## Question paper with Solution

## PHYSICS - 2 Sep 2020. - SHIFT - 1

Q. 1 The mass density of a spherical galaxy $K$ varies as $\frac{K}{r}$ over a large distance ' $r$ ' from its centre. In that region, a small star is in a circular orbit of radius $R$. Then the period of revolution, $T$ depends on $R$ as:
(1) $\mathrm{T}^{2} \propto \mathrm{R}$
(2) $\mathrm{T}^{2} \propto \mathrm{R}^{3}$
(3) $\mathrm{T}^{2} \propto \frac{1}{\mathrm{R}^{3}}$
(4) $\mathrm{T} \propto \mathrm{R}$

Sol. (1)


Mass of galaxy $=\int_{0}^{R} \rho d v$
$=\int_{0}^{R} \frac{k}{r} 4 \pi r^{2} d r$
$=4 \pi k \int_{0}^{R} r d r$
$M=\frac{4 \pi k R^{2}}{2}=k_{1} R^{2}$
$F=m \omega^{2} R$
$\frac{G M m}{R^{2}}=m \omega^{2} R$
$\frac{\mathrm{Gk}_{1} \mathrm{R}^{2}}{\mathrm{R}^{2}}=\omega^{2} \mathrm{R}$
$\therefore \omega^{2}=\frac{\mathrm{k}_{2}}{\mathrm{R}}$
$\omega=\sqrt{\frac{k_{2}}{R}}$
$\mathrm{T}=\frac{2 \pi}{\omega}=2 \pi \sqrt{\frac{\mathrm{R}}{\mathrm{k}_{2}}}$
$T=k_{3} \sqrt{R}$
$T^{2} \propto \mathrm{R}$
Q. 2 An amplitude modulated wave is represented by the expression $\mathrm{v}_{\mathrm{m}}=5(1+0.6 \cos 6280 \mathrm{t})$ $\sin \left(211 \times 10^{4} \mathrm{t}\right)$ volts. The minimum and maximum amplitudes of the amplitude modulated wave are, respectively :
(1) $\frac{3}{2} \mathrm{~V}, 5 \mathrm{~V}$
(2) $5 \mathrm{~V}, 8 \mathrm{~V}$
(3) $3 \mathrm{~V}, 5 \mathrm{~V}$
(4) $\frac{5}{2} \mathrm{~V}, 8 \mathrm{~V}$

Sol. (4)
$\frac{A_{m}}{A_{c}}=0.6$
$=(5+3 \cos 6280 t) \sin \left(211 \times 10^{4} t\right)$
maximum Amp. $=5+3=8 \mathrm{~V}$
minimum Amp. $=5-3=2 \mathrm{~V}$
from the given option nearest value of minimum Amplitude $=\frac{5}{2} \mathrm{~V}$
Q. 3 A spherical mirror is obtained as shown in the figure from a hollow glass sphere. If an object is positioned in front of the minor, what will be the nature and magnification of the image of the object? (Figure drawn as schematic and not to scale)

(1) Erect, virtual and unmagnified
(2) Inverted, real and magnified
(3) Erect, virtual and magnified
(4) Inverted, real and unmagnified

Sol. (4)

$\therefore$ beyond Ci.e. $-\infty<\mathrm{u}<\mathrm{C}$
$\therefore$ real, inverted
and unmagnified
Q. 4 A cylindrical vessel containing a liquid is rotated about its axis so that the liquid rises at its sides as shown in the figure. The radius of vessel is 5 cm an and the angular speed of rotation is $\omega$ rad s $\mathrm{s}^{-1}$. The difference in the height, $h$ (in cm ) of liquid at the centre of vessel and at the side will be :

10 cm
(3) $\frac{25 \omega^{2}}{2 g}$
(4) $\frac{2 \omega^{2}}{5 g}$
(1) $\frac{5 \omega^{2}}{2 g}$
(2) $\frac{2 \omega^{2}}{25 g}$

Sol. (3)


10 cm
$y=\frac{\omega^{2} x^{2}}{2 g}$
at $x=5 \mathrm{~cm}, \mathrm{y}=\mathrm{h}$
$h=\frac{\omega^{2}(5)^{2}}{2 g}=\frac{25 \omega^{2}}{2 g}$
Q. 5 If speed $V$, area $A$ and force $F$ are chosen as fundamental units, then the dimension of Young's modulus will be
(1) $\mathrm{FA}^{2} \mathrm{~V}^{-3}$
(2) $F A^{2} V^{-2}$
(3) $\mathrm{FA}^{-1} \mathrm{~V}^{0}$
(4) $F A^{2} V^{-1}$

Sol. (3)
$\mathrm{Y}=\mathrm{k}[\mathrm{F}]^{\mathrm{x}}[\mathrm{A}]^{\mathrm{y}}[\mathrm{V}]^{\mathrm{z}}$
$\left[\mathrm{ML}^{1} \mathrm{~T}^{-2}\right]=\left[\mathrm{MLT}^{-2}\right]^{x}\left[\mathrm{~L}^{2}\right]^{\mathrm{y}}\left[\mathrm{LT}^{-1}\right]^{\mathrm{z}}$
$\left[\mathrm{ML}^{1} \mathrm{~T}^{-2}\right]=\left[\mathrm{M}^{\times} \mathrm{L}^{\left.\mathrm{x}+2 \mathrm{y}+\mathrm{z} \mathrm{T}^{-2 x-z}\right]}\right.$
$x=1,-2 x-z=-2, x+2 y+z=-1$
$\Rightarrow z=0$
$\Rightarrow y=-1$
Q. 6 A bead of mass $m$ stays at point $P(a, b)$ on a wire bent in the shape of a parabola $y=4 C x^{2}$ and rotating with angular speed $\omega$ (see figure). The value of $\omega$ is (neglect friction):

(1) $\sqrt{\frac{2 g}{C}}$
(2) $2 \sqrt{g C}$
(3) $\sqrt{\frac{2 g C}{a b}}$
(4) $2 \sqrt{2 g C}$

Sol. (4)
$y=4 c x^{2}$
$\frac{d y}{d x}=8 c x$
$\tan \theta=\frac{m \omega^{2} a}{m g}$
$\tan \theta=\frac{d y}{d x}=8 c x$
$8 c x=\frac{\omega^{2} a}{g}$

$(x=a), 8 c a=\frac{\omega^{2} a}{g}$
$\sqrt{8 c g}=\omega$
$2 \sqrt{2 g c}=\omega$
Q. 7 Magnetic materials used for making permanent magnets ( $P$ ) and magnets in a transformer ( $T$ ) have different properties of the following, which property best matches for the type of magnet required ?
(1) P : Small retentivity, large coercivity
(2) P : Large retentivity, large coercivity
(3) T : Large retentivity, large coercivity
(4) T : Large retentivity, small coercivity

Sol. (2)
Permanent magnet must retain for long use and should not be easily demagnetised.
Q. 8 Interference fringes are observed on a screen by illuminating two thin slits 1 mm apart with a light source $(\lambda=632.8 \mathrm{~nm})$. The distance between the screen and the slits is 100 cm . If a bright fringe is observed on screen at a distance of 1.27 mm from the central bright fringe, then the path difference between the waves, which are reaching this point from the slits is close to :
(1) $2.05 \mu \mathrm{~m}$
(2) 2.87 nm
(3) 2 nm
(4) $1.27 \mu \mathrm{~m}$

## Sol. (4)

given, $\mathrm{d}=1 \mathrm{~mm}$
$\lambda=632.8 \mathrm{~nm}$
$D=100 \mathrm{~cm}$
$y=1.27 \mathrm{~mm}$
$\Delta \mathrm{X}=\mathrm{d} \sin \theta$
$\because(\theta=$ small $)$
$\Delta \mathrm{x}=\mathrm{d} \tan \theta$
$\Delta \mathrm{x}=\frac{\mathrm{dy}}{\mathrm{D}}=\frac{1 \mathrm{x} 10^{-3} \times 1.27 \times 10^{-3}}{100 \times 10^{-2}}$
$=1.27 \times 10^{-6} \mathrm{~m}$
$=1.27 \mu \mathrm{~m}$
Q. 9 A gas mixture consists of 3 moles of oxygen and 5 moles of argon at temperature T. Assuming the gases to be ideal and the oxygen bond to be rigid, the total internal energy (in units of RT) of the mixture is:
(1) 11
(2) 13
(3) 15
(4) 20

Sol. (3)
$\mathrm{U}=\mathrm{n}_{1} \mathrm{C}_{\mathrm{v}_{1}} \mathrm{~T}+\mathrm{n}_{2} \mathrm{C}_{\mathrm{v}_{2}} \mathrm{~T}$
$=3 \times \frac{5}{2} R T+5 \times \frac{3}{2} R T$
$=\frac{30}{2} R T=15 R T$
Q. 10 A plane electromagnetic wave, has frequency of $2.0 \times 10^{10} \mathrm{~Hz}$ and its energy density is $1.02 \times 10^{-8} \mathrm{~J} / \mathrm{m}^{3}$ in vacuum. The amplitude of the magnetic field of the wave is close
$\left(\frac{1}{4 \pi \varepsilon_{0}}=9 \times 10^{9} \frac{\mathrm{Nm}^{2}}{\mathrm{C}^{2}}\right.$ and speed of light $\left.=3 \times 10^{8} \mathrm{~ms}^{-1}\right):$
(1) 160 nT
(2) 150 nT
(3) 180 nT
(4) 190 nT

