h$
$h=\frac{3 t^{2}}{4}$
$Q=\left(\frac{3 t^{2}}{4}, 3 t\right)$
Equation of NQ
$y=\frac{3 t}{\left(\frac{3 t^{2}}{4}-3 t^{2}\right)} \quad\left(x-3 t^{2}\right)$
$y=\frac{-4 t}{3 t^{2}}\left(x-3 t^{2}\right)$
put $x=0$
$y=\frac{-4}{3 t}\left(-3 t^{2}\right)=4 t$
$4 \mathrm{t}=\frac{4}{3} \Rightarrow \mathrm{t}=\frac{1}{3}$
$P N=6 t=6 \cdot \frac{1}{3}=2$
$M=\left[\frac{1}{3}, 1\right], Q\left[\frac{1}{12}, 1\right]$
$\mathrm{MQ}=\frac{1}{3}-\frac{1}{12}=\frac{1}{4}$
Q. 3 If $\Delta=\left|\begin{array}{ccc}x-2 & 2 x-3 & 3 x-4 \\ 2 x-3 & 3 x-4 & 4 x-5 \\ 3 x-5 & 5 x-8 & 10 x-17\end{array}\right|=A x^{3}+B x^{2}+C x+D$, then $B+C$ is equal to:
(1) 1
(2)-1
(3) -3
(4) 9

Sol. 3

$$
\begin{aligned}
& \left|\begin{array}{ccc}
x-2 & 2 x-3 & 3 x-4 \\
2 x-3 & 3 x-4 & 4 x-5 \\
3 x-5 & 5 x-8 & 10 x-17
\end{array}\right|=A x^{3}+B x^{2}+C x+D \\
& R_{2} \rightarrow R_{2}-2 R_{1}, R_{3} \rightarrow R_{3}-3 R_{1} \\
& \left|\begin{array}{ccc}
x-2 & 2 x-3 & 3 x-4 \\
1 & -x+2 & -2 x+3 \\
1 & -x+1 & x-5
\end{array}\right|
\end{aligned}
$$

$$
R_{3} \rightarrow R_{3}-R_{2}
$$

$\left|\begin{array}{ccc}x-2 & 2 x-3 & 3 x-4 \\ 1 & -x+2 & -2 x+3 \\ 0 & -1 & 3 x-8\end{array}\right|=A x^{3}+B x^{2}+C x+D$
$\Rightarrow-1[(3-2 x)(x-2)-(3 x-4)]+(3 x-8)[(x-2)(-x+2)-(2 x-3)]=A x^{3}+B x^{2}+C x+D$
$\Rightarrow 3 x-2 x^{2}-6+4 x-3 x+4+(3 x-8)\left[-x^{2}+4 x-4-2 x+3\right]=A x^{3}+B x^{2}+C x+D$
$A=-3, B=12, C=-15$
$B+C=-3$
Q. 4 The foot of the perpendicular drawn form the point $(4,2,3)$ to the line joining the points $(1,-2,3)$ and $(1,1,0)$ lies on the plane:
(1) $x-y-2 z=1$
(2) $x-2 y+z=1$
(3) $2 x+y-z=1$
(4) $x+2 y-z=1$

Sol. 3
$\vec{r}=(1,-2,3)+\lambda(0,3,-3)$

$\overrightarrow{\mathrm{pm}} \perp \overrightarrow{\mathrm{b}}$
$\overrightarrow{\mathrm{pm}} \cdot \overrightarrow{\mathrm{b}}=0$
$(-3,3 \lambda-4,-3 \lambda) \cdot(0,3,-3)=0$
$\Rightarrow 0+9 \lambda-12+9 \lambda=0 \Rightarrow \lambda=\frac{12}{18}=\frac{2}{3}$
$m=(1,0,1)$ are on $2 x+y-z=1$
Q. 5 If $y^{2}+\log _{e}\left(\cos ^{2} x\right)=y, x \in\left(-\frac{\pi}{2}, \frac{\pi}{2}\right)$, then
(1) $\left|y^{\prime}(0)\right|+\left|y^{\prime \prime}(0)\right|=1$
(2) $y^{\prime \prime}(0)=0$
(3) $\left|y^{\prime}(0)\right|+\left|y^{\prime \prime}(0)\right|=3$
(4) $\left|y^{\prime \prime}(0)\right|=2$