Sol. (1)
energy density $=\frac{B_{0}^{2}}{2 \mu_{0}}$
$\& C=\frac{1}{\sqrt{\mu_{0} \epsilon_{0}}}$
$\mu_{0}=\frac{1}{C^{2} \epsilon_{0}}$
$B=\sqrt{U \times 2 \mu_{0}}$
$=\sqrt{1.02 \times 10^{-8} \times 2 \times \frac{1}{9 \times 10^{16}} 4 \pi \times 9 \times 10^{9}}$
$=\sqrt{25.62 \times 10^{-15}}$
$\cong \sqrt{25600 \times 10^{-18}}$
$\cong 160 \times 10^{-9}$
$=160 \mathrm{nT}$
Q. 11 Consider four conducting materials copper, tungsten, mercury and aluminium with resistivity $\rho_{C^{\prime}} \rho_{T^{\prime}}$ $\rho_{m}$ and $\rho_{A}$ respectively. Then :
(1) $\rho_{C}>\rho_{A}>\rho_{T}$
(2) $\rho_{A}>\rho_{M}>\rho_{C}$
(3) $\rho_{A}>\rho_{T}>\rho_{C}$
(4) $\rho_{M}>\rho_{A}>\rho_{C}$

## Sol. (4)

(Theoretical concept)
Q. 12 A beam of protons with speed $4 \times 10^{5} \mathrm{~ms}^{-1}$ enters a uniform magnetic field of 0.3 T at an angle of $60^{\circ}$ to the magnetic field. The pitch of the resulting helical path of protons is close to : (Mass of the pr oton $=1.67 \times 10^{-27} \mathrm{~kg}$, charge of the proton $=1.69 \times 10^{-19} \mathrm{C}$ )
(1) 4 cm
(2) 2 cm
(3) 12 cm
(4) 5 cm

Sol. (1)
pitch $=\mathrm{V} \cos 60^{\circ}, T=\frac{\mathrm{V}}{2} \frac{2 \pi \mathrm{~m}}{\mathrm{eB}}$
$=4 \times 10^{5} \times \frac{1}{2} \times \frac{2 \pi}{0.3}\left(\frac{\mathrm{~m}}{\mathrm{e}}\right)$
$=\frac{4 \pi \times 10^{5} \times 10^{-8}}{0.3}$
$=\frac{4 \times 3.14 \times 10^{-3}}{3 \times 10^{-1}}$
$\simeq 4 \times 10^{-2} \mathrm{~m}$
$\approx 4 \mathrm{~cm}$
Q. 13 Two identical strings $X$ and $Z$ made of same material have tension $T_{x}$ and $T_{z}$ in them. If their fundamental frequencies are 450 Hz and 300 Hz , respectively, then the ratio $\mathrm{T}_{x} / \mathrm{T}_{z}$ is :
(1) 2.25
(2) 1.25
(3) 0.44
(4) 1.5

Sol. (1)
$f=\frac{1}{2 \mathrm{~L}} \sqrt{\frac{\mathrm{~T}}{\mu}}$
given, $\mu_{x}=\mu_{z} \& L_{x}=L_{z}$ as identical
$\therefore f \propto \sqrt{T}$
$\Rightarrow \frac{T_{x}}{\mathrm{~T}_{z}}=\frac{f_{x}^{2}}{f_{z}^{2}}=\left(\frac{450}{300}\right)^{2}=\left(\frac{3}{2}\right)^{2}=\frac{9}{4}$
$\frac{T_{x}}{T_{y}}=2.25$
Q. 14 A uniform cylinder of mass $M$ and radius $R$ is to be pulled over a step of height a ( $a<R$ ) by applying a force $F$ at its centre 'O' perpendicular to the plane through the axes of the cylinder on the edge of the step (see figure). The minimum value of $F$ required is

(1) $M g \sqrt{\left(\frac{R}{R-a}\right)^{2}-1}$
(2) $M g \sqrt{1-\frac{a^{2}}{R^{2}}}$
(3) $M g \frac{a}{R}$
(4) $M g \sqrt{1-\left(\frac{R-a}{R}\right)^{2}}$

Sol. (4)

$\cos \theta=\frac{\sqrt{R^{2}-(R-a)^{2}}}{R}$
$=\sqrt{\frac{R^{2}}{R^{2}}-\left(\frac{R-a}{R}\right)^{2}}$
$=\sqrt{1-\left(\frac{R-a}{R}\right)^{2}}$
to pull up, $\tau_{\mathrm{F}} \geq \tau_{\mathrm{mg}}$
$F R \geq m g \cos \theta R$
for $\min F, F_{\text {min }}=m g \cos \theta$
$F_{\text {min }}=m g \sqrt{1-\left(\frac{R-a}{R}\right)^{2}}$
Q. 15 In a reactor, 2 kg of ${ }_{92} \mathrm{U}^{235}$ fuel is fully used up in 30 days. The energy released per fission is 200 MeV . Given that the Avogadro number, $\mathrm{N}=6.023 \times 10^{26}$ per kilo mole and $1 \mathrm{eV}=1.6 \times 10^{-19} \mathrm{~J}$. The power output of the reactor is close to
(1) 60 MW
(2) 54 MW
(3) 125 MW
(4) 35 MW

Sol. (1)
$\mathrm{n}($ moles $)=\frac{2 \mathrm{~kg}}{235 \mathrm{gm}}=\frac{2000}{235}$
no. of nucleus $=N_{A} \times n$

$$
\begin{aligned}
& =6.022 \times 10^{23} \times \frac{2000}{235} \\
& =51.25 \times 10^{23}
\end{aligned}
$$

total energy released $=200 \times 51.25 \times 10^{23} \mathrm{MeV}$

$$
\begin{aligned}
& =102.5 \times 10^{25} \mathrm{MeV} \\
& =102.5 \times 10^{25} \times 10^{6} \times 1.6 \times 10^{-16} \mathrm{~J} \\
& =164 \times 10^{6} \mathrm{MJ}
\end{aligned}
$$

power $=\frac{164 \times 10^{6} \mathrm{MJ}}{30 \times 24 \times 60 \times 60 \mathrm{~S}}$

$$
\begin{aligned}
& =0.063 \times 10^{3} \mathrm{MW} \\
& \cong 60 \mathrm{MW}
\end{aligned}
$$

Q. 16 A charged particle (mass $m$ and charge $q$ ) moves along $X$ axis with velocity $V_{0}$. When it passes through the origin it enters a region having uniform electric field $\vec{E}=-E \hat{j}$ which extends upto $x=d$. Equation of path of electron in the region $x>d$ is :

(1) $y=\frac{q E d^{2}}{m V_{0}^{2}} x$
(2) $y=\frac{q E d}{m V_{0}^{2}}\left(\frac{d}{2}-x\right)$
(3) $y=\frac{q E d}{m V_{0}^{2}}(x-d)$
(4) $y=\frac{q E d}{m V_{0}^{2}} x$

Sol. (2)

$-y=\frac{1}{2} a t^{2}$
$-\mathrm{y}=\frac{1}{2} \frac{\mathrm{qE}}{\mathrm{m}} \mathrm{t}^{2}$
$\mathrm{X}=\mathrm{V}_{\mathrm{o}} \mathrm{t}$
$\Rightarrow \mathrm{t}=\frac{\mathrm{x}}{\mathrm{V}_{0}}$
for $x \leq d$,
$y=-\frac{1}{2} \frac{q E}{m} \frac{x^{2}}{V_{0}^{2}}$
$\left.\frac{d y}{d x}\right|_{x=d}=-\frac{1}{2} \frac{q E}{m} \times\left.\frac{2 x}{V_{0}^{2}}\right|_{x=d}$
Slope $=m=\tan \theta=-\frac{q E d}{\mathrm{mV}_{0}^{2}}$
equation of straight line,
$y=(\tan \theta) x+c$
$=-\left(\frac{q E d}{m v_{0}^{2}}\right) x+c$
(now for $C$, at $x=d, y=-\frac{q E d^{2}}{2 m v_{0}^{2}}$ put in (4)
$-\frac{q E d^{2}}{2 m v_{0}^{2}}=-\frac{q E d^{2}}{\mathrm{mV}_{0}^{2}}+\mathrm{c}$
$\Rightarrow \mathrm{c}=\frac{\mathrm{qEd}}{} \mathrm{m}^{2}$
for $x>d$, as no $\vec{E}$
$y=-\left(\frac{q E d}{m v_{0}^{2}}\right) x+\frac{q E d^{2}}{2 m v_{0}^{2}}$
$y=\frac{q E d}{m v_{0}^{2}}\left(\frac{d}{2}-x\right)$
Q. 17 Train A and train B are running on parallel tracks in the opposite directions with speeds of $36 \mathrm{~km} /$ hour and $72 \mathrm{~km} /$ hour, respectively. A person is walking in train $A$ in the direction opposite to its motion with a speed of $1.8 \mathrm{~km} /$ hour. Speed (in $\mathrm{ms}^{-1}$ ) of this person as observed from train B will be close to : (take the distance between the tracks as negligible)
(1) $29.5 \mathrm{~ms}^{-1}$
(2) $30.5 \mathrm{~ms}^{-1}$
(3) $31.5 \mathrm{~ms}^{-1}$
(4) $28.5 \mathrm{~ms}^{-1}$

Sol. (1)
$\overrightarrow{V_{m}}=\overrightarrow{V_{m / A}}+\overrightarrow{V_{A}}$
$=(-1.8 \hat{\mathrm{i}}+36 \hat{\mathrm{i}}) \mathrm{km} / \mathrm{h}$
$=\left(-1.8 \times \frac{5}{18}+36 \times \frac{5}{18}\right) \mathrm{m} / \mathrm{s}$
$=(-0.5 \hat{i}+10 \hat{i}) \mathrm{m} / \mathrm{s}$
$=9.5 \hat{i} \mathrm{~m} / \mathrm{s}$
$\overline{V_{m / B}}=\overrightarrow{V_{M}}-\overrightarrow{V_{B}}$
$=9.5 \hat{i}-(-20 \hat{i}) \mathrm{m} / \mathrm{s}$
$=29.5 \mathrm{~m} / \mathrm{s} \hat{\mathrm{i}}$

Q. 18 Shown in the figure is rigid and uniform one meter long rod $A B$ held in horizontal position by two strings tied to its ends and attached to the ceiling. The rod is of mass ' $m$ ' and has another weight of mass 2 m hung at a distance of 75 cm from $A$. The tension in the string at $A$ is:

(1) 0.75 mg
(2) 0.5 mg
(3) 1 mg
(4) 2 mg

Sol. (3)

$\tau_{\text {net }}$ about $B=0$
$\mathrm{T} \times 100 \mathrm{~cm}=\mathrm{mg} \times 50 \mathrm{~cm}+2 \mathrm{mg} \times 25 \mathrm{~cm}$
$\mathrm{T}=1 \mathrm{mg}$
Q. 19 The least count of the main scale of a vernier callipers is 1 mm . Its vernier scale is divided into 10 divisions and coincide with 9 divisions of the main scale. When jaws are touching each other, the $7^{\text {th }}$ division of vernier scale coincides with a division of main scale and the zero of vernier scale is lying right side of the zero of main scale. When this vernier is used to measure length of a cylinder the zero of the vernier scale between 3.1 cm and 3.2 cm and $4^{\text {th }}$ VSD coincides with a main scale division. The length of the cylinder is: (VSD is vernier scale division)
(1) 3.21 cm
(2) 3.07 cm
(3) 2.99 cm
(4) 3.2 cm

Sol. (2)
L.C. $=1 \mathrm{MSD}-1 \mathrm{VSD}$
L.C. $=0.1 \mathrm{MSD}$
$1 \mathrm{MSD}=1 \mathrm{~mm}$
L. C. $=0.1 \mathrm{~mm}$

+ ve zero error $=+7 \times$ L.C.

$$
=0.7 \mathrm{~mm}
$$

Reading $=(3.1 \mathrm{~cm}+4 \times$ L.C $)-$ zero error
$=3.1 \mathrm{~cm}+0.4 \mathrm{~mm}-0.7 \mathrm{~mm}$
$=3.1 \mathrm{~cm}-0.03 \mathrm{~cm}$ (as given $1 \mathrm{MSD}=1 \mathrm{~mm}$ )
$=3.07 \mathrm{~cm}$
Q. 20 A particle of mass $m$ with an initial velocity $u \hat{i}$ collides perfectly elastically with a mass 3 m at rest. It moves with a velocity $v \hat{j}$ after collision, then $v$ is given by:
(1) $v=\frac{1}{\sqrt{6}} u$
(2) $\mathrm{v}=\sqrt{\frac{2}{3}} \mathrm{u}$
(3) $v=\frac{u}{\sqrt{3}}$
(4) $v=\frac{u}{\sqrt{2}}$

Sol. (4)

(before collision)

$\overrightarrow{p_{i}}=\overrightarrow{p_{f}}$
$m u \hat{i}+0=m v \hat{j}+3 m \vec{V}_{2}$
$\frac{m u \hat{i}}{3 m}-\frac{m v \hat{j}}{3 m}=\vec{V}_{2}$
$\overrightarrow{V_{2}}=\frac{u}{3} \hat{i}-\frac{v}{3} \hat{j}$
now, $K \cdot E$ conserved as elastic collision
$\Sigma K E_{\mathrm{f}}=\Sigma K \mathrm{E}_{\mathrm{f}}$
$\frac{1}{2} m u^{2}=\frac{1}{2} m v^{2}+\frac{1}{2} 3 m\left(\frac{u^{2}}{9}+\frac{v^{2}}{9}\right)$
$\Rightarrow u^{2}=v^{2}+\frac{u^{2}}{3}+\frac{v^{2}}{3}$
$\frac{2}{3} u^{2}=\frac{4}{3} v^{2}$
$\Rightarrow \mathrm{v}=\frac{\mathrm{u}}{\sqrt{2}}$


A small block starts slipping down from a point $B$ on an inclined plane $A B$, which is making an angle $\theta$ with the horizontal section $B C$ is smooth and the remaining section $C A$ is rough with a coefficient of friction $\mu$. It is found that the block comes to rest as it reaches the bottom (point $A$ ) of the inclined plane. If $B C=2 A C$, the coefficient of friction is given by $\mu=k \tan \theta$. The value of $k$ is $\qquad$ -.
Sol. (3)

from work energy theorem
$\mathrm{W}_{\mathrm{g}}+\mathrm{W}_{\mathrm{f}}=\Delta \mathrm{kE}$
$\mathrm{mg} 3 \mathrm{x} \sin \theta-\mu \mathrm{mg} \cos \theta \mathrm{x}=0-0$
$\Rightarrow m g 3 x \sin \theta=\mu m g \cos \theta x$
$3 \tan \theta=\mu$
$k=3$
Q. 22 An engine takes in 5 moles of air at $20^{\circ} \mathrm{C}$ and 1atm, and compresses it adiabaticaly to $1 / 10^{\text {th }}$ of the original volume. Assuming air to be a diatomic ideal gas made up of rigid molecules, the change in its internal energy during this process comes out to be $X \mathrm{~kJ}$. The value of $X$ to the nearest integer is $\qquad$ .

Sol. (46)

$$
\begin{aligned}
& \mathrm{T}_{2} \mathrm{~V}_{2}^{\gamma-1}=\mathrm{T}_{1} \mathrm{~V}_{1}^{\gamma-1} \\
& \mathrm{~T}_{2}=\mathrm{T}_{1}\left(\frac{\mathrm{~V}_{1}}{\mathrm{~V}_{2}}\right)^{\gamma-1} \\
& =293\left(\frac{\mathrm{~V}}{\mathrm{~V} / 10}\right)^{\frac{7}{5}-1} \\
& \mathrm{~T}_{2}=293 \times(10)^{2 / 5} \\
& \Delta \mathrm{U}=\mathrm{nC}_{\mathrm{v}} \Delta \mathrm{~T}=5 \times \frac{5}{2} \mathrm{R}\left(293 \times 10^{\frac{2}{5}}-293\right)
\end{aligned}
$$

$$
\begin{aligned}
& =\frac{25}{2} \mathrm{R} \times 293\left(10^{\frac{2}{5}}-1\right)=\frac{25 \mathrm{R}}{2} \times 293(2.5-1) \\
& =\frac{25 \times 8.314 \times 293 \times 1.5}{2} \\
& =45675 \mathrm{~J}=46 \mathrm{~kJ}
\end{aligned}
$$

Q. 23 When radiation of wavelength $\lambda$ is used to illuminate a metallic surface, the stopping potential is $V$.

When the same surface is illuminated with radiation of wavelength $3 \lambda$, the stopping potential is $\frac{\mathrm{V}}{4}$.
If the threshold wavelength for the metallic surface is $n \lambda$ then value of $n$ will be $\qquad$ -
Sol. (9)
$\frac{\mathrm{hc}}{\lambda}=\phi+\mathrm{eV}$
$\frac{\mathrm{hc}}{3 \lambda}=\phi+\frac{\mathrm{eV}}{4}$
$\frac{\text { eq.(1) }}{\text { eq. (2) }} \quad 3=\frac{\phi+e V}{\phi+\frac{e V}{4}}$
$3 \phi+\frac{3 e V}{4}=\phi+e V$
$2 \phi=\frac{\mathrm{eV}}{4}$
$\phi=\frac{\mathrm{eV}}{8}$
$\frac{\mathrm{hc}}{\lambda}=\frac{\mathrm{eV}}{8}+e V$
$=\frac{9}{8} \mathrm{eV}$
$\therefore \mathrm{eV}=\frac{8}{9} \frac{\mathrm{hc}}{\lambda}$
so $\phi=\frac{h c}{\lambda}-\frac{8}{9} \frac{h c}{\lambda}$
$\phi=\frac{1}{9} \frac{h c}{\lambda}$
$\frac{\mathrm{hc}}{\lambda_{\mathrm{th}}}=\frac{\mathrm{hc}}{9 \lambda}$
$\therefore \lambda_{\text {th }}=9 \lambda$
Q. 24 A circular coil of radius 10 cm is placed in uniform magnetic field of $3.0 \times 10^{-5} \mathrm{~T}$ with its plane perpendicular to the field initially. It is rotated at constant angular speed about an axis along the diameter of coil and perpendicular to magnetic field so that it undergoes half of rotation in 0.2 s . The maximum value of EMF induced (in $\mu \mathrm{V}$ ) in the coil will be close to the integer $\qquad$ .
Sol. (15)
$\phi=\mathrm{BA} \cos \omega \mathrm{t}$
$E=\frac{-d \phi}{d t}=B A \omega \sin \omega t$
$E_{\max }=B A \omega \quad\left(\omega=\frac{\pi}{0.2}\right)$
$=3 \times 10^{-5} \times \pi \mathrm{R}^{2} \times \frac{\pi}{0.2}$
$=15 \times 10^{-6} \mathrm{~V}$
$=15 \mu \mathrm{~V}$
Q. 25 A $5 \mu \mathrm{~F}$ capacitor is charged fully by a 220 V supply. It is then disconnected from the supply and is connected in series to another uncharged $2.5 \mu \mathrm{~F}$ capacitor. If the energy change during the charge redistribution is $\frac{X}{100}$ J then value of $X$ to the nearest integer is $\qquad$ -.