Sol. 4
$2 y y^{\prime}+2(-\tan x)=y^{\prime}$
diff. w.r.t.x
$2 y y^{\prime \prime}+2\left(y^{\prime}\right)^{2}-2 \sec ^{2} x=y^{\prime \prime}$
Put $x=0$ in given equation we get $y=0,1$
from (1) $x=0, y=0 \Rightarrow y^{\prime}(0)=0$
$x=0, y=1, \Rightarrow y^{\prime}(0)=0$
from (2) $x=0, y=0, y^{\prime}(0)=0 \Rightarrow y^{\prime \prime}(0)=-2$
$x=0, y=1, y^{\prime}(0)=0 \Rightarrow y^{\prime \prime}(0)=2$
$\left|y^{\prime \prime}(0)\right|=2$
Q. $62 \pi-\left(\sin ^{-1} \frac{4}{5}+\sin ^{-1} \frac{5}{13}+\sin ^{-1} \frac{16}{25}\right)$ is equal to:
(1) $\frac{5 \pi}{4}$
(2) $\frac{3 \pi}{2}$
(3) $\frac{7 \pi}{4}$
(4) $\frac{\pi}{2}$

Sol. 2
$2 \pi-\left[\tan ^{-1}\left(\frac{4}{3}\right)+\tan ^{-1}\left(\frac{5}{12}\right)+\tan ^{-1} \frac{16}{63}\right]$
$2 \pi-\tan ^{-1}\left(\frac{\frac{4}{3}+\frac{5}{12}}{1-\frac{4}{3} \cdot \frac{5}{12}}\right)-\tan ^{-1}\left(\frac{16}{63}\right)$
$\Rightarrow 2 \pi-\tan ^{-1}\left(\frac{48+15}{36-20}\right)-\tan ^{-1}\left(\frac{16}{63}\right)$
$\Rightarrow 2 \pi-\left[\tan ^{-1}\left(\frac{63}{16}\right)+\cot ^{-1}\left(\frac{63}{16}\right)\right]$
$\Rightarrow 2 \pi-\frac{\pi}{2}$
$=\frac{3 \pi}{2}$
Q. 7 A hyperbola having the transverse axis of length $\sqrt{2}$ has the same foci as that of the ellipse $3 x^{2}+4 y^{2}=12$, then this hyperbola does not pass through which of the following points ?
(1) $\left(\sqrt{\frac{3}{2}}, \frac{1}{\sqrt{2}}\right)$
(2) $\left(1,-\frac{1}{\sqrt{2}}\right)$
(3) $\left(\frac{1}{\sqrt{2}}, 0\right)$
(4) $\left(-\sqrt{\frac{3}{2}}, 1\right)$

Sol. 1

$$
\begin{aligned}
& \frac{x^{2}}{4}+\frac{y^{2}}{3}=1 \\
& b_{1}^{2}=a_{1}^{2}\left(1-e_{1}^{2}\right) \\
& 3=4\left(1-e_{1}^{2}\right)
\end{aligned}
$$

$e_{1}=\frac{1}{2}$
focus $=\left( \pm \mathrm{a}_{1} \mathrm{e}_{1}, 0\right)$
$=( \pm 1,0)$
Length of transverse axis $2 a_{2}=\sqrt{2} \rightarrow a_{2}=\frac{1}{\sqrt{2}}$
$\mathrm{a}_{2} \mathrm{e}_{2}=1$
$=\mathrm{e}_{2}=\sqrt{2}$
$b_{2}{ }^{2}=a_{2}{ }^{2}\left(e_{2}{ }^{2}-1\right)$
$\mathrm{b}_{2}{ }^{2}=\frac{1}{2}(2-1)=\frac{1}{2}$
equation of Hyperbola
$x^{2}-y^{2}=\frac{1}{2}$
Q. 8 For the frequency distribution:

Variate $(x):$
Frequency $(\mathrm{f}):$$\quad \begin{array}{llll}\mathrm{x}_{1} & \mathrm{x}_{2} & \mathrm{x}_{3} \ldots \mathrm{x}_{15} \\ \mathrm{f}_{1} & \mathrm{f}_{2} & \mathrm{f}_{3} \ldots . \mathrm{f}_{15}\end{array}$
where $0<x_{1}<x_{2}<x_{3}<\ldots<x_{15}=10$ and $\sum_{i=1}^{15} f_{i}>0$, the standard deviation cannot be:
(1) 1
(2) 4
(3) 6
(4) 2

Sol. 3
$\sigma^{2} \leq \frac{1}{4}(M-m)^{2}$
( $M=$ upper bound of value of any random variable,
$\mathrm{m}=$ Lower bound of value of any random variable)
$\sigma^{2} \leq \frac{1}{4}(10-0)^{2}$
$\sigma^{2}<25$
$-5<\sigma<5$
$\sigma \neq 6$
Q. 9 A die is thrown two times and the sum of the scores appearing on the die is observed to be a multiple of 4. Then the conditional probability that the score 4 has appeared atleast once is:
(1) $\frac{1}{3}$
(2) $\frac{1}{4}$
(3) $\frac{1}{8}$
(4) $\frac{1}{9}$

Sol. 4
Total Possibilities $=(1,3),(3,1),(2,2)$,
$(2,6),(6,2)(4,4)$
$(3,5),(5,3)(6,6)$
fav. $=1=(4,4)$
prob. $=\frac{1}{9}$
Q. 10 If the number of integral terms in the expansion of $\left(3^{\frac{1}{2}}+5^{\frac{1}{8}}\right)^{n}$ is exactly 33 , then the least value of $n$ is:
(1) 128
(2) 248
(3) 256
(4) 264

Sol. 3

$$
\begin{aligned}
& \mathrm{T}_{\mathrm{r}+1}={ }^{\mathrm{n}} \mathrm{C}_{\mathrm{r}}\left(3^{\frac{1}{2}}\right)^{\mathrm{n}-\mathrm{r}}\left(5^{\frac{1}{8}}\right)^{\mathrm{r}} \\
& \frac{\mathrm{n}-\mathrm{r}}{2} \rightarrow \mathrm{n}-\mathrm{r}=0,2,4,6,8, \ldots \ldots \\
& \quad \frac{r}{8} \rightarrow r=0,8,16,24 \ldots \ldots \ldots \ldots \ldots . \\
& \text { common } r=0,8,16,24 \ldots \ldots \ldots \\
& \text { no. of integral term }=33 . \\
& \mathrm{L}=0+(33-1) \times 8 \rightarrow L=32 \times 8 \\
& =256
\end{aligned}
$$

Q. $11 \int_{-\pi}^{\pi}|\pi-|x|| d x$ is equal to:
(1) $\pi^{2}$
(2) $\frac{\pi^{2}}{2}$
(3) $\sqrt{2} \pi^{2}$
(4) $2 \pi^{2}$

Sol. 1
$\int_{-\pi}^{\pi}|\pi-|x|| d x$
even function
$2 \int_{0}^{\pi}|\pi-x| d x$
$=2 \int_{0}^{\pi}(\pi-x) d x \Rightarrow 2\left[\pi x-\frac{x^{2}}{2}\right]_{0}^{\pi}$
$=2\left[\frac{\pi^{2}}{2}\right]=\pi^{2}$
Q. 12 Consider the two sets:
$A=\left\{m \in R\right.$ : both the roots of $x^{2}-(m+1) x+m+4=0$ are real $\}$ and $B=[-3,5)$.
Which of the following is not true ?
(1) $A-B=(-\infty,-3) \cup(5, \infty)$
(2) $A \cap B=\{-3\}$
(3) $B-A=(-3,5)$
(4) $A \cup B=R$