## Sol. (40) Our Answer

NTA Answer 36
heat $=U_{i}-U_{f}$
$=\frac{1}{2} \frac{C_{1} C_{2}}{C_{1}+C_{2}}\left(V_{1}-V_{2}\right)^{2}$
$=\frac{1}{2} \frac{5 \times 2.5}{7.5}(220-0)^{2}$
$=\frac{5}{6} \times 220 \times 220 \times 10^{-6} \mathrm{~J}$
$=40,333.33 \times 10^{-6} \mathrm{~J}$
$\frac{x}{100}=0.4 \mathrm{~J}$
$x=40$

## Question Paper with Solution

## CHEMISTRY - 2 Sep 2020. - SHIFT - 1

1. The increasing order of the following compounds towards HCN addition is:

(i)

(ii)

(iii)

(iv)
(1) (iii) $<$ (i) $<$ (iv) $<$ (ii)
(2) (iii) < (iv) < (i) < (ii)
(3) (i) $<$ (iii) < (iv) < (ii)
(4) (iii) $<$ (iv) $<$ (ii) $<$ (i)

Sol. 1
In $\mathrm{HCN}, \mathrm{CN}^{-}$is acts as nucleophile, attack first that -CHO group which has maximum positive charge. The magnitude of the (+ve) charge increases by $-M$ and $-I$ group. So reactivity order will be


So, option (1) is correct answer.
2. Which of the following is used for the preparation of colloids?
(1) Van Arkel Method
(2) Ostwald Process
(3) Mond Process
(4) Bredig's Arc Method

Sol. 4
Bredig's Arc method
Chapter name surface chemistry
3. An open beaker of water in equilibrium with water vapour is in a sealed container. When a few grams of glucose are added to the beaker of water, the rate at which water molecules:
(1) leaves the vapour increases
(2) leaves the solution increases
(3) leaves the vapour decreases
(4) leaves the solution decreases

Sol. 1


> Backward shift vapours $\downarrow$

Hence Rate at which water molecules leaves the vap. increases.
4. For octahedral Mn (II) and tetrahedral $\mathrm{Ni}(\mathrm{II})$ complexes, consider the following statements:
(I) both the complexes can be high spin.
(II) Ni (II) complex can very rarely be low spin.
(III) with strong field ligands, Mn (II) complexes can be low spin.
(IV) aqueous solution of Mn (II) ions is yellow in colour.

The correct statements are:
(1) (I), (III) and (IV) only
(2) (I), (II) and (III) only
(3) (II), (III) and (IV) only
(4) (I) and (II) only

Sol. 2
$\mathrm{Mn}^{2+}$ [Ar]3d ${ }^{5}$ it can form low spin as well as high spin complex depending upon nature of ligand same of $\mathrm{Ni}^{2+}$ ion with coordination no 4. It can be $\mathrm{dsp}^{2}$ or $\mathrm{sp}^{3} \mathrm{i}$ :e low spin or high spin depending open nature of ligand.
5. The statement that is not true about ozone is:
(1) in the stratosphere, it forms a protective shield against UV radiation.
(2) in the atmosphere, it is depleted by CFCs.
(3) in the stratosphere, CFCs release chlorine free radicals $(\mathrm{Cl})$ which reacts with $\mathrm{O}_{3}$ to give chlorine dioxide radicals.
(4) it is a toxic gas and its reaction with NO gives $\mathrm{NO}_{2}$.

Sol. 3
$\dot{\mathrm{Cl}}+\mathrm{O}_{3} \longrightarrow \mathrm{Cl} \dot{\mathrm{O}}+\mathrm{O}_{2}$
Chlorine monoxide
Hence option (3)
6. Consider the following reactions:
(i) Glucose $+\mathrm{ROH} \xrightarrow{\text { dry } \mathrm{HCl}}$ Acetal $\xrightarrow[\left(\mathrm{CH}_{3} \mathrm{CO}\right)_{2} \mathrm{O}]{\text { xeq. of }}$ acetyl derivative
(ii) Glucose $\xrightarrow{\mathrm{Ni} / \mathrm{H}_{2}} A \xrightarrow[\left(\mathrm{CH}_{3} \mathrm{CO}\right)_{2} \mathrm{O}]{\text { eq.of }}$ acetyl derivative
(iii) Glucose $\xrightarrow[\left(\mathrm{CH}_{3} \mathrm{CO}\right)_{2} \mathrm{O}]{\text { zeq.of }}$ acetyl derivative
' $x$ ', ' $y$ ' and ' $z$ ' in these reactions are respectively.
(1) $4,5 \& 5$
(2) $5,4 \& 5$
(3) $5,6 \& 5$
(4) $4,6 \& 5$

Sol. 4
(i) Glucose $+\mathrm{ROH} \xrightarrow{\text { dry } \mathrm{HCl}}$

(ii) Glucose $\xrightarrow{{\mathrm{Ni} / \mathrm{H}_{2}}^{C}} \mathrm{CH}_{2} \mathrm{OH} \xrightarrow[\left(\mathrm{CH}_{3} \mathrm{CO}\right)_{2} \mathrm{O}]{6 \text { eq }}$ acetyl derivative

(iii) Glucose $\xrightarrow[\left(\mathrm{CH}_{3} \mathrm{CO}\right)_{2} \mathrm{O}]{5 \mathrm{e} . \text { or }}$ Acetyl derivative
$\left(\mathrm{CH}_{3} \mathrm{CO}\right)_{2} \mathrm{O}$ reacts with -OH group to form acetyl derivative, so as the no. of -OH group no. of eq.
of $\left(\mathrm{CH}_{3} \mathrm{CO}\right)_{2} \mathrm{O}$ will be used
So, $x=4$
$y=6$
$z=5$
So, option (4) will be correct answer.
7. The IUPAC name for the following compound is:

(1) 2,5-dimethyl-5-carboxy-hex-3-enal
(2) 2,5-dimethyl-6-oxo-hex-3-enoic acid
(3) 6-formyl-2-methyl-hex-3-enoic acid
(4) 2,5-dimethyl-6-carboxy-hex-3-enal

Sol. 2


2,5-Dimethyl-6-oxohex-3-enoic acid
8. For the following Assertion and Reason, the correct option is

Assertion (A): When Cu (II) and sulphide ions are mixed, they react together extremely quickly to give a solid.

Reason (R): The equilibrium constant of $\mathrm{Cu}^{2+}(\mathrm{aq})+\mathrm{S}^{2-}(\mathrm{aq}) \rightleftharpoons \mathrm{CuS}(\mathrm{s})$ is high because the solubility product is low.
(1) ( $\mathbf{A}$ ) is false and ( $\mathbf{R}$ ) is true.
(2) Both ( $\mathbf{A}$ ) and ( $\mathbf{R}$ ) are false.
(3) Both (A) and (R) are true but (R) is not the explanation for (A).
(4) Both ( $\mathbf{A}$ ) and ( $\mathbf{R}$ ) are true but ( $\mathbf{R}$ ) is the explanation for $(\mathbf{A})$.

Sol. 4
$(A)$ is $(B)$ true \&
$(R)$ is correct explanation of $(A)$
Ans. 4
9. Which one of the following graphs is not correct for ideal gas?

I

II

III

IV
$d=$ Density, $P=$ Pressure, $T=$ Temperature
(1) I
(2) IV
(3) III
(4) II

## Sol. 4

For ideal Gas

$$
d=\frac{P \times M}{R T}
$$

$\mathrm{d} v / \mathrm{s} \mathrm{T} \rightarrow$ Hyperbolic

$d v / s \frac{1}{T} \rightarrow$ St. line

$\mathrm{d} v / \mathrm{s} p \rightarrow$ St line

$\therefore \quad$ 'II' Graph is incorrect Ans (4)
10. While titrating dilute HCl solution with aqueous NaOH , which of the following will not be required?
(1) Bunsen burner and measuring cylinder
(2) Burette and porcelain tile
(3) Clamp and phenolphthalein
(4) Pipette and distilled water

Sol. 1
Bunsen Burner \& measuring cylinder are not Required. As titration is already on exothermic pro cess
Ans.(1)
11. In Carius method of estimation of halogen, 0.172 g of an organic compound showed presence of 0.08 g of bromine. Which of these is the correct structure of the compound?
(1) $\mathrm{H}_{3} \mathrm{C}-\mathrm{Br}$
(2)

(3)

(4) $\mathrm{H}_{3} \mathrm{C}-\mathrm{CH}_{2}-\mathrm{Br}$

## Sol. 3

carius method
mass $\%$ of $^{\prime} \mathrm{Br}^{\prime}=\frac{0.08}{0.172} \times 100=\frac{8000}{172}=46.51 \%$
option (1) mass $\%=\frac{80}{95} \times 100$
(2) mass $\%=\frac{2 \times 80 \times 100}{252}$
(3) mass $\%=\frac{1 \times 80 \times 100}{80+72+6+14}=\frac{8000}{172} \%$
(4) mass $\%=\frac{1 \times 80 \times 100}{109} \%$

Option (3) matches with the given mass percentage value Ans (3)
12. On heating compound (A) gives a gas (B) which is a constituent of air. This gas when treated with $\mathrm{H}_{2}$ in the presence of a catalyst gives another gas (C) which is basic in nature. (A) should not be:
(1) $\left(\mathrm{NH}_{4}\right)_{2} \mathrm{Cr}_{2} \mathrm{O}_{7}$
(2) $\mathrm{NaN}_{3}$
(3) $\mathrm{NH}_{4} \mathrm{NO}_{2}$
(4) $\mathrm{Pb}\left(\mathrm{NO}_{3}\right)_{2}$

Sol. 4
The gas (B) is $\mathrm{N}_{2}$ which is found in air
$\mathrm{N}_{2}+3 \mathrm{H}_{2} \xlongequal{\mathrm{Fe} / \mathrm{Mo}} 2 \mathrm{NH}_{3}$ (Haber's process)
(Basic in nature)
$\mathrm{NH}_{3}+\mathrm{H}_{2} \mathrm{O} \rightarrow \mathrm{NH}_{4} \mathrm{OH}$ (weak base)
$\left(\mathrm{NH}_{4}\right)_{2} \mathrm{Cr}_{2} \mathrm{O}_{7} \longrightarrow \mathrm{~N}_{2}+\mathrm{Cr}_{2} \mathrm{O}_{3}+\mathrm{H}_{2} \mathrm{O}$
$\mathrm{NaN}_{3} \longrightarrow \mathrm{~N}_{2}+\mathrm{Na}$
$\mathrm{NH}_{4} \mathrm{NO}_{2} \longrightarrow \mathrm{~N}_{2}+\mathrm{H}_{2} \mathrm{O}$
$\mathrm{Pb}\left(\mathrm{NO}_{3}\right)_{2} \longrightarrow \mathrm{PbO}+\mathrm{NO}_{2}+\mathrm{O}_{2}$
13. The major product in the following reaction is:

(1)

(2)

(3)

(4)


Sol. 3


Option (3) is correct answer.
14. In general, the property (magnitudes only) that shows an opposite trend in comparison to other properties across a period is:
(1) Ionization enthalpy
(2) Electronegativity
(3) Atomic radius
(4) Electron gain enthalpy

Sol. 3
Ionisation energy, electronegativity \& electron gain enthalpy increase across a period but atomic radius decreases
15. The figure that is not a direct manifestation of the quantum nature of atoms is:
(1)

(2)
Kinetic
energy of
photoelectrons
$\rightarrow$ Increasing wavelength

(3)
(4)


Sol. 4
Internal energy of 'Ar' or any gas, has nothing to do with Quantum nature of atom hence


Ans. option (4)
16. The major aromatic product $C$ in the following reaction sequence will be :

(1)

(2)

(3)

(4)


Sol. 3


Option (3) is correct answser.
17. Consider that a $d^{6}$ metal ion ( $M^{2+}$ ) forms a complex with aqua ligands, and the spin only magnetic moment of the complex is 4.90 BM. The geometry and the crystal field stabilization energy of the complex is:
(1) tetrahedral and $-0.6 \Delta_{t}$
(2) tetrahedral and $-1.6 \Delta_{t}+1 P$
(3) octahedral and $-1.6 \Delta_{0}$
(4) octahedral and $-2.4 \Delta_{0}+2 P$

Sol. 1

$\mu \operatorname{spin}=4.9 \mathrm{BM}$
CFSE $=-0.4 \times 4 \Delta_{0}+0.6 \times 2 \Delta_{0}$
$=[1.6+1.2] \Delta_{0}$
$=-0.4 \Delta_{0}$


$$
\begin{aligned}
\text { CFSE } & =-0.6 \times 3 \Delta_{\mathrm{t}}+0.4 \times 3 \Delta_{\mathrm{t}} \\
& =-1.8 \Delta_{\mathrm{t}}+1.2 \Delta_{\mathrm{t}} \\
& =-0.6 \Delta_{\mathrm{t}}
\end{aligned}
$$

18. If $A B_{4}$ molecule is a polar molecule, a possible geometry of $A B_{4}$ is:
(1) Square planar
(2) Tetrahedral
(3) Square pyramidal
(4) Rectangular planar

Sol. 1
Incorrect question Option 1 is more appropriate with respect to given option
(Chemical bonding)
(Options are incorrect)
19. Which of the following compounds will show retention in configuration on nucleophilic substitution by $\mathrm{OH}^{-}$ion?
(1)

(2)

(3)

(4)


Sol. 1
In $\mathrm{CH}_{3}-\mathrm{C}_{\mathrm{C}}^{\mathrm{CH}}-\mathrm{CH}_{2} \mathrm{Br}$ attack of $\mathrm{OH}^{-}$is not on chiral carbon, it is adjacent to chiral carbon, so configuration of chiral carbon remains constant.
20. The metal mainly used in devising photoelectric cells is:
(1) Li
(2) Cs
(3) Rb
(4) Na

Sol. 2
'Cs' is used in photoelectric cell as its ionisation energy is lowest Hence Ans (2)
21. The mass of gas adsorbed, $x$, per unit mass of adsorbate, $m$, was measured at various pressures, $p$. A graph between $\log \frac{x}{m}$ and $\log p$ gives a straight line with slope equal to 2 and the intercept equal to 0.4771 . The value of $\frac{x}{m}$ at a pressure of 4 atm is: (Given $\log 3=0.4771$ )
Sol. 48
$\frac{\mathrm{x}}{\mathrm{m}}=K \mathrm{P}^{1 / \mathrm{n}}$
$\log (x / m)=\log _{(k)}+\frac{1}{n} \log (p)$
$y=c+m x$
Intercept $C=\log _{k}=0.4771$
slop $=\frac{1}{n}=2, k=3$
$\begin{aligned} \frac{x}{m} & =k(P)^{1 / n} \\ & =3(4)^{2}\end{aligned} \quad$ at $P=4 \mathrm{~atm}$
$\frac{x}{m}=3 \times 16=48$ Ans
22. The Gibbs energy change (in J) for the given reaction at $\left[\mathrm{Cu}^{2+}\right]=\left[\mathrm{Sn}^{2+}\right]=1 \mathrm{M}$ and 298 K is:
$\mathrm{Cu}(\mathrm{s})+\mathrm{Sn}^{2+}$ (aq.) $\rightarrow \mathrm{Cu}^{2+}$ (aq.) $+\mathrm{Sn}(\mathrm{s})$
$\left(\mathrm{E}_{\mathrm{Sn}^{2+} \mid \mathrm{Sn}}^{\circ}=-0.16 \mathrm{~V}, \mathrm{E}_{\mathrm{Cu}^{2+} \mid \mathrm{Cu}}^{0}=0.34 \mathrm{~V}\right.$, Take $\left.\mathrm{F}=96500 \mathrm{C} \mathrm{mol}^{-1}\right)$
Sol. 96500
$\mathrm{Cu}(\mathrm{s})+\mathrm{Sn}^{+2}(\mathrm{aq}) \rightleftharpoons \mathrm{Cu}^{+2}(\mathrm{aq})+\mathrm{Sn}(\mathrm{s})$
$\mathrm{E}_{\text {cell }}=-0.16-0.34$
$=-0.50$
$\Delta \mathrm{G}^{0}=-\mathrm{nFE} \mathrm{E}^{0}$
$=-2 \times 96500 \times(-0.5)$
$=+96500$
$\Delta \mathrm{G}=\Delta \mathrm{G}^{0}+\mathrm{RT} \ell \mathrm{n} \mathrm{Q}$
$=96500+\frac{25}{3} \times 298 \times 2.303 \log (1)$
$\Delta \mathrm{G} \quad=96500$ Joules
23. The internal energy change (in J) when 90 g of water undergoes complete evaporation at $100^{\circ} \mathrm{C}$ is
$\qquad$ .
(Given: $\Delta \mathrm{H}_{\text {vap }}$ for water at $373 \mathrm{~K}=41 \mathrm{~kJ} / \mathrm{mol}, \mathrm{R}=8.314 \mathrm{JK}^{-1} \mathrm{~mol}^{-1}$ )
Sol. $\mathrm{H}_{2} \mathrm{O}(\ell) \longrightarrow \mathrm{H}_{2} \mathrm{O}(\mathrm{g})$

$$
\begin{aligned}
\Delta \mathrm{E}_{\text {vap }} & =\Delta \mathrm{H}_{\text {vap }}-\Delta \mathrm{ngRT} \\
& =41000 \times 5-5 \times 8.314 \times 373 \\
& =189494.39
\end{aligned}
$$

24. The oxidation states of iron atoms in compounds ( $A$ ), ( $B$ ) and ( $C$ ), respectively, are $x, y$ and $z$. The sum of $x, y$ and $z$ is $\qquad$ .
$\underset{(\mathrm{A})}{\mathrm{Na}_{4}\left[\mathrm{Fe}(\mathrm{CN})_{5}(\mathrm{NOS})\right]}$
$\mathrm{Na}_{4}\left[\mathrm{FeO}_{4}\right]$
(B)
$\left[\mathrm{Fe}_{2}(\mathrm{CO})_{9}\right]$
(C)

Sol. 6
$\mathrm{Na}_{4}\left[\mathrm{Fe}^{+2}(\mathrm{CN})_{5}(\mathrm{NOS})\right]$
$\mathrm{Na}_{4}\left[\mathrm{Fe}^{+4} \mathrm{O}_{4}\right]$
$\left[\mathrm{Fe}_{2}{ }^{\circ}(\mathrm{CO})_{9}\right]$
25. The number of chiral carbons present in the molecule given below is $\qquad$ .