Sol. 1
$\mathrm{D} \geq 0$
$(m+1)^{2}-4(m+4) \geq 0$
$\Rightarrow m^{2}-2 m-15 \geq 0$
$(m-5)(m+3) \geq 0$
$m \in(-\propto,-3] \cup[5, \propto)$
$A=(-\infty,-3] \cup[5, \propto)$
$B=[-3,5)$
$A-B=(-\propto,-3) \cup[5, \propto)$
$A \cup B=R$
Q. 13 The proposition $\mathrm{p} \rightarrow \sim(\mathrm{p} \wedge \sim \mathrm{q})$ is equivalent to:
(1) $(\sim p) \vee(\sim q)$
(2) $(\sim p) \wedge q$
(3) $q$
(4) $(\sim p) \vee q$

Sol. 4
$\sim(p \wedge \sim q) \rightarrow \sim p \vee q$
$p \rightarrow(\sim p \vee q)$
$\Rightarrow \sim p \vee(\sim p \vee q)$
$\Rightarrow \sim p \vee q$
Q. 14 The function, $f(x)=(3 x-7) x^{2 / 3}, x \in R$ is increasing for all $x$ lying in:
(1) $\left(-\infty,-\frac{14}{15}\right) \cup(0, \infty)$
(2) $\left(-\infty, \frac{14}{15}\right)$
(3) $(-\infty, 0) \cup\left(\frac{14}{15}, \infty\right)$
(4) $(-\infty, 0) \cup\left(\frac{3}{7}, \infty\right)$

Sol. 3

$$
f(x)=(3 x-7) \cdot \frac{2}{3 x^{\frac{1}{3}}}+x^{\frac{2}{3}} \cdot 3
$$

$$
=\frac{6 x-14+9 x}{3 x^{\frac{1}{3}}}
$$

$=\frac{15 x-14}{3 x^{\frac{1}{3}}}$

$f(x)>0 \uparrow \Rightarrow x \in(-\infty, 0) \cup\left(\frac{14}{15}, \infty\right)$
Q. 15 If the first term of an A.P. is 3 and the sum of its first 25 terms is equal to the sum of its next 15 terms, then the common difference of this A.P. is:
(1) $\frac{1}{6}$
(B) $\frac{1}{5}$
(C) $\frac{1}{4}$
(D) $\frac{1}{7}$

Sol. 1
$a=3$
$\frac{25}{2}[2 a+24 d]=\frac{15}{2}[2 \times(a+25 d)+14 d]$
$\Rightarrow 50 \mathrm{a}+600 \mathrm{~d}=15[2 \mathrm{a}+50 \mathrm{~d}+14 \mathrm{~d}]$
$\Rightarrow 20 \mathrm{a}+600 \mathrm{~d}=960 \mathrm{~d}$
$\Rightarrow 60=360 \mathrm{~d}$
$d=\frac{1}{6}$
Q. 16 The solution curve of the differential equation, $\left(1+e^{-x}\right)\left(1+y^{2}\right) \frac{d y}{d x}=y^{2}$, which passes through the point $(0,1)$, is:
(1) $y^{2}=1+y \log _{e}\left(\frac{1+e^{-x}}{2}\right)$
(2) $y^{2}+1=y\left(\log _{e}\left(\frac{1+e^{-x}}{2}\right)+2\right)$
(3) $y^{2}+1=y\left(\log _{e}\left(\frac{1+e^{x}}{2}\right)+2\right)$
(4) $y^{2}=1+y\left(\log _{e}\left(\frac{1+e^{x}}{2}\right)\right)$

Sol. 4

$$
\begin{aligned}
& \int\left(\frac{1+y^{2}}{y^{2}}\right) d y=\int\left(\frac{1}{1+e^{-x}}\right) d x \\
& \int \frac{1}{y^{2}} d y+\int d y=\int\left(\frac{e^{x}}{e^{x}+1}\right) d x
\end{aligned}
$$

$\Rightarrow \frac{-1}{y}+y=\ln \left|e^{x}+1\right|+c$
$x=0, y=1$
$\Rightarrow-1+1=\ln 2+c \Rightarrow c=-\ln 2$
$\Rightarrow \frac{-1}{y}+y=\ln \left|e^{x}+1\right|-\ln 2$
$\Rightarrow \mathrm{y}^{2}=1+\mathrm{y}\left[\ln \left(\frac{\mathrm{e}^{\mathrm{x}}+1}{2}\right)\right]$
Q. 17 The area (in sq. units) of the region $\left\{(x, y): 0 \leq y \leq x^{2}+1,0 \leq y \leq x+1, \frac{1}{2} \leq x \leq 2\right\}$ is
(1) $\frac{23}{16}$
(2) $\frac{79}{16}$
(3) $\frac{23}{6}$
(4) $\frac{79}{24}$