## Sol. 5



## Question Paper with Solution

## MATHEMATICS - 2 Sep 2020-SHIFT - 1

Q. 1 A line parallel to the straight line $2 x-y=0$ is tangent to the hyperbola $\frac{x^{2}}{4}-\frac{y^{2}}{2}=1$ at the point $\left(x_{1}, y_{1}\right)$. Then $x_{1}^{2}+5 y_{1}^{2}$ is equal to :
(1) 6
(2) 10
(3) 8
(4) 5

Sol. 1
$\mathrm{T}: \frac{\mathrm{XX}}{1} 4-\frac{\mathrm{Yy}}{1}{ }_{2}^{2}=1$
$t: 2 x-y=0$ is parallel to $T$
$\Rightarrow T: 2 x-y=\lambda$
Now compare (1) \& (2)
$\frac{x_{1}}{\frac{4}{2}}=\frac{y_{1}}{2}=\frac{1}{\lambda}$
$x_{1}=8 / \lambda \& y_{1}=2 / \lambda$
$\left(x_{1}, y_{1}\right)$ lies on hyperbola $\Rightarrow \frac{64}{4 \lambda^{2}}-\frac{4}{2 \lambda^{2}}=1$
$\Rightarrow 14=\lambda^{2}$
Now $=x_{1}^{2}+5 y_{1}^{2}$
$=\frac{64}{\lambda_{2}}+5 \frac{4}{\lambda_{2}}$
$=\frac{84}{14}$
$=6$ Ans.
Q. 2 The domain of the function $f(x)=\sin ^{-1}\left(\frac{|x|+5}{x^{2}+1}\right)$ is $(-\infty,-a] \cup[a, \infty)$. Then a is equal to :
(1) $\frac{\sqrt{17}-1}{2}$
(2) $\frac{\sqrt{17}}{2}$
(3) $\frac{1+\sqrt{17}}{2}$
(4) $\frac{\sqrt{17}}{2}+1$

Sol. 3
$-1 \leq \frac{|x|+5}{x^{2}+1} \leq 1$
$-x^{2}-1 \leq|x|+5 \leq x^{2}+1$

## case-I

$-x^{2}-1 \leq|x|+5$
this inequality is always right $\forall x \in R$

## case - II

$|x|+5 \leq x^{2}+1$
$x^{2}-|x| \geq 4$
$|x|^{2}-|x|-4 \geq 0$
$\left(|x|-\left(\frac{1+\sqrt{17}}{2}\right)\right)\left(|x|-\left(\frac{1-\sqrt{17}}{2}\right)\right) \geq 0$
$|x| \leq \frac{1-\sqrt{17}}{2} \cup|x| \geq \frac{1+\sqrt{17}}{2}$
not possible
$x \in\left(-\infty, \frac{-1-\sqrt{17}}{2}\right] \cup\left[\frac{1+\sqrt{17}}{2}, \infty\right)$
$a=\frac{1+\sqrt{17}}{2}$
Q. 3 If a function $f(x)$ defined by $f(x)=\left\{\begin{array}{l}a e^{x}+b e^{-x},-1 \leq x<1 \\ c x^{2}, \quad 1 \leq x \leq 3 \\ a x^{2}+2 c x, 3<x \leq 4\end{array}\right.$ be continuous for some $a, b, c \in R$ and $f^{\prime}(0)+f^{\prime}(2)=e$, then the value of $a$ is :
(1) $\frac{1}{\mathrm{e}^{2}-3 \mathrm{e}+13}$
(2) $\frac{e}{e^{2}-3 e-13}$
(3) $\frac{e}{e^{2}+3 e+13}$
(4) $\frac{e}{e^{2}-3 e+13}$

Sol. 4
$f(x)$ is continuous

$$
\begin{aligned}
& \text { at } x=1 \Rightarrow a e+\frac{b}{e}=c \\
& \text { at } x=3 \Rightarrow 9 c=9 a+6 c \Rightarrow c=3 a \\
& \text { Now } f^{\prime}(0)+f^{\prime}(2)=e \\
& \Rightarrow a-b+4 c=e \\
& \Rightarrow a-e(3 a-a e)+4.3 a=e \\
& \Rightarrow a-3 a e+a e^{2}+12 a=e \\
& \Rightarrow 13 a-3 a e+a e^{2}=e \\
& \Rightarrow a=\frac{e}{13-3 e+e^{2}}
\end{aligned}
$$

Q. 4 The sum of the first three terms of a G.P. is $S$ and their product is 27 . Then all such $S$ lie in :
(1) $(-\infty,-9] \cup[3, \infty)$
(2) $[-3, \infty)$
(3) $(-\infty, 9]$
(4) $(-\infty,-3] \cup[9, \infty)$

## Sol. 4

$\frac{\mathrm{a}}{\mathrm{r}} \cdot \mathrm{a} \cdot \mathrm{ar}=27 \Rightarrow \mathrm{a}=3$

$$
\begin{aligned}
& \frac{a}{r}+a+a r=S \\
& \frac{1}{r}+1+r=\frac{S}{3} \\
& r+\frac{1}{r}=\frac{S}{3}-1 \\
& r+\frac{1}{r} \geq 2 \text { or } r+\frac{1}{r} \leq-2 \\
& \frac{S}{3} \geq 3 \text { or } \frac{S}{3} \leq-1 \\
& S \geq 9 \text { or } S \leq-3 \\
& S \in(-\infty,-3] \cup[9, \infty)
\end{aligned}
$$

Q. 5 If $R=\left\{(x, y): x, y \in Z, x^{2}+3 y^{2} \leq 8\right\}$ is a relation on the set of integers $Z$, then the domain of $R^{-1}$ is :
(1) $\{-1,0,1\}$
(2) $\{-2,-1,1,2\}$
(3) $\{0,1\}$
(4) $\{-2,-1,0,1,2\}$

Sol. 1
$3 y^{2} \leq 8-x^{2}$
R : $\{(0,1),(0,-1),(1,0),(-1,0),(1,1),(1,-1)$
$(-1,1),(-1,-1),(2,0),(-2,0),(-2,0),(2,1),(2,-1),(-2,1),(-2,-1)\}$
$\Rightarrow R:\{-2,-1,0,1,2\} \rightarrow\{-1,0,-1\}$
Hence $R^{-1}:\{-1,0,1\} \rightarrow\{-2,-1,0,1,2\}$
Q. 6 The value of $\left(\frac{1+\sin \frac{2 \pi}{9}+i \cos \frac{2 \pi}{9}}{1+\sin \frac{2 \pi}{9}-i \cos \frac{2 \pi}{9}}\right)^{3}$ is :
(1) $-\frac{1}{2}(1-i \sqrt{3})$
(2) $\frac{1}{2}(1-i \sqrt{3})$
(3) $-\frac{1}{2}(\sqrt{3}-i)$
(4) $\frac{1}{2}(\sqrt{3}-i)$

Sol. 3

$$
\begin{aligned}
& \left(\frac{1+\sin \frac{2 \pi}{9}+i \cos \frac{2 \pi}{9}}{1+\sin \frac{2 \pi}{9}-i \cos \frac{2 \pi}{9}}\right)^{3} \\
& =\left(\frac{1+\cos \left(\frac{\pi}{2}-\frac{2 \pi}{9}\right)+i \sin \left(\frac{\pi}{2}-\frac{2 \pi}{9}\right)}{1+\cos \left(\frac{\pi}{2}-\frac{2 \pi}{9}\right)-i \sin \left(\frac{\pi}{2}-\frac{2 \pi}{9}\right)}\right)^{3}
\end{aligned}
$$

$=\left(\frac{1+\cos \frac{5 \pi}{18}+i \sin \frac{5 \pi}{18}}{1+\cos \frac{5 \pi}{18}-i \sin \frac{5 \pi}{18}}\right)^{3}$
$=\left(\frac{2 \cos \frac{5 \pi}{36}\left\{\cos \frac{5 \pi}{36}+i \sin \frac{5 \pi}{36}\right\}}{2 \cos \frac{5 \pi}{36}\left\{\cos \frac{5 \pi}{36}-i \sin \frac{5 \pi}{36}\right\}}\right)^{3}$
$=\left(\frac{\operatorname{cis}\left(\frac{5 \pi}{36}\right)}{\operatorname{cis}\left(\frac{-5 \pi}{36}\right)}\right)^{3}$
$=\operatorname{cis}\left(\frac{5 \pi}{36} \times 3+\frac{5 \pi}{36} \times 3\right)$
$=\operatorname{cis}\left(\frac{10 \pi}{12}\right)$
$=\operatorname{cis}\left(\frac{5 \pi}{6}\right)=-\frac{\sqrt{3}}{2}+\frac{i}{2}$
Q. 7 Let $P(h, k)$ be a point on the curve $y=x^{2}+7 x+2$, nearest to the line, $y=3 x-3$. Then the equation of the normal to the curve at $P$ is:
(1) $x+3 y-62=0$
(2) $x-3 y-11=0$
(3) $x-3 y+22=0$
(4) $x+3 y+26=0$

Sol.

$$
C: y=x^{2}+7 x+2
$$

Let $P$ : $(h, k)$ lies on


Curve $=k=h^{2}+7 h+2$
Now for shortest distance
$M_{T} l_{p}^{c}=m_{L}=2 h+7=3$
$\mathrm{h}=-2$
$k=-8$
$P:(-2,-8)$
equation of normal to the curve is perpendicular to $L: 3 x-y=3$
$N: x+3 y=\lambda$
$\downarrow$ Pass $(-2,-8)$
$\lambda=-26$
$N: x+3 y+26=0$
Q. 8 Let $A$ be a $2 \times 2$ real matrix with entries from $\{0,1\}$ and $|A| \neq 0$. Consider the following two statements:
(P) If $A \neq I_{2}$, then $|A|=-1$
(Q) If $|A|=1$, then $\operatorname{tr}(A)=2$,
where $I_{2}$ denotes $2 \times 2$ identity matrix and $\operatorname{tr}(A)$ denotes the sum of the diagonal entries of $A$. Then:
(1) Both (P) and (Q) are false
(2) (P) is true and (Q) is false
(3) Both (P) and (Q) are true
$(4)(P)$ is false and $(Q)$ is true

Sol. 4
$P: A=\left[\begin{array}{ll}1 & 1 \\ 0 & 1\end{array}\right] \neq I_{2} \&|A| \neq 0 \&|A|=1$ (false)
$\mathrm{Q}: \mathrm{A}=\left[\begin{array}{ll}1 & 1 \\ 0 & 1\end{array}\right]=1$ then $\operatorname{Tr}(\mathrm{A})=2$ (true)
Q. 9 Box I contains 30 cards numbered 1 to 30 and Box II contains 20 cards numbered 31 to 50. A box is selected at random and a card is drawn from it. The number on the card is found to be a nonprime number. The probability that the card was drawn from Box I is:
(1) $\frac{4}{17}$
(2) $\frac{8}{17}$
(3) $\frac{2}{5}$
(4) $\frac{2}{3}$

Sol. 2

$$
1 \text { to } 30
$$

boxI
Prime on I
$\{2,3,5,7,11,13,17,19,23,29\}$
31 to 50
box II
Prime on II
$\{31,37,41,43,47\}$
A : selected number on card is non - prime
$P(A)=P(I) \cdot P(A / I)+P(I I) \cdot P(A / I I)$
$=\frac{1}{2} \times \frac{20}{30}+\frac{1}{2} \cdot \frac{15}{20}$
Now, $\mathrm{P}(\mathrm{I} / \mathrm{A})=\frac{\mathrm{P}(\mathrm{II}) \cdot \mathrm{P}(\mathrm{A} / \mathrm{I})}{\mathrm{P}(\mathrm{A})}$
$=\frac{\frac{1}{2} \cdot \frac{20}{30}}{\frac{1}{2} \cdot \frac{20}{30}+\frac{1}{2} \cdot \frac{15}{20}}=\frac{\frac{2}{3}}{\frac{2}{3}+\frac{3}{4}}=\frac{8}{17}$
Q. 10 If $p(x)$ be a polynomial of degree three that has a local maximum value 8 at $x=1$ and a local minimum value 4 at $x=2$; then $p(0)$ is equal to :
(1) 12
(2) -12
(3) -24
(4) 6

Sol. 2
$p^{\prime}(1)=0 \& p^{\prime}(2)=0$
$p^{\prime}(x)=a(x-1)(x-2)$
$p(x)=a\left(\frac{x^{3}}{3}-\frac{3 x^{2}}{2}+2 x\right)+b$
$p(1)=8 \Rightarrow a\left(\frac{1}{3}-\frac{3}{2}+2\right)+b=8$
$p(2)=4 \Rightarrow a\left(\frac{8}{3}-\frac{3.4}{2}+2.2\right)+b=4$
from equation (i) and (ii)
$a=24 \& b=-12$
$p(0)=b=-12$
Q. 11 The contrapositive of the statement "If I reach the station in time, then I will catch the train" is:
(1) If I will catch the train, then I reach the station in time.
(2) If I do not reach the station in time, then I will catch the train.
(3) If I do not reach the station in time, then I will not catch the train.
(4) If I will not catch the train, then I do not reach the station in time.

Sol. 4
Statement p and q are true
Statement, then the contra positive of the implication $p \rightarrow q=(\sim q) \rightarrow(\sim p)$
hence correct Ans. is 4
Q. 12 Let $\alpha$ and $\beta$ be the roots of the equation, $5 x^{2}+6 x-2=0$. If $S_{n}=\alpha^{n}+\beta^{n}, n=1,2,3, \ldots \ldots$, then:
(1) $5 \mathrm{~S}_{6}+6 \mathrm{~S}_{5}+2 \mathrm{~S}_{4}=0$
(2) $6 \mathrm{~S}_{6}+5 \mathrm{~S}_{5}=2 \mathrm{~S}_{4}$
(3) $6 \mathrm{~S}_{6}+5 \mathrm{~S}_{5}+2 \mathrm{~S}_{4}=0$
(4) $5 \mathrm{~S}_{6}+6 \mathrm{~S}_{5}=2 \mathrm{~S}_{4}$

Sol. 4
$5 x^{2}+6 x-2=0<_{\beta}^{\alpha}=5 \alpha^{2}+6 \alpha=2$
$6 \alpha-2=-5 \alpha^{2}$
Simillarly
$6 \beta-2=-5 \beta^{2}$
$\mathrm{S}_{6}=\alpha^{6}+\beta^{6}$
$S_{5}=\alpha^{5}+\beta^{5}$
$S_{4}=\alpha^{4}+\beta^{4}$
Now $6 \mathrm{~S}_{5}-2 \mathrm{~S}_{4}$
$=6 \alpha^{5}-2 \alpha^{4}+6 \beta^{5}-2 \beta^{4}$
$=a^{4}(6 \alpha-2)+\beta^{4}(6 \beta-2)$
$=\alpha^{4}\left(-5 \alpha^{2}\right)+\beta^{4}\left(-5 \beta^{2}\right)$
$=-5\left(\alpha^{6}+\beta^{6}\right)$
$=-5 \mathrm{~S}_{6}$
$=6 \mathrm{~S}_{5}+5 \mathrm{~S}_{6}=2 \mathrm{~S}_{4}$
Q. 13 If the tangent to the curve $y=x+\sin y$ at a point $(a, b)$ is parallel to the line joining $\left(0, \frac{3}{2}\right)$ and $\left(\frac{1}{2}, 2\right)$, then:
(1) $b=\frac{\pi}{2}+a$
(2) $|a+b|=1$
(3) $|b-a|=1$
(4) $b=a$

Sol. 3
$\left.\frac{d y}{d x}\right|_{p(a, b)} ^{c}=\frac{2-\frac{3}{2}}{\frac{1}{2}-0}$
$1+\cos b=1 \mid p:(a, b)$ lies on curve
$\cos b=0 \quad b=a+\sin b$
$b=a \pm 1$
$\mathrm{b}-\mathrm{a}= \pm 1$
$|\mathrm{b}-\mathrm{a}|=1$
Q. 14 Area (in sq. units) of the region outside $\frac{|x|}{2}+\frac{|y|}{3}=1$ and inside the ellipse $\frac{x^{2}}{4}+\frac{y^{2}}{9}=1$ is:
(1) $3(\pi-2)$
(2) $6(\pi-2)$
(3) $6(4-\pi)$
(4) $3(4-\pi)$

Sol. 2

$$
c_{1}: \frac{|x|}{2}+\frac{|y|}{3}=1
$$


$\mathrm{A}=4\left(\frac{\pi \mathrm{ab}}{4}-\frac{1}{2} \cdot 2 \cdot 3\right)$
$\mathrm{A}=\pi .2 .3-12$
$A=6(\pi-2)$
Q. 15 If $|x|<1,|y|<1$ and $x \neq y$, then the sum to infinity of the following series $(x+y)+\left(x^{2}+x y+y^{2}\right)+\left(x^{3}+x^{2} y+x y^{2}+y^{3}\right)+\ldots$ is:
(1) $\frac{x+y+x y}{(1-x)(1-y)}$
(2) $\frac{x+y-x y}{(1-x)(1-y)}$
(3) $\frac{x+y+x y}{(1+x)(1+y)}$
(4) $\frac{x+y-x y}{(1+x)(1+y)}$

Sol. 2
$(x+y)+\left(x^{2}+x y+y^{2}\right)+\left(x^{3}+x^{2} y+x y^{2}+y^{3}\right)+\ldots \infty$
$=\frac{1}{(x-y)}\left\{\left(x^{2}-y^{2}\right)+\left(x^{3}-y^{3}\right)+\left(x^{4}-y^{4}\right)+\ldots \infty\right\}$
$=\frac{\frac{x^{2}}{1-x}-\frac{y^{2}}{1-y}}{x-y}$
$=\frac{x^{2}(1-y)-y^{2}(1-x)}{(1-x)(1-y)(x-y)}$
$=\frac{\left(x^{2}-y^{2}\right)-x y(x-y)}{(1-x)(1-y)(x-y)}=\frac{((x+y)-x y)(x-y)}{(1-x)(1-y)(x-y)}$
$=\frac{x+y-x y}{(1-x)(1-y)}$
Q. 16 Let $\alpha>0, \beta>0$ be such that $\alpha^{3}+\beta^{2}=4$. If the maximum value of the term indepen dent of $x$ in the binomial expansion of $\left(\alpha x^{\frac{1}{9}}+\beta x^{-\frac{1}{6}}\right)^{10}$ is $10 k$, then $k$ is equal to:
(1) 176
(2) 336
(3) 352
(4) 84