Sol. 4


$$
\begin{aligned}
& A=\int_{\frac{1}{2}}^{1}\left(x^{2}+1\right) d x+\int_{1}^{2}(x+1) d x \\
& \left(\frac{x^{3}}{3}+x\right)_{\frac{1}{2}}^{1}+\left(\frac{x^{2}}{2}+x\right)_{1}^{2} \\
& =\left(\frac{1}{3}+1\right)-\left(\frac{1}{24}+\frac{1}{2}\right)+\left((2+2)-\left(\frac{3}{2}\right)\right) \\
& =\left(\frac{4}{3}-\frac{13}{24}\right)+\left(\frac{5}{2}\right)
\end{aligned}
$$

$=\left(\frac{32-13}{24}\right)+\left(\frac{5}{2}\right)=\frac{19+60}{24}=\frac{79}{24}$
Q. 18 If $\alpha$ and $\beta$ are the roots of the equation $x^{2}+p x+2=0$ and $\frac{1}{\alpha}$ and $\frac{1}{\beta}$ are the roots of the equation $2 \mathrm{x}^{2}+2 \mathrm{qx}+1=0$, then $\left(\alpha-\frac{1}{\alpha}\right)\left(\beta-\frac{1}{\beta}\right)\left(\alpha+\frac{1}{\beta}\right)\left(\beta+\frac{1}{\alpha}\right)$ is equal to :
(1) $\frac{9}{4}\left(9+p^{2}\right)$
(2) $\frac{9}{4}\left(9+q^{2}\right)$
(3) $\frac{9}{4}\left(9-p^{2}\right)$
(4) $\frac{9}{4}\left(9-q^{2}\right)$

Sol. 3
$\alpha+\beta=-p, \alpha \beta=2$
$\frac{1}{\alpha}+\frac{1}{\beta}=-q, \frac{1}{\alpha \beta}=\frac{1}{2}$
$\frac{\alpha+\beta}{\alpha \beta}=-q \Rightarrow \frac{-p}{2}=-q$
$\Rightarrow p=2 q$
$\left(\alpha+\frac{1}{\beta}\right)\left(\beta+\frac{1}{\alpha}\right)=\alpha \beta+\frac{1}{\alpha \beta}+2$
$=2+\frac{1}{2}+2=\frac{9}{2}$
$\left(\alpha-\frac{1}{\alpha}\right)\left(\beta-\frac{1}{\beta}\right)=\alpha \beta+\frac{1}{\alpha \beta}-\frac{\alpha}{\beta}-\frac{\beta}{\alpha}$
$=2+\frac{1}{2}-\left[\frac{\alpha^{2}+\beta^{2}}{\alpha \beta}\right]$
$=\frac{5}{2}-\left[\frac{(\alpha+\beta)^{2}-2 \alpha \beta}{\alpha \beta}\right]$
$=\frac{5}{2}-\left[\frac{\mathrm{p}^{2}-4}{2}\right]$
$=\frac{9-\mathrm{p}^{2}}{2}$

$$
\begin{aligned}
& \left(\alpha-\frac{1}{\alpha}\right)\left(\beta-\frac{1}{\beta}\right)\left(\alpha+\frac{1}{\beta}\right)\left(\beta+\frac{1}{\alpha}\right)=\left(\frac{9-\mathrm{p}^{2}}{2}\right)\left(\frac{9}{2}\right) \\
& =\frac{9}{4}\left(9-\mathrm{p}^{2}\right)
\end{aligned}
$$

Q. 19 The lines $\vec{r}=(\hat{i}-\hat{j})+1(2 \hat{i}+\hat{k})$ and $\vec{r}=(2 \hat{i}-\hat{j})+m(\hat{i}+\hat{j}-\hat{k})$
(1) do not intersect for any values of $I$ and $m$
(2) intersect when $\mathrm{I}=1$ and $\mathrm{m}=2$
(3) intersect when $\mathrm{I}=2$ and $\mathrm{m}=\frac{1}{2}$
(4) intersect for all values of I and m

Sol. 1
$\frac{2}{1} \neq \frac{0}{1} \neq \frac{1}{-1} \rightarrow$ lines are intersecting
$\vec{r}=(1+2 l) \hat{i}-\hat{j}+l \hat{k}$
$\vec{r}=(2+m) \hat{i}+(m-1) \hat{j}-m \hat{k}$
compare coff. of $\hat{i}, \hat{j}, \hat{k}$
$1+2 l=2+m \quad\left|\begin{array}{c}-1=m-1 \\ m=0\end{array}\right| \quad I=0$
Lines do not intersect
Q. 20 Let [ t$]$ denote the greatest integer $\leq \mathrm{t}$. if for some $\lambda \in \quad \mathrm{R}-\left\{\begin{array}{ll}0, & 1\end{array}\right\}$ $\lim _{x \rightarrow 0}\left|\frac{1-x+|x|}{\lambda-x+[x]}\right|=L$, then $L$ is equal to:
(1) 0
(2) 2
(3) $\frac{1}{2}$
(4) 1

Sol. 2
$\lim _{x \rightarrow 0}\left|\frac{1-x+|x|}{\lambda-x+[x]}\right|=L$
$\lim _{h \rightarrow 0}\left|\frac{1-h+h}{\lambda-h+[h]}\right|$
$\lim _{h \rightarrow 0}\left|\frac{1}{\lambda-h+0}\right|=\left|\frac{1}{\lambda}\right|[x]=0$
$\lim _{h \rightarrow 0}\left|\frac{1+h+h}{\lambda+h+[-h]}\right|$
$=\left|\frac{1}{\lambda-1}\right| \quad[-\mathrm{h}]=-1$
$\because|\lambda|=|\lambda-1|$
$\lambda^{2}=\lambda^{2}-2 \lambda+1 \Rightarrow \lambda=\frac{1}{2}$
$\mathrm{L}=2$
Q. 21 If $\lim _{x \rightarrow 0}\left\{\frac{1}{x^{8}}\left(1-\cos \frac{x^{2}}{2}-\cos \frac{x^{2}}{4}+\cos \frac{x^{2}}{2} \cos \frac{x^{2}}{4}\right)\right\}=2^{-k}$, then the vlaue of $k$ is $\ldots \ldots$.

Sol. 8
$\lim _{x \rightarrow 0} \frac{\left(1-\cos \frac{x^{2}}{2}\right)}{\left(\frac{x^{2}}{2}\right)^{2}} \frac{\left(1-\cos \frac{x^{2}}{4}\right)}{\left(\frac{x^{2}}{4}\right)^{2}} \cdot \frac{\left(\frac{x^{2}}{2}\right) \cdot\left(\frac{x^{2}}{4}\right)^{2}}{x^{8}}$
$\lim _{x \rightarrow 0} \frac{1}{4} \cdot \frac{1}{4} \cdot \frac{1}{16} \Rightarrow \frac{1}{256}=2^{-k}$
$2^{-8}=2^{-k} \Rightarrow k=8$
Q. 22 The diameter of the circle, whose centre lies on the line $x+y=2$ in the first quadrant and which touches both the lines $x=3$ and $y=2$, is ..........
Sol. 2


$$
\begin{aligned}
& p=r \\
& \text { for } y=2
\end{aligned}
$$

$$
r=\left|\frac{2-\alpha-2}{1}\right|=|\alpha|
$$

$$
\text { for } x=3
$$

$$
r=\left|\frac{\alpha-3}{1}\right|=|\alpha-3|
$$

$$
|\alpha|=|\alpha-3|
$$

$$
\Rightarrow \alpha^{2}+\alpha^{2}-6 \alpha+9 \Rightarrow \alpha=\frac{3}{2}
$$

$$
2 \alpha=3=2 r
$$

Q. 23 The value of $(0.16)^{\log _{2.5}\left(\frac{1}{3}+\frac{1}{3^{2}}+\frac{1}{3^{2}}+\ldots \ldots . . . \text {.to } \infty\right)}$ is equal to..............