Sol. 2
For term independent of $x$
$T_{r+1}={ }^{10} C_{r}\left(\alpha X^{\frac{1}{9}}\right)^{10-r} \cdot\left(\beta X^{-\frac{1}{6}}\right)^{r}$
$T_{r+1}={ }^{10} C_{r} \alpha^{10-r} \beta r \cdot x^{\frac{10-r}{9}} \cdot x^{-\frac{r}{6}}$
$\because \frac{10-r}{9}-\frac{r}{6}=0 \Rightarrow r=4$
$T_{5}={ }^{10} C_{r} \alpha^{6} \cdot \beta^{4}$
$\because \mathrm{AM} \geq \mathrm{GM}$
Now $\frac{\left(\frac{\alpha^{3}}{2}+\frac{\alpha^{3}}{2}+\frac{\beta^{2}}{2}+\frac{\beta^{2}}{2}\right)}{4} \geq \sqrt[4]{\frac{\alpha^{6} \cdot \beta^{4}}{2^{4}}}$
$\left(\frac{4}{4}\right)^{4} \geq \frac{\alpha^{6} \beta^{4}}{2^{4}}$
$\alpha^{6} \cdot \beta^{4} \leq 2^{4}$
${ }^{10} C_{4} \cdot \alpha^{6} \cdot \beta^{4} \leq{ }^{10} C_{4} 2^{4}$
$\mathrm{T}_{5} \leq^{10} \mathrm{C}_{4} 2^{4}$
$T_{5} \leq \frac{10!}{6!4!} .2^{4}$
$\mathrm{T}_{5} \leq \frac{10 \cdot 9 \cdot 8 \cdot 7.2^{4}}{4 \cdot 3 \cdot 2 \cdot 1}$
maximum value of $T_{5}=10.3 .7 .16=10 \mathrm{k}$
$\mathrm{k}=16.7 .3$
$k=336$
Q. 17 Let $S$ be the set of all $\lambda \in \mathrm{R}$ for which the system of linear equations
$2 x-y+2 z=2$
$x-2 y+\lambda z=-4$
$x+\lambda y+z=4$
has no solution. Then the set $S$
(1) is an empty set.
(2) is a singleton.
(3) contains more than two elements.
(4) contains exactly two elements.

Sol. 4
For no solution
$\Delta=0 \& \Delta_{1}\left|\Delta_{2}\right| \Delta_{3} \neq 0$
$\Delta=\left|\begin{array}{ccc}2 & -1 & 2 \\ 1 & -2 & \lambda \\ 1 & \lambda & 1\end{array}\right|=0$
$2\left(-2-\lambda^{2}\right)+1(1-\lambda)+2(\lambda+2)=0$
$-4-2 \lambda^{2}+1-\lambda+2 \lambda+4=0$
$-2 \lambda^{2}+\lambda+1=0$
$2 \lambda^{2}-\lambda-1=0 \Rightarrow \lambda=1,-1 / 2$
Equation has exactly 2 solution
Q. 18 Let $X=\{x \in N: 1 \leq x \leq 17\}$ and $Y=\{a x+b: x \in X$ and $a, b \in R, a>0\}$. If mean and variance of elements of $Y$ are 17 and 216 respectively then $a+b$ is equal to:
(1)-27
(2) 7
(3)-7
(4) 9

Sol. 3
$X:\{1,2, \ldots .17\}$
$Y:\{a x+b: x \in X \& a, b \in R, a>0\}$
Given $\operatorname{Var}(\mathrm{Y})=216$
$\frac{\sum y_{1}^{2}}{n}-(\text { mean })^{2}=216$
$\frac{\sum y_{1}^{2}}{17}-289=216$
$\sum y_{1}=8585$
$(a+b)^{2}+(2 a+b)^{2}+\ldots+(17 a+b)^{2}=8585$
$105 a^{2}+b^{2}+18 a b=505 \ldots$ (1)
Now $\sum \mathrm{y}_{1}=17 \times 17$
$a(17 \times 9)+17 . b=17 \times 17$
$9 a+b=17 \ldots(2)$
from equation (1) \& (2)
$a=3 \& b=-10$
$a+b=-7$
Q. 19 Let $y=y(x)$ be the solution of the differential equation, $\frac{2+\sin x}{y+1} \cdot \frac{d y}{d x}=-\cos x, y>0, y(0)=1$. If $y(\pi)=a$, and $\frac{d y}{d x}$ at $x=\pi$ is $b$, then the ordered pair $(a, b)$ is equal to:
(1) $\left(2, \frac{3}{2}\right)$
(2) $(1,1)$
(3) $(2,1)$
(4) $(1,-1)$

Sol. 2
$\int \frac{d y}{y+1}=\int \frac{-\cos x d x}{2+\sin x}$
$\ln |y+1|=-\ln |2+\sin x|+k$
$\downarrow(0,1)$
$k=\ln 4$
Now C: $(y+1)(2+\sin x)=4$
$y(\pi)=a \Rightarrow(a+1)(2+0)=4 \Rightarrow(a=1)$
$\left.\frac{d y}{d x}\right|_{x=\pi}=b \Rightarrow b=-(-1)\left(\frac{2+0}{1+1}\right)$
$\Rightarrow b=1$
$(a, b)=(1,1)$
Q. 20 The plane passing through the points $(1,2,1),(2,1,2)$ and parallel to the line, $2 x=3 y, z=1$ also passes through the point:
(1) $(0,-6,2)$
(2) $(0,6,-2)$
(3) $(-2,0,1)$
(4) $(2,0,-1)$

Sol. 3
$L:\left\{\begin{array}{cc}2 x=3 y \\ z=1\end{array}<_{Q:(3,2,1)}^{P:(0,0,1)}\right.$
$\vec{V}_{\mathrm{L}}$ Dr of line $(3,2,0)$

$\overrightarrow{\mathrm{n}}_{\mathrm{p}}=\overrightarrow{\mathrm{AB}} \times \overrightarrow{\mathrm{V}}_{\mathrm{L}}$
$\vec{n}_{p}=\langle 1,-1,1\rangle \times\langle 3,2,0\rangle$
$\vec{n}_{p}=\langle-2,+3,5\rangle$
Plane : $-2(x-1)+3(y-2)+5(z-1)=0$

Plane : $-2 x+3 y+5 z+2-6-5=0$
Plane: $2 x-3 y-5 z=-9$
Q. 21 The number of integral values of $k$ for which the line, $3 x+4 y=k$ intersects the circle, $x^{2}+y^{2}-2 x-$ $4 y+4=0$ at two distinct points is
Sol. 9
$c:(1,2) \& r=1$
$|c p|<r$

$$
\left|\frac{3.1+4.2-k}{5}\right|<1
$$

$|11-k|<5$
$-5<k-11<5$
$6<k<16$
$k=7,8,9, \ldots \ldots, 15 \Rightarrow$ total 9 value of $k$

Q. 22 Let $\vec{a}, \vec{b}$ and $\vec{c}$ be three unit vectors such that $|\vec{a}-\vec{b}|^{2}+|\vec{a}-\vec{c}|^{2}=8$. Then $|\vec{a}+2 \vec{b}|^{2}+|\vec{a}+2 \vec{c}|^{2}$ is equal to :
Sol. 2
$|\vec{a}-\vec{b}|^{2}+|\vec{a}-\vec{c}|^{2}=8$
$(\vec{a}-\vec{b}) \cdot(\vec{a}-\vec{b})+(\vec{a}-\vec{c})(\vec{a}-\vec{c})=8$
$a^{2}+b^{2}-2 a \cdot b+a^{2}+c^{2}-2 a \cdot c=8$
$2 a^{2}+b^{2}+c^{2}-2 a \cdot b-2 a \cdot c=8$
$a . b+a . c=-2$
Now $|\vec{a}+2 \vec{b}|^{2}+|\vec{a}+2 \vec{c}|^{2}$
$=2 a^{2}+4 b^{2}+4 c^{1}+4 \bar{a} \cdot \bar{b}+4 \bar{a} \cdot \bar{c}$
$=2+4+4+4(-2)$
$=2$
Q. 23 If the letters of the word 'MOTHER' be permuted and all the words so formed (with or without meaning) be listed as in a dictionary, then the position of the word 'MOTHER' is..........
Sol. 309
EHMORT

| E - - - | $=5!$ |
| :---: | :---: |
| H - - - | $=5!$ |
| M E - - - | $=4!$ |
| M H - - - | $=4!$ |
| M O E-- | $=3!$ |
| MOH - - | $=3!$ |
| M OR -- | $=3!$ |
| M OTE-- | $=2!$ |
| MOTHER | = 1 |
|  | 309 |

Q.24. If $\lim _{x \rightarrow 1} \frac{x+x^{2}+x^{3}+\ldots+x^{n}-n}{x-1}=820,(n \in N)$ then the value of $n$ is equal to :

Sol. 40
$\lim _{x \rightarrow 1} \frac{(x-1)}{x-1}+\frac{\left(x^{2}-1\right)}{x-1}+\ldots+\frac{\left(x^{n}-1\right)}{x-1}=820$
$\Rightarrow 1+2+3+\ldots \ldots+n=820$
$\Rightarrow \Sigma \mathrm{n}=820$
$\Rightarrow \frac{\mathrm{n}(\mathrm{n}+1)}{2}=820$
$\Rightarrow \mathrm{n}=40$
Q. 25 The integral $\int_{0}^{2}| | x-1|-x| d x$ is equal to :

Sol. 1.5

$$
\begin{aligned}
& \int_{0}^{2}| | x-1|-x| d x \\
& =\int_{0}^{1}|1-x-x| d x+\int_{1}^{2}|x-1-x| d x \\
& =\int_{0}^{1}|2 x-1| d x+\int_{1}^{2} 1 d x \\
& =\int_{0}^{\frac{1}{2}}(1-2 x) d x+\int_{\frac{1}{2}}^{1}(2 x-1) d x+\int_{1}^{2} 1 d x \\
& =\left[\left(\frac{1}{2}-0\right)-\left(\frac{1}{4}-0\right)\right]+\left(1-\frac{1}{4}\right)-\left(1-\frac{1}{2}\right)+1 \\
& =\frac{1}{2}-\frac{1}{4}+\frac{3}{4}-\frac{1}{2}+1 \\
& =\frac{3}{2}
\end{aligned}
$$