Sol. 4

$$
\begin{aligned}
& \frac{1}{3}+\frac{1}{3^{2}}+\frac{1}{3^{3}}+\ldots \ldots \infty=\frac{\frac{1}{3}}{1-\frac{1}{3}}=\frac{1}{2} \\
& \log _{2.5}\left(\frac{1}{2}\right) \Rightarrow \log _{\frac{5}{2}} \frac{1}{2} \\
& .16=\frac{16}{100}=\frac{4}{25}=\left(\frac{2}{5}\right)^{2}
\end{aligned}
$$

$$
\begin{aligned}
& \Rightarrow\left(\frac{2}{5}\right)^{2 \log _{\frac{5}{2}} \frac{1}{2}}=\left(\frac{5}{2}\right)^{-2 \log _{\frac{5}{2}} \frac{1}{2}} \\
& \Rightarrow\left(\frac{5}{2}\right)^{\log _{\frac{5}{2}}\left(\frac{1}{2}\right)^{-2}} \\
& =4
\end{aligned}
$$

Q. 24 Let $A=\left[\begin{array}{ll}x & 1 \\ 1 & 0\end{array}\right], x \in R$ and $A^{4}=\left[a_{i j}\right]$. If $a_{11}=109$, then $a_{22}$ is equal to .............

Sol. 10

$$
\begin{aligned}
& A=\left[\begin{array}{ll}
x & 1 \\
1 & 0
\end{array}\right] \\
& A^{2}=\left[\begin{array}{ll}
x & 1 \\
1 & 0
\end{array}\right]\left[\begin{array}{ll}
x & 1 \\
1 & 0
\end{array}\right]=\left[\begin{array}{cc}
x^{2}+1 & x \\
x & 1
\end{array}\right] \\
& A^{3}=\left[\begin{array}{cc}
x^{2}+1 & x \\
x & 1
\end{array}\right]\left[\begin{array}{ll}
x & 1 \\
1 & 0
\end{array}\right] \\
& =\left[\begin{array}{cc}
x^{3}+x+x & x^{2}+1 \\
x^{2}+1 & x
\end{array}\right] \\
& A^{4}=\left[\begin{array}{cc}
x^{3}+2 x & x^{2}+1 \\
x^{2}+1 & x
\end{array}\right]\left[\begin{array}{ll}
x & 1 \\
1 & 0
\end{array}\right] \\
& {\left[\begin{array}{cc}
x^{4}+2 x^{2}+x^{2}+1 & x^{3}+2 x \\
x^{3}+x+x & x^{2}+1
\end{array}\right]} \\
& \begin{array}{l}
a_{11} \Rightarrow x^{4}+3 x^{2}+1=109 \\
x^{4}+3 x^{2}-108=0 \\
\Rightarrow\left(x^{2}+12\right)\left(x^{2}-9\right)=0 \\
x= \pm 3 \\
a_{11}=x^{2}+1=10
\end{array}
\end{aligned}
$$

Q. 25 If $\left(\frac{1+i}{1-i}\right)^{\frac{m}{2}}=\left(\frac{1+i}{i-1}\right)^{\frac{n}{3}}=1,(m, n \in N)$ then the greatest common divisor of the least values of $m$ and $n$
$\qquad$
Sol. 4

$$
\begin{aligned}
& {\left[\frac{(1+i)(1+i)}{(1+i)(1-i)}\right]^{\frac{m}{2}}=\left[\left(\frac{1+i}{-1+i}\right)\left(\frac{-1-i}{-1-i}\right)\right]^{\frac{n}{3}}=1} \\
& =\left(\frac{2 i}{2}\right)^{\frac{m}{2}}=1 \left\lvert\,\left(\frac{-1-i-i+1}{1+1}\right)^{\frac{n}{3}}=1\right. \\
& m=8
\end{aligned}
$$

greatest common divisor of $m \& n$ is 4

